

# 1    **A genome-scale metabolic model for *Methylococcus*** 2    ***capsulatus* predicts reduced efficiency uphill electron** 3    **transfer to pMMO.**

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# 1    **Abstract**

## 2    Background:

3    Genome-scale metabolic models allow researchers to calculate yields,  
 4    to predict consumption and production rates, and to study the effect of  
 5    genetic modifications *in silico*, without running resource-intensive  
 6    experiments. While these models have become an invaluable tool for  
 7    optimizing industrial production hosts like *E. coli* and *S. cerevisiae*, few  
 8    such models exist for one-carbon (C1) metabolizers.

## 9    Results:

10    Here we present a genome-scale metabolic model for *Methylococcus*  
 11    *capsulatus*, a well-studied obligate methanotroph, which has been used  
 12    as a production strain of single cell protein (SCP). The model was  
 13    manually curated, and spans a total of 877 metabolites connected via  
 14    898 reactions. The inclusion of 730 genes and comprehensive  
 15    annotations, make this model not only a useful tool for modeling  
 16    metabolic physiology, but also a centralized knowledge base for *M.*  
 17    *capsulatus*. With it, we determined that oxidation of methane by the  
 18    particulate methane monooxygenase is most likely driven through uphill  
 19    electron transfer operating at reduced efficiency as this scenario  
 20    matches best with experimental data from literature.

# 1 Conclusions:

2 The metabolic model will serve the ongoing fundamental research of C1  
3 metabolism, and pave the way for rational strain design strategies  
4 towards improved SCP production processes in *M. capsulatus*.

## 5 **Keywords**

6 COBRA, Genome-scale metabolic reconstruction, C1 Metabolism,  
7 Single Cell Protein, Methanotrophy

## 8 **Availability of Data and Materials**

9 The metabolic model, scripts and corresponding datasets generated  
10 and/or analysed during the current study are available in the GitHub  
11 repository.

- 12 • Archived: [https://github.com/ChristianLieven/memote-m-](https://github.com/ChristianLieven/memote-m-capsulatus)  
13 [capsulatus](https://github.com/ChristianLieven/memote-m-capsulatus)
- 14 • Most Recent: [https://github.com/ChristianLieven/memote-m-](https://github.com/ChristianLieven/memote-m-capsulatus/tree/develop)  
15 [capsulatus/tree/develop](https://github.com/ChristianLieven/memote-m-capsulatus/tree/develop)

16  
17 The data that support the biomass equation constructed in this study  
18 are available from Unibio at

- 19 • <http://www.unibio.dk/end-product/chemical-composition-1>

- 1       • <http://www.unibio.dk/end-product/chemical-composition-2>
- 2       Restrictions may apply to the availability of these data. Data are
- 3       however available from the authors upon reasonable request and with
- 4       permission of Unibio.

## 5       **Background**

6       The Gram-negative, obligate-aerobe *Methylococcus capsulatus* is a  
7       methane oxidizing, gamma-proteobacterium. Since its initial isolation by  
8       Foster and Davis [1], the organism has been subject to a wide array of  
9       studies. The global role of *M. capsulatus* as a participant in the carbon  
10      cycle has been elucidated [2, 3] as well as its effects on human [4] and  
11      animal health and disease [5]. Specifically the latter studies have been  
12      triggered by a considerable commercial interest in *M. capsulatus* as the  
13      primary microbe used for the production of Single Cell Protein (SCP) as  
14      animal feed starting in the 70s [6]. Now that hydraulic fracturing has  
15      made natural gas a cheap and abundant feedstock [7], the application  
16      of *M. capsulatus* for this purpose is being explored again [8, 9]. Another  
17      portion of studies, however, has focused on uncovering the biochemical  
18      and genetic basis of the organism's unique metabolism [10]. Yet, the  
19      greatest interest has been the role and function of the initial enzyme in  
20      methanotrophy, methane monooxygenase [11], responsible for  
21      oxidation of methane to methanol.

1 *Methylococcus capsulatus* is able to express two distinct types of  
 2 methane monooxygenases: a soluble form of methane monooxygenase  
 3 (sMMO) and a particulate, or membrane-bound form (pMMO). The  
 4 expression of these enzymes is strongly influenced by the extracellular  
 5 concentration of copper; when *M. capsulatus* is grown in the presence  
 6 of low levels of copper the sMMO is active, while the pMMO is  
 7 predominantly active at high levels. Both enzymes require an external  
 8 electron donor to break a C-H bond in methane. While the electron  
 9 donor for the sMMO is NADH [12–15], the native reductant to the  
 10 pMMO has not yet been elucidated due to difficulties to purify the  
 11 enzyme and assay its activity *in vitro* [11]. Three hypotheses regarding  
 12 the mode of electron transfer to the pMMO have been suggested  
 13 previously: 1) formaldehyde oxidation provides the electrons required to  
 14 oxidize methane in the form of NADH, while the electrons that are  
 15 released from the oxidation of methanol are passed via a cytochrome  
 16 ‘redox arm’ to a terminal oxidase, generating ATP. 2) A direct coupling  
 17 between the pMMO and the methanol dehydrogenase (MDH) allows an  
 18 efficient exchange of electrons, leading to the latter reaction driving the  
 19 former. 3) Finally, it is possible that electrons are transferred to the  
 20 pMMO from MDH through the ubiquinol pool by the action of a reversely  
 21 operating ubiquinol-cytochrome-c reductase. This mode is referred to as  
 22 ‘uphill’ electron transfer [17].

1 **Figure 1. Overview of the respiratory chain in *Methylococcus capsulatus***  
 2 **as implemented in iMcBath.** Black text in the center of the symbols denotes  
 3 the common abbreviation, while the faint grey text below denotes the  
 4 corresponding reaction ID in the metabolic model.

5 A genome-scale metabolic model (GEM) not only represents a  
 6 knowledge base that combines the available physiological, genetic and  
 7 biochemical information of a single organism [16], but also provides a  
 8 testbed for rapid prototyping of a given hypothesis [17]. Hence, we  
 9 present the first manually curated genome-scale metabolic model for  
 10 *Methylococcus capsulatus*, with the intent of supplying the basis for  
 11 hypothesis-driven metabolic discovery and clearing the way for future  
 12 efforts in metabolic engineering [18]. Using the GEM, we investigate the  
 13 nature of electron transfer in *M. capsulatus* by comparing the model's  
 14 predictions against experimental data from Leak and Dalton [19].  
 15 Furthermore, we compare its predictions to those of the model of *M.*  
 16 *buryatense* [20] and explain notable differences by considering the  
 17 proposed electron transfer modes.

# 1    **Results and Discussion**

## 2    **Reconstruction**

3    The presented genome-scale metabolic reconstruction of *M. capsulatus*  
4    termed iMcBath was based on BMID000000141026, an automatic draft  
5    published as part of the Path2Models project [21]. The whole genome  
6    sequence of *Methylococcus capsulatus* (Bath) (GenBank AE017282.2  
7    [22]) was used to aid the curation process and to supply annotations  
8    (see material & methods). This metabolic reconstruction consists of 840  
9    enzymatic reactions that interconvert 783 unique metabolites. The total  
10    number of reactions, including exchange, demand and the biomass  
11    reactions, is 892. The model attributes reactions and metabolites to  
12    three distinct compartments: The cytosol, the periplasm, and the  
13    medium that is referred to as extracellular in the model. Gene-Protein-  
14    Reaction rules (GPR) associated with 730 unique genes support 85.5%  
15    of the included reactions with, leaving 122 reactions without a GPR. The  
16    GPRs include representation of 24 enzyme complexes.

17    The model is made available in the community-standard SML format  
18    (Level 3 Version 2 with FBC).

## 19    **Biomass reaction**

20    In stoichiometric models biological growth is represented as the flux  
21    through a special demand reaction. The so-called biomass reaction

1 functions as a drain of metabolites, which are either highly reduced non-  
 2 polymeric macromolecules such as lipids and steroids or precursors to  
 3 typical biopolymers such as nucleic acids, proteins or long-chain  
 4 carbohydrates. The stoichiometry of an individual precursor was  
 5 calculated from the principal composition of *M. capsulatus* as reported  
 6 by Unibio [23]. The monomer composition of individual macromolecules  
 7 was calculated from different sources. A detailed account of the  
 8 resources is provided in the methods section and an overview of the  
 9 biomass reaction is given in Supplement Table 1.

10 The growth-associated maintenance (GAM) and the non-growth  
 11 associated maintenance (NGAM) requirements for *M. capsulatus* are  
 12 yet to be determined experimentally. Therefore, a GAM value of 23.087  
 13 mmol ATP gDW<sup>-1</sup> h<sup>-1</sup> was estimated according to Thiele et al. [16]  
 14 based on the data for *E. coli* published by Neidhardt et al. [24]. The  
 15 value for GAM is expected to increase with the growth rate of the cells  
 16 [25]. Like de la Torre et al. had done for *M. buryatense* [20], we  
 17 assumed the non-growth associated maintenance (NGAM) of *M.*  
 18 *capsulatus* to be similar to that of *E. coli* thus setting it to 8.39 mmol  
 19 ATP gDW<sup>-1</sup> h<sup>-1</sup> [26].



# 1 Metabolism

2 Much of the focus in curating the initial draft model BMID000000141026  
3 was put on the central carbon metabolism and respiration of *M.*  
4 *capsulatus*.

5 Three possible modes of electron transfer to the pMMO have been  
6 proposed.

7 1) In the *redox-arm* mode [27], the methanol dehydrogenase (MDH)  
8 passes electrons via cytochrome c555 (cL) [28] and cytochrome c553  
9 (cH) [29] to either a CBD- or AA3-type terminal oxidase [30] and thus  
10 contributes to building up a proton motive force (PMF) and the synthesis  
11 of ATP (Figure 1). The electrons required for the oxidation of methane  
12 are provided through ubiquinone (Q8H2), which in turn received  
13 electrons from various dehydrogenases involved in the oxidation of  
14 formaldehyde to CO<sub>2</sub>. Although no binding site has been identified,  
15 there is support for pMMO reduction by endogenous quinols [31].

16 2) With the *direct coupling* mode, the MDH is able to directly pass  
17 electrons to the pMMO [32–34]. Here, cytochrome c555 is the electron  
18 donor to the pMMO instead of ubiquinol. This mode is supported by  
19 results from cryoelectron microscopy which indicates that the pMMO  
20 and the MDH form a structural enzyme complex [35].

3) The *uphill electron transfer* mode supposes that the electrons from cytochrome c553 can reach the ubiquinol-pool facilitated by the PMF at the level of the ubiquinol-cytochrome-c reductase. This mode was proposed by Leak and Dalton [36] as it could explain the observed reduced efficiency.

To determine which of these modes is most likely active in *M. capsulatus*, Leak and Dalton [36] developed mathematical models for each mode based on previous calculations [37, 38]. However, their models did not reflect the experimental results. De la Torre et al. constructed a genome-scale metabolic model (GEM) for *Methylobacterium buryatense* to investigate growth yields and energy requirements in different conditions [20]. They found that the *redox-arm* mode correlated least with their experimental data for *M. buryatense*.

To study which of the three modes of electron transfer is active in *M. capsulatus*, they were each implemented in the model. The implementation is illustrated in Figure 2. To include the *redox-arm* we implemented the reaction representing the particulate methane monooxygenase, in the model termed as PMMOipp, utilizing Q8H2 as a cofactor. Accordingly, a variant of the pMMO reaction was added to account for a possible direct coupling to the MDH. In this variant reaction, termed PMMODCipp, the cofactor is cytochrome c555, represented as the metabolite focytcc555\_p in the model. To enable an

1 *uphill electron transfer*, the reaction representing the ubiquinol-  
2 cytochrome-c reductase (CYOR\_q8ppi in the model), was constrained  
3 to be reversible while keeping PMMOipp active.

4 **Figure 2. The three possible modes of electron transfer to the pMMO**  
5 **mapped onto an excerpt of the respiratory chain.** 1) Redox-arm: The  
6 methanol dehydrogenase transfers electrons to the terminal oxidase, while  
7 the pMMO draws electrons from the quinone pool. 2) Direct coupling:  
8 Electrons from the oxidation of methanol are transferred directly to the pMMO.  
9 3) Uphill electron transfer: Electrons from the methanol dehydrogenase feed  
10 back into the ubiquinol-pool. Black text denotes the common name, while faint  
11 grey text denotes the reaction ID in the metabolic model.

12 Following the path of carbon through metabolism downstream from the  
13 MDH, the model includes both the reaction for a ubiquinone-dependent  
14 formaldehyde dehydrogenase [39], termed FALDDHipp, and an NAD-  
15 dependent version, termed ALDD1 (Figure 3). Despite of the initial  
16 evidence for the latter reaction [40] having been dispelled by Adeosun  
17 et al. [41], it was added to allow further investigation into the presence  
18 of a putative enzyme of that function. An additional pathway for  
19 formaldehyde oxidation represented in both genome and model is the  
20 tetrahydromethanopterin(THMPT)-linked pathway [42].

**Figure 3. Three different formaldehyde oxidation pathways are represented in the model.** Black text denotes the common name, while faint grey text denotes the reaction ID in the metabolic model.

Formaldehyde assimilation in *M. capsulatus* occurs primarily through the ribulose monophosphate (RuMP)-pathway. As outlined by Anthony [10], the RuMP-pathway has four hypothetical variants. Based on the annotated genome published by Ward et al. [22], we identified not only both C6 cleavage pathways depending either on the 2-keto, 3-deoxy, 6-phosphogluconate (KDPG) aldolase (EDA) or the fructose bisphosphate aldolase (FBA), but also the transaldolase (TA) involved in the rearrangement phase that regenerates ribulose 5-phosphate. The alternative to a transaldolase-driven rearrangement step is the use of a sedoheptulose biphosphatase, which was not included in the initial annotation. Strøm et al. could not detect specific activity using cell-free preparations [43]. Yet, we decided to add a corresponding reaction for two reasons. First, the FBA has been characterized to reversibly catalyze sedoheptulose biphosphate cleavage [44], which is reflected by the reaction FBA3 in the model. Second, the pyrophosphate-dependent 6-phosphofructokinase (PFK\_3\_ppi) was reported to have higher affinity and activity to the reversible phosphorylation of sedoheptulose phosphate than compared to fructose 6-phosphate [45]. Thus, all of the resulting four combinations that make up the RuMP-pathway are represented in this metabolic model (Figure 4).

1 **Figure 4. All four variants of the RuMP pathway are represented in the**  
 2 **metabolic model.** If there is a common enzyme name the black text denotes  
 3 the common name, while faint grey text denotes the reaction ID in the  
 4 metabolic model. Otherwise the black text is also the reaction ID in the model.

5 The genome of *Methylococcus capsulatus* encodes genes for a  
 6 complete Calvin Benson Bassham (CBB) cycle [46–48] and a partial  
 7 Serine pathway for formaldehyde assimilation [22]. It was argued by  
 8 Taylor et al. [49] that *M. capsulatus* can metabolize glycolate (a product  
 9 of the oxygenase activity of the ribulose biphosphate carboxylase  
 10 (RBPC)) via this pathway. Both Taylor and Ward, further suggested the  
 11 presence of unique key enzymes to complete the Serine pathway, such  
 12 as hydroxymethyl transferase, hydroxypyruvate reductase and malate-  
 13 CoA lyase. However, since the gene annotation did not reflect this and  
 14 the RuMP pathway is reportedly the main pathway for formaldehyde  
 15 assimilation [50], these putative reactions were not included.

16 All genes of the TCA cycle were identified in the genome sequence and  
 17 all associated reactions were included accordingly [22]. Because no  
 18 activity of the 2-oxoglutarate dehydrogenase has so far been measured  
 19 *in vivo* [50, 51], the associated reactions have been constrained to be  
 20 blocked (both lower and upper bounds were set to zero). This way they  
 21 can easily be activated if needed. For instance, if a growth condition is  
 22 discovered where activity for this enzyme can be detected.

1 Based on reactions already present in BMID000000141026, the  
 2 information in the genome annotation, and the measured biomass  
 3 composition, we curated the biosynthetic pathways of all proteinogenic  
 4 amino acids, nucleotides, fatty acids, phospholipids, panthothenate,  
 5 coenzyme A, NAD, FAD, FMN, riboflavin, thiamine, myo-inositol, heme,  
 6 folate, cobalamine, glutathione, squalene, lanosterol, peptidoglycan.  
 7 Since no corresponding genes could be identified, reactions catalyzing  
 8 the biosynthesis of lipopolysaccharide (LPS) were adopted from  
 9 iJO1366 [52] under the assumption that the biosynthesis steps among  
 10 gram-negative bacteria require amounts of ATP comparable to  
 11 *Escherichia coli*.

## 12 **Figure 5. Oxidation of ammonia to nitrite.**

13 *Methylococcus capsulatus* is able to metabolize the nitrogen sources  
 14 ammonium ( $\text{NH}_4$ ) and nitrate ( $\text{NO}_3$ ) in a variety of ways. When the  
 15 extracellular concentration of  $\text{NH}_4$  is high, alanine dehydrogenase  
 16 (ADH) is the primary pathway for nitrogen assimilation into biomass  
 17 [53]. In addition, the two monooxygenases are able to oxidize  
 18 ammonium to hydroxylamine [15, 54], which is then further oxidized by  
 19 specific enzymes first to nitrite [55] (Figure 5), and even to dinitrogen  
 20 oxide [56].  $\text{NO}_3$  is reduced to  $\text{NH}_4$  via nitrite and ultimately assimilated  
 21 via the glutamine synthetase/ glutamine synthase (GS/GOGAT)  
 22 pathway. Furthermore, it has been shown that *M. capsulatus* is able to

1 fix atmospheric nitrogen ( $N_2$ ) [57]. The nitrogenase gene cluster has  
 2 been identified [58] and annotated accordingly [22], and the  
 3 corresponding reactions have been included in the model. As the  
 4 enzyme has not yet been specifically characterized, the nitrogenase  
 5 reaction was adapted from iAF987 [59]. A schema showing these  
 6 reactions side-by-side is displayed in Figure 6.

7 **Figure 6. Nitrogen assimilation and fixation pathways represented in the**  
 8 **model.** Black text in the centre of the symbols denotes reaction IDs. Common  
 9 names are used for metabolites.

10 A metabolic map of all the reactions in the model was built using Escher  
 11 [60] and is available as Supplement Figure 1.

## 12 Extension and manual curation

13 Starting reconstruction on the basis of an automated draft required  
 14 additional effort to be able to use it for calculations. The automated draft  
 15 used two sets of ID namespaces, BiGG [61] and MetaNetX (MNX) [62].  
 16 Hence, the first step in the curation efforts consisted of unifying the  
 17 namespace by mapping all metabolite and reaction identifiers from MNX  
 18 into the human-readable BiGG namespace. Individual metabolites and  
 19 reactions with unmappable IDs, that could not be identified in the BiGG  
 20 database and for which there was little evidence in the form of GPR  
 21 rules, were removed from the model. Several metabolite formulas

1 contained polymer units, and many reactions lacked EC numbers.  
 2 Using the MNX spreadsheet exports 'chem\_prop.tsv' and  
 3 'reac\_prop.tsv' from version 1.0 [63] and 2.0 [64] most of these issues  
 4 were resolved. Due to said malformed and missing metabolite formulae,  
 5 many reactions were not mass-charge-balanced. We used the  
 6 'check\_mass\_balance' function in cobrapy [65] to identify and balance  
 7 99.4% of the reactions in the model. The remaining five reactions,  
 8 belonging to fatty acid biosynthesis, involve a lumped metabolite that  
 9 represents the average fatty acid content of *M. capsulatus*.

10 Another challenge with extending the automated draft was that many  
 11 reactions were inverted meaning that reactants and products were  
 12 switched, and constrained to be irreversible. Consulting the  
 13 corresponding reactions in MetaCyc [66], these instances were  
 14 corrected manually. For each precursor in the biomass reaction, the  
 15 corresponding biosynthetic pathway was gap-filled and manually  
 16 curated. To identify the appropriate pathways, MetaCyc pathway maps  
 17 from the *M. capsulatus* Bath specific database were used to compare  
 18 the reactions that were present in the draft reconstruction  
 19 (<https://biocyc.org/organism-summary?object=MCAP243233>).

20 In order to enable other researchers to integrate iMcBath into their  
 21 respective workflows and to simplify model cross-comparisons, we



1 included MIRIAM-compliant annotations for a majority of the model's  
2 reactions (96%) and metabolites (98%) [67].

3 As a last step we added transport reactions that were not in the draft  
4 reconstruction. Inferring membrane transport reactions from the  
5 genome sequence is difficult, as usually the precise 3D structure of  
6 transport proteins dictates which metabolite classes can be transported  
7 [68]. Even if the substrates are known, the energy requirements of  
8 transport are often undefined. Working on protein sequence matches  
9 using PsortB 3.0 [69], combined with BLAST [70] matches against  
10 TransportDB [71] and the Transporter Classification Database (TCDB)  
11 [72], we were able to identify 56 additional transport reactions. We have  
12 limited the number of transporters to be added, focusing specifically on  
13 transporters with known mechanisms and transport of metabolites  
14 already included in the model. A list of putative transport-associated  
15 genes that we did not consider is available at  
16 <https://github.com/ChristianLieven/memote-m-capsulatus>. Some of  
17 these genes were reported by Ward et al. to facilitate the uptake of  
18 sugars [22]. Kelly et al. suggest [50] these genes may allow a function  
19 similar to *Nitrosomas europaea*, which is able grow on fructose as a  
20 carbon-source with energy from ammonia oxidation [73]. This list is a  
21 good starting point to study potential alternate carbon source use by *M.*  
22 *capsulatus*.

# 1 Validation of the Model

2 To determine which combination of the three aforementioned electron  
 3 transfer modes is active in *Methylococcus capsulatus*, we constrained  
 4 the model based on experiments conducted by Leak and Dalton [19].  
 5 Since the three modes relate to how the pMMO receives electrons, we  
 6 focused on the data generated by growing *M. capsulatus* in high-copper  
 7 medium, which is the condition in which pMMO is predominantly active.  
 8 We used the average of carbon and oxygen-limited measurements as a  
 9 reference. Having constrained the model, we compared the Leak and  
 10 Dalton's measurements for the ratio of oxygen consumption to methane  
 11 consumption ( $O_2/CH_4$ ) to the predictions of the model (Figure 7A). We  
 12 considered the  $O_2/CH_4$  ratio to be a key metric for the respiratory chain  
 13 in *M. capsulatus*, as it is a function of the mode of electron transfer to  
 14 the pMMO. The central carbon metabolism was left unconstrained.

15 Under the assumption that the mode of electron transfer to pMMO  
 16 would be independent of the source of nitrogen, we compared the  
 17  $O_2/CH_4$  ratios of the model constrained to employ one of the three  
 18 modes of electron transfer exclusively to the corresponding reference  
 19 values of *M. capsulatus* grown on either  $NO_3$  or  $NH_4$ . However, neither  
 20 of the modes adequately represented the measured  $O_2/CH_4$  ratios of  
 21 about 1.43 and 1.6, respectively. Although Leak and Dalton had  
 22 proposed that the *reverse* or *uphill electron transfer* is the most

1 probable mode [36], the model predictions allowing for an unbounded  
2 *uphill transfer* did not support this (Figure 7), as the efficiency was  
3 almost comparable to predictions using *direct coupling*.

4 We altered the efficiency of the three modes to determine whether the  
5 fit could be improved. For the *redox arm*, we gradually decreased the  
6 mol protons required for the synthesis of 1 mol ATP, thereby improving  
7 the efficiency. This did not change the O<sub>2</sub>/CH<sub>4</sub> ratio (See Supplement  
8 Table 2). We decreased the efficiency of the *direct coupling* mode  
9 (PMMODCipp) by forcing a portion of the flux through the regular  
10 pMMO reaction (PMMOipp) using ratio constraints. Although this  
11 affected the O<sub>2</sub>/CH<sub>4</sub> ratio, to achieve a ratio representing the measured  
12 value of 1.43 would require a large decrease in efficiency with almost  
13 85% of the incoming carbon to be routed through the regular pMMO  
14 reaction (See Supplement Table 2). Because of this, we consider a  
15 *direct coupling* to be possible, albeit unlikely. Lastly, we iteratively  
16 constrained the lower bound of the reaction associated with the  
17 ubiquinol-cytochrome-c reductase (CYOR\_q8ppi), to reduce the  
18 efficiency of the *uphill-electron transfer*.

19 As Figure 7B shows, by constraining the reverse flux through this  
20 reaction, it is possible to modulate the ratio of O<sub>2</sub>/CH<sub>4</sub> consumption. We  
21 decided to fit the lower bound of ubiquinol-cytochrome-c reductase only  
22 to the reference O<sub>2</sub>/CH<sub>4</sub> ratio of 1.43 with cells grown on NO<sub>3</sub>, in order

1 to avoid an overlap with the effects of NO<sub>2</sub> production. Leak and Dalton  
2 pointed out, that the unexpectedly high O<sub>2</sub>/CH<sub>4</sub> ratio of 1.6 was the  
3 product of latent NH<sub>4</sub> oxidation rather than assimilation [19] leading to  
4 elevated levels of NO<sub>2</sub>. They were uncertain whether this increase could  
5 be attributed to the energetic burden of reducing NH<sub>4</sub> or the uncoupling  
6 effect of NO<sub>2</sub>.

7 To investigate this effect, we introduced a ratio constraint (see Methods)  
8 coupling the uptake of NH<sub>4</sub> to the excretion of NO<sub>2</sub> and explored a  
9 number of values for this ratio (Figure 7C). According to the simulations,  
10 the energy spent on reducing about 50% of incoming NH<sub>4</sub> to NO<sub>2</sub> is  
11 sufficient to account for the observed, high O<sub>2</sub>/CH<sub>4</sub> ratio of 1.6. Although  
12 this shows that the loss of energy could be significant enough to  
13 account for the increased ratio, this does not exclude a potential  
14 combined effect because of energy decoupling. Regardless it shows  
15 that predictions using the metabolic model can accurately reflect the *in*  
16 *vivo* behavior of *M. capsulatus*.

17 **Figure 7 Validation of iMcBath. A:** Using each of the three electron transfer  
18 modes exclusively (Redox-Arm, Direct Coupling, Uphill Transfer), the ratios of  
19 oxygen to methane consumption (O<sub>2</sub>/CH<sub>4</sub>) predicted by iMcBath were  
20 compared to experimental values from Leak and Dalton [19]. Since none of  
21 the three modes matched the reference, the efficiency of the Uphill Transfer  
22 was reduced by iteratively constraining the flux through the ubiquinol-  
23 cytochrome c reductase (CYOR\_q8) reaction as shown in **B**. Using a lower

bound of -2.8 improved the fit to the reference value for cells grown on NO<sub>3</sub> (**A**  
– Uphill Transfer Fit) **C**: To account for the energy loss through NH<sub>3</sub> oxidation,  
several ratios of NH<sub>4</sub> uptake to NO<sub>2</sub> production were considered. The closest  
fit was achieved with a ratio of around 0.5 (**A** – + ½ NO<sub>2</sub> Production)

## Comparison with other models

We compared iMcBath, the preceding automated draft reconstruction  
BMID000000141026, and a genome-scale metabolic model of the  
gram-negative, gamma-proteobacterium *Methylobacterium buryatense*  
strain 5G(B1) [20] to illustrate how much the model has progressed  
from the draft through manual curation and how the mode of electron  
transfer affects growth parameters within the group of gamma-  
proteobacteria (See Table 1).

Unsurprisingly, the automated draft generally performs quite poorly in  
comparison with the curated models. It's not possible to produce  
biomass from methane, even in rich media conditions, which is  
indicative of gaps in the pathways leading towards individual biomass  
components. Moreover, ATP can be produced without the consumption  
of substrate, which in turn means that key energy reactions are not  
constrained thermodynamically to have the correct reversibility. In the  
draft, 51% of the reactions are not supported by Gene-Protein-Reaction  
rules (GPR), while in iMb5G(B1) this percentage is only 32% and in  
iMcBath the percentage of these reactions is under 20%. GPRs allow

1 modelers to distinguish between isozymes and enzyme complexes by  
 2 connecting gene IDs either through boolean ‘OR’ or ‘AND’ rules. In  
 3 iMcBath, 25 complexes in total were curated and formulated as GPR.  
 4 Neither the draft model nor iMb5G(B1) make this distinction.

5 In the automated draft, the oxidation of methane was only possible  
 6 through a reaction representing the sMMO (MNXR6057). In iMcBath,  
 7 this was corrected by also implementing reactions that represent the  
 8 pMMO, one with ubiquinone as the electron donor (PMMOipp), and  
 9 another that receives electrons from the MDH (PMMODCipp).

10 *Methylococcus capsulatus*, like *Methylobacterium buryatense*,  
 11 expresses both the soluble and the particulate monooxygenase  
 12 depending on the availability of copper in the medium. In iMb5G(B1)  
 13 however only the particulate monooxygenase is represented by the  
 14 reaction “pMMO”. The ability of *M. capsulatus* to grow on ammonia,  
 15 nitrate and nitrogen has been characterized experimentally [53]. In  
 16 addition to that, however, iMcBath also predicts growth on nitrite and  
 17 urea. Nitrite is an intermediate in the assimilation of nitrate, yet also  
 18 reported to elicit toxic effects [19, 74], hence *in vivo* growth may only be  
 19 possible on low concentrations. Growth on urea is possible in the model  
 20 since we identified *MCA1662* as an ATP-driven urea transporter and  
 21 gap-filled a urease reaction when curating the cobalamine biosynthesis  
 22 pathway. As a consequence of this, urea can be taken up into the  
 23 cytosol and broken down into ammonia and carbon dioxide, the former

1 of which is then assimilated. Further experimental work is necessary to  
 2 verify this *in vivo*. *M. buryatense* 5G(B1) is reported to grow on nitrate,  
 3 ammonia and urea, yet without adding the respective transport  
 4 reactions iMb5G(B1) only grows on nitrate. While the draft model for *M.*  
 5 *capsulatus* contains exchange reactions for all the tested nitrogen  
 6 sources except for urea, it couldn't grow at all, which again is likely  
 7 because of gaps in the pathways leading to biomass precursor  
 8 metabolites.

9 The difference between *M. capsulatus* and *M. buryatense* becomes  
 10 apparent when comparing the growth energetics of iMcBath and  
 11 iMb5G(B1). iMb5G(B1) produces more  $\text{mmol gDW}^{-1} \text{ h}^{-1}$  ATP than  
 12 iMcBath. Because of this the hypothetical growth-rates predicted by  
 13 iMb5G(B1) are higher than those of iMcBath regardless of the  
 14 respective nitrogen source. This difference is likely a direct  
 15 consequence of the mode of electron transfer to the monooxygenase  
 16 and thus the efficiency of the methane oxidation reaction in total. When  
 17 comparing the ratio of the uptake rate of oxygen and the uptake rate of  
 18 methane for the two models, we can see that the resulting values in  
 19 iMb5G(B1) are lower than in iMcBath. It was recently reported, that  
 20 instead of the reverse-electron transfer and redox-arm mode active in  
 21 *M. capsulatus*, a mixture of *direct coupling* from pMMO to MDH and  
 22 *reverse electron transfer* seems to be the most likely mode in *M.*  
 23 *buryatense* 5G(B1) [20].

1 **Table 1. Model Comparison.** The presented reconstruction iMcBath, the  
2 automated draft BMID000000141026 and iMb5G(B1), a genome-scale  
3 reconstruction of *Methylobacterium buryatense* strain 5G(B1).

	iMcBath	BMID000000141026	iMb5G(B1)
<b>Structure</b>			
	SMMOi,		
Methane Oxidation	PMMOipp,	MNXR6057	pMMO
	PMMODCipp		
Growth on N-Sources	Urea, NO <sub>2</sub> ,	No Growth	NO <sub>3</sub>
	NO <sub>3</sub> , NH <sub>4</sub> , N <sub>2</sub>		
<b>Performance</b>			
ATP Production Closed			
Exchanges	False	True	False
ATP Production Rate - NH <sub>4</sub>	0.775	1000	1
ATP Production Rate - NO <sub>3</sub>	0.365	1000	0.519
Growth Rate - pMMO - NH <sub>4</sub>	0.299	No Growth	0.34
Growth Rate - pMMO - NO <sub>3</sub>	0.205	No Growth	0.27
O <sub>2</sub> /CH <sub>4</sub> Ratio - pMMO - NH <sub>4</sub>	1.434	No Growth	1.156
O <sub>2</sub> /CH <sub>4</sub> Ratio - pMMO - NO <sub>3</sub>	1.415	No Growth	1.116
<b>Specifications</b>			
Reactions without GPR	174	950	129
Enzyme Complexes	43	0	0
Total # Genes	730	589	313
Total # Metabolites	877	935	403
Total # Reactions	898	1858	402



# 1 Conclusion

2 IMCBATH is the first, manually curated, genome-scale metabolic model  
 3 for *Methylococcus capsulatus*. With iMcBath, we combine biochemical  
 4 knowledge of half a century of research on *Methylococcus capsulatus*  
 5 into a single powerful resource, providing the basis for targeted  
 6 metabolic engineering, process design and optimization, omic-data  
 7 contextualization and comparative analyses. We applied the metabolic  
 8 model to study the complex electron transfer chain of *M. capsulatus*, by  
 9 analyzing the three modes of electron transfer that had been proposed  
 10 previously [36]. We did so by corresponding each mode with the flux  
 11 through a reaction in the model, and consequently comparing the  
 12 predicted O<sub>2</sub>/CH<sub>4</sub> ratios to experimentally measured values by Leak and  
 13 Dalton [19]. Simulation and experiment agreed either when the model  
 14 was constrained to employ the *uphill electron transfer* at reduced  
 15 efficiency or *direct coupling* at strongly reduced efficiency for *M.*  
 16 *capsulatus* grown *in silico* on NO<sub>3</sub> as the source of nitrogen. Although  
 17 an *uphill electron-transfer* seems more likely, further experimental  
 18 validation is required as neither mode can be ruled out exclusively. The  
 19 experimentally observed effect of NH<sub>4</sub> oxidation to NO<sub>2</sub> could be  
 20 replicated by considering the energy burden alone.

21 Future applications of the metabolic model could include hypothesis  
 22 testing of the regulation of the MMO in other growth conditions [50],

1 studying the effects of the predicted hydrogenases on the energy  
2 metabolism of *M. capsulatus* [22], and exploring venues of metabolic  
3 engineering for an improved production of metabolites [75]. Another

## 4 **Methods**

### 5 Model Curation

6 After mapping the reaction and metabolite identifiers from the MetaNetX  
7 namespace to the BiGG namespace, we proceeded with the curation  
8 efforts as follows: First, we chose a subsystem of interest, then we  
9 picked a pathway and using information from either the genome  
10 sequence, published articles, the metacyc or uniprot databases, and  
11 lastly, we enriched each enzymatic step in said pathway with as much  
12 information as possible. Information that was added included for  
13 instance: GPR, reaction localization, EC numbers, a confidence score,  
14 possible cofactors and inhibitors and cross references to other  
15 databases such as KEGG, BiGG and MetaNetX. For each metabolite  
16 involved in these reactions, we defined the stoichiometry, charge and  
17 elemental formula, either based on the corresponding entries in the  
18 BiGG database or on clues from literature.

19 If reactions from a pathway were present in the draft, we checked their  
20 constraints and directionality. This was necessary as many of the  
21 irreversible reactions in the draft reconstruction seemed to have been

1 'inverted' when compared to the corresponding reactions in the  
2 reference databases, which made flux through them impossible in  
3 normal growth conditions.

4 The energy metabolism and methane oxidation were curated first.  
5 Except for the reaction representing the sMMO, all reactions were newly  
6 constructed, as they were absent in the draft. Then, in order to achieve  
7 sensible FBA solutions for growth on methane, the central carbon  
8 metabolism, amino acid and nucleotide biosynthesis pathways were  
9 manually curated. Simultaneous to the manual curation a metabolic  
10 pathway map was created in Escher, which helped us to maintain a  
11 visual checklist of curated pathways.

12 The automated draft contained a rudimentary, generic biomass  
13 reaction, which only accounted for the production of proteins, DNA and  
14 RNA, but not for the biosynthesis of a cell wall and cell membrane, the  
15 maintenance of a cofactor pool, the turnover of trace elements or the  
16 energetic costs associated with growth. After calculating a more specific  
17 biomass composition for *M. capsulatus*, further pathway curation was  
18 necessary to achieve growth *in silico*. This included the biosynthesis  
19 pathways of fatty acids (even, odd and cyclopropane varieties),  
20 phospholipids, coenzyme A, Ubiquinol, Lanosterol, FAD, FMN,  
21 Riboflavin, NAD and NADP, Heme, Glutathione, Cobalamin, Thiamine,  
22 Myo-Inositol, and Lipopolysaccharides.

1 To account for the reported differences in ammonia assimilation of *M.*  
 2 *capsulatus* when grown in the presence of excess ammonia versus the  
 3 growth on either atmospheric nitrogen or nitrate, we curated the  
 4 nitrogen metabolism including the oxidation of ammonia to nitrite via  
 5 hydroxylamine, the reduction of nitrate and nitrite, the glutamine  
 6 synthetase/ glutamate synthase reactions and the alanine  
 7 dehydrogenase. Reversible degradation reactions producing ammonia  
 8 that would potentially bypass the characterized ammonia assimilation  
 9 pathways were constrained to be irreversible accordingly.

10 After we had enriched the annotations already in the draft with  
 11 annotations from the metabolic models iJO1366 [52], iRC1080 [76],  
 12 iJN678 [77] and iHN637 [78], they were converted into a MIRIAM-  
 13 compliant format. As a final step, we manually added transport  
 14 reactions to reflect the uptake energetics of cofactors.

15 Throughout the reconstruction process, we iteratively tested and  
 16 validated the reconstruction. For instance, we checked the mass and  
 17 charge balance of each reaction, attempting to manually balance those  
 18 that weren't balanced. In the majority of cases metabolites were missing  
 19 formula or charge definitions. In order to remove energy generating  
 20 cycles, problematic reactions were manually constrained to be  
 21 irreversible. Validation was carried out against growth data [19], which  
 22 was also the point of reference for the parameter fitting.

# 1 Biomass Composition

2 For the principal components of biomass, measurements were made  
 3 available through the website of our collaborators Unibio [23]. This  
 4 included specifically the content of RNA (6.7%), DNA (2.3%), crude fat  
 5 (9.1%), and glucose (4.5%) as a percentage of the cell dry weight. We  
 6 did not use the percentage values for crude protein (72.9%) and N-free  
 7 extracts (7.6%) as these measurements are fairly inaccurate relying on  
 8 very generalized factors. The percentage value of Ash 550 (8.6%)  
 9 measurements was inconsistent with the sum of its individual  
 10 components (4.6%) and was hence excluded.

11 On the homepage of UniBio, we were also able to find g/kg  
 12 measurements of all amino acids except for glutamine and asparagine,  
 13 trace elements and vitamins, which could directly be converted into  
 14 mmol/g DW. However, we omitted some of data: The stoichiometries for  
 15 Selenium, Cadmium, Arsenic, Lead and Mercury were not included in  
 16 the biomass reaction as their values were negligibly small. Beta-  
 17 Carotene (Vitamin A) and Gama-Tocopherol (Vitamin E) were omitted  
 18 because no genes were found supporting their biosynthesis, in addition  
 19 to both being reportedly below the detection limit [79].

20 For the lack of better measurements, and assuming that *M. buryatense*  
 21 and *M. capsulatus* are more similar than *M. capsulatus* and *E. coli*, the  
 22 stoichiometries of glutamine and asparagine, intracellular metabolites

1 such as ribulose-5-phosphate, organic cofactors such as coenzyme A,  
2 and cell wall components such as LPS were introduced from de la Torre  
3 et al. [20].

4 Using the GC content calculated from the genome sequence [22] and  
5 the percentage measurements from Unibio for RNA and DNA, we were  
6 able to calculate the stoichiometries of all individual nucleobases.

7 Unibio's measurements that 94% of the crude fat portion were fatty  
8 acids conflicted with previously published results, which indicated that in  
9 fact phospholipids are likely to be the main lipid component in *M.*  
10 *capsulatus* [80, 81]. Thus, we assumed 94% of the crude fat to be  
11 phospholipids. This meant that 6% of the crude fat was composed of  
12 fatty acids, the distributions of which were again provided by Unibio.  
13 However, in order to calculate the stoichiometry of each fatty acid we  
14 recalculated the distribution to exclude the unidentified part. Makula had  
15 also measured the composition of the phospholipid pool itself [80], from  
16 which we calculated the corresponding stoichiometries for  
17 phosphatidylethanolamine, phosphatidylcholine, phosphatidylglycerol  
18 and cardiolipin.

19 Bird and colleagues had reported the percentage of squalene and  
20 sterols of dry weight [82], which we converted into mmol/g DW without  
21 further corrections. Since the genes for hopanoid synthesis were  
22 identified we included diplopterol in the biomass reaction [22, 83]. For a

1 lack of more detailed measurements we estimated a similar contribution  
2 to the overall composition as squalene. We specifically used lanosterol  
3 to represent the general class of sterols in the biomass reaction, since  
4 we had only been able to reconstruct the synthesis pathway of  
5 lanosterol and since lanosterol is a key precursor metabolite in the  
6 biosynthesis of all other sterols.

7 The growth associated maintenance energy requirements were  
8 calculated in accordance with protocol procedures [16].

## 9 Transport Reactions

10 The identification of transport reactions involved the two databases for  
11 transport proteins, the Transporter Classification Database (TCDB) [84]  
12 and the Transport Database (TransportDB) [71], and two computational  
13 tools, PSORTb v3.0 [69] and BLASTp [70]. We employed the following  
14 semi-automatic way of inferring the putative function of transport  
15 proteins in *M. capsulatus*.

16 Using the protein sequences in AE017282.2\_protein.faa [22], we  
17 determined the subcellular location of each protein using PSORTb v3.0.  
18 We filtered the results and focused only on proteins with a final score  
19 larger than 7.5, which the authors of PSORTb consider to be  
20 meaningful. We combined this list with the *M. capsulatus* specific  
21 entries from the TransportDB, which allowed us to use the PSORT-

1 scores as an additional measure of confidence. At this point, 242  
 2 putative transport proteins were identified. From this list we then  
 3 selected all proteins which were predicted to transport metabolites and  
 4 were already included in the model, which reduced the number to 133.  
 5 Since for many of the entries, the exact mechanism of transport is  
 6 unknown, we combined the previously selected transporters with the  
 7 results from running BLAST against the TCDB. The e-value and  
 8 bitscore from BLAST provided yet another measure to confidently  
 9 assess the correctness of the automatic TransportDB predictions, and  
 10 the Transporter Classification-Numbers (TC-numbers) allowed us to  
 11 gather information on the mechanism of transport. This led to a list of 97  
 12 transport proteins with high confidence, which was filtered once more as  
 13 follows.

14 We checked the general description for a given specific TC-number,  
 15 and then considered the BLAST result to read about a given transporter  
 16 in more detail, especially with regards to finding the specific substrates.  
 17 When we were able to identify the corresponding transport reaction in  
 18 the BiGG database, we incorporated only the simplest, smallest set of  
 19 reactions. In cases of conflicts between our own BLAST results and the  
 20 automatic TransportDB transporter annotation, we preferentially trusted  
 21 the BLAST results. Thus we ultimately added 75 transporter-encoding  
 22 genes connected via GPR to 56 distinct transport reactions.



# 1 Ratio Constraint

2 To identify which mode of electron transfer is active in *M. capsulatus*,  
 3 we fit the solutions of FBA using iMcBath to measured values [19]. The  
 4 authors had experimentally determined the O<sub>2</sub>/CH<sub>4</sub> ratio and the growth  
 5 yield of *M. capsulatus* in several conditions. They varied the nitrogen  
 6 source using KNO<sub>3</sub>, NH<sub>4</sub>CL, and both simultaneously; the concentration  
 7 of copper in the medium, which directly affects the activity of either  
 8 sMMO or pMMO; and whether the culture was oxygen or carbon limited.

9 Since each electron transfer mode respectively is represented by the  
 10 flux through one specific reaction in the model (PMMOipp,  
 11 PMMODCipp, CYOR\_q8ppi), we were able to investigate each simply  
 12 by constraining the corresponding reaction.

13 Secondly, we accounted for the differential expression of ADH in the  
 14 presence of excess NH<sub>4</sub> in the medium versus the GS/GOGAT in the  
 15 presence of NO<sub>4</sub> or N<sub>2</sub> by blocking the corresponding reactions  
 16 accordingly [53].

17 Several studies have shown that NH<sub>4</sub> is a co-metabolic substrate to the  
 18 methane monooxygenases in *M. capsulatus* leading to the production of  
 19 hydroxylamine first and nitrite later [54, 74, 85]. Hence, when simulating  
 20 the growth on NH<sub>4</sub> we assumed that varying ratios *r* of the nitrogen  
 21 taken up would eventually be converted into nitrite (Figure 7).

$$(1) \quad v_{EX\_nh4\_e} + r v_{EX\_no2\_e} = 0$$

## 2 Stoichiometric Modeling

3 The reactome of an organism can be represented mathematically as a  
 4 stoichiometric matrix  $S$ . Each row of  $S$  corresponds to a unique  
 5 compound, while each column corresponds to a metabolic reaction.  
 6 Hence the structure of the matrix for an organism with  $m$  compounds  
 7 and  $n$  reactions equals  $m \times n$ . The values in each row denote the  
 8 stoichiometric coefficients of that metabolite in all reactions. The  
 9 coefficients are either negative for compounds that are educts, positive  
 10 for those that are products, or zero for those that are not involved in a  
 11 given metabolic reaction.

12 The vector  $v$  of length  $n$  contains as values the turnover rates of each  
 13 reaction. These rates are commonly referred to as fluxes and are given  
 14 the unit  $\text{mmol gDW}^{-1} \text{ h}^{-1}$ . Vector  $v$  is also referred to as the flux vector,

## 15 Declarations

16 Ethics approval and consent to participate:

17 “Not applicable”

1    Consent for publication:

2    “Not applicable”

3    Availability of Data and Materials:

4    The metabolic model, scripts and corresponding datasets generated  
5    and/or analysed during the current study are available in the GitHub  
6    repository.

7    • Archived:

8    <https://github.com/ChristianLieven/memote-m-capsulatus>

9    • Most Recent:

10

11    The data that support the biomass equation constructed in this study  
12    are available from Unibio at

13    • <http://www.unibio.dk/end-product/chemical-composition-1>

14    • <http://www.unibio.dk/end-product/chemical-composition-2>

15    Restrictions may apply to the availability of these data. Data are  
16    however available from the authors upon reasonable request and with  
17    permission of Unibio.

18    Competing interests:

19    Unibio is a collaborator in the “Environmentally Friendly Protein  
20    Production (EFPro2)” project. CL, MJH, and NS are publicly funded

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2 Denmark and thus declare no financial or commercial conflict of  
3 interest.

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7 Production (EFPro2)”)

#### 8 Authors’ contributions:

9 CL collected and analysed the data, reconstructed the metabolic model,  
10 drafted and revised the manuscript. LP and SBJ provided feedback on  
11 the metabolic behaviour of *M. capsulatus*, discussed improvements to  
12 the model and revised the manuscript. KVG, MJH and NS conceived  
13 and supervised the study and revised the manuscript critically. All  
14 authors read and approved the manuscript.

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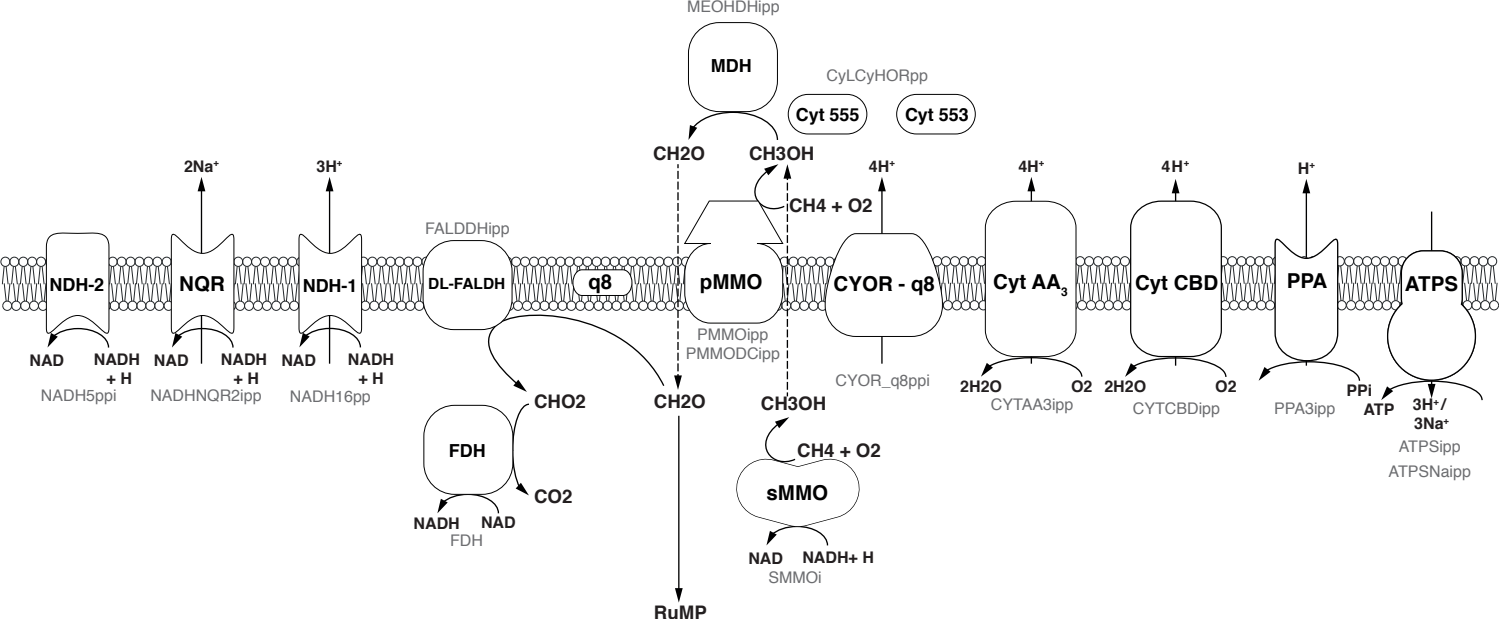
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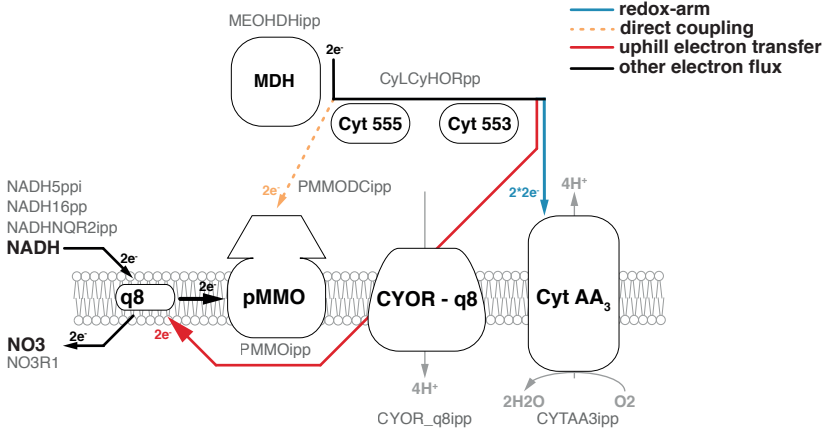


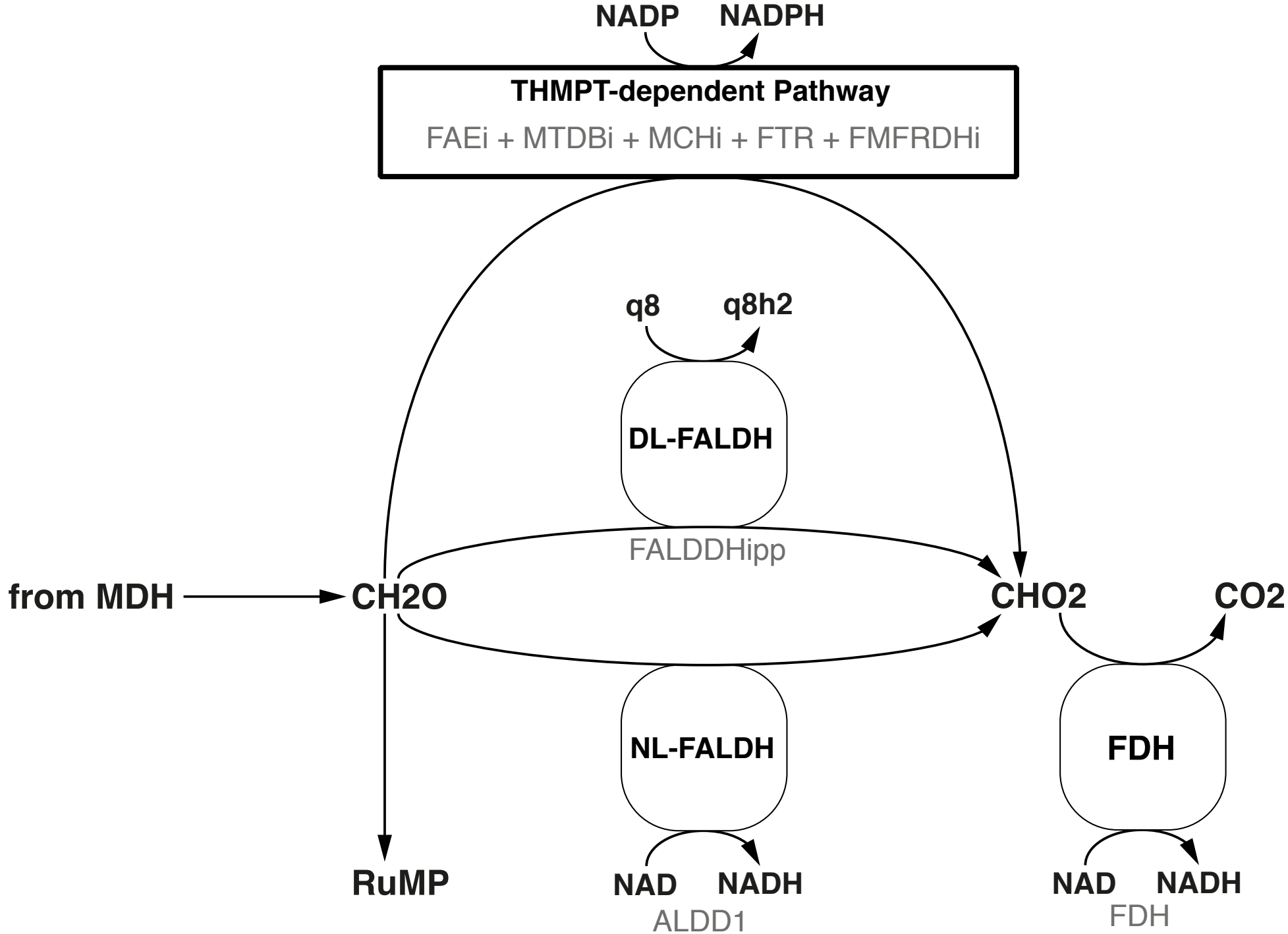
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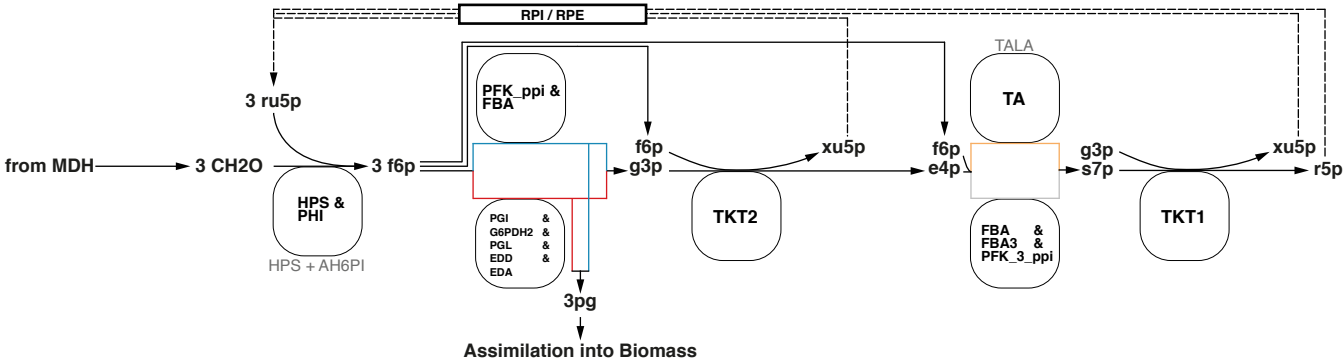
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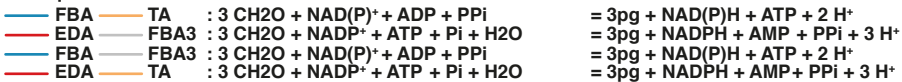


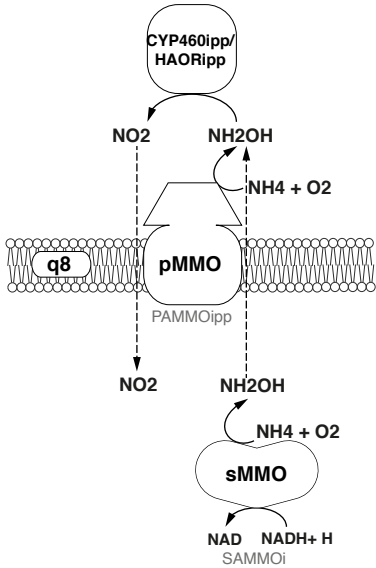


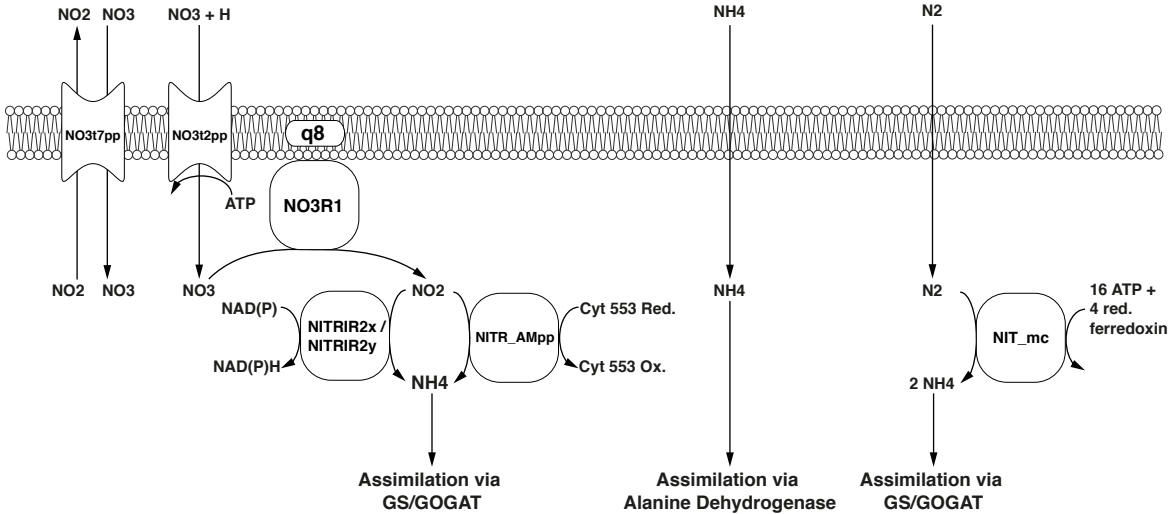




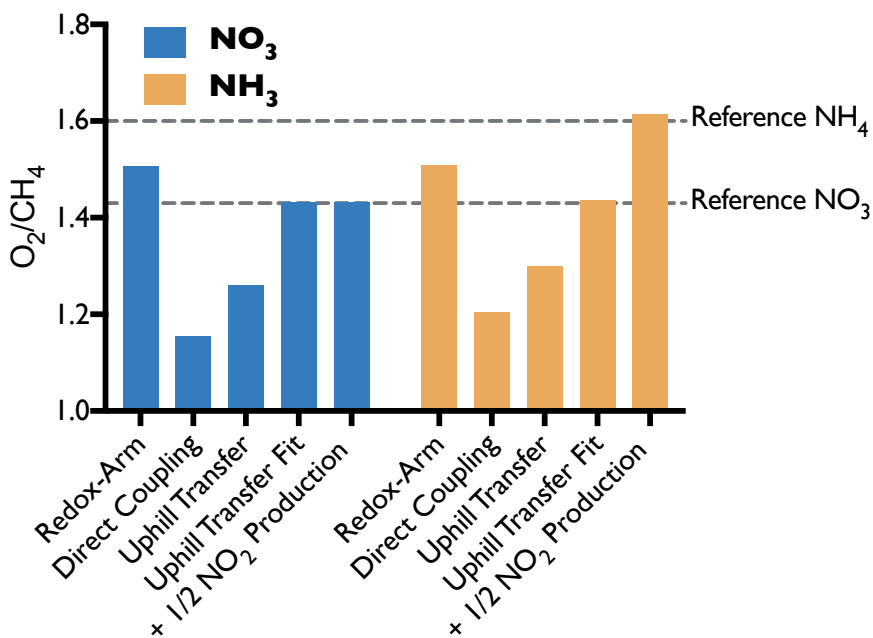
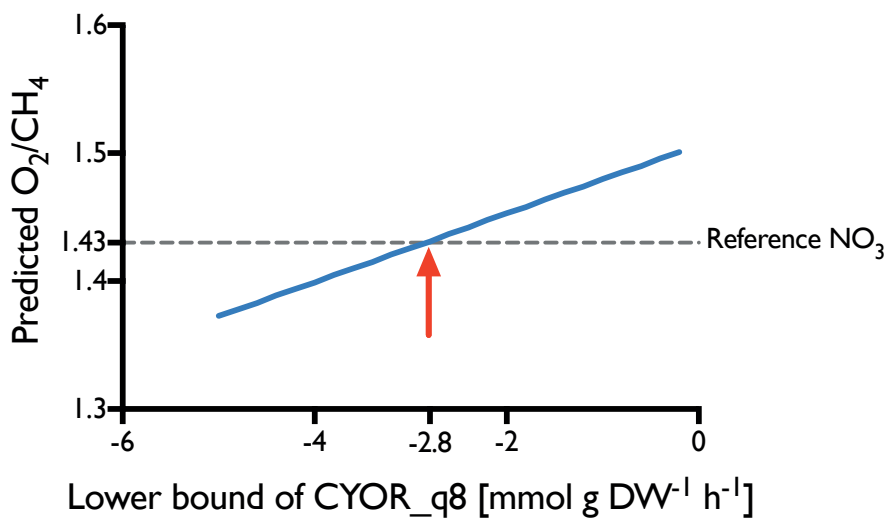
Net equation for each combination in the model:









**A****B****C**