

## **Soil N<sub>2</sub>O emission potential falls along a denitrification phenotype gradient linked to differences in microbiome, rainfall and carbon availability**

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### **Highlights**

- N<sub>2</sub>O emission potential is linked to microbiome changes associated with rainfall, but not to pH.
- Sequential vs. concurrent denitrification phenotypes differing in NO and N<sub>2</sub>O accumulation are identified.
- High N<sub>2</sub>O accumulation is associated with increased NO accumulation.
- Sequentiality of N<sub>2</sub>O production/reduction determines soil N<sub>2</sub>O emission potential.
- Sequentiality of N<sub>2</sub>O reduction was susceptible to manipulation via carbon addition.

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2 **linked to differences in microbiome, rainfall and carbon availability**

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15 **Keywords:** Denitrification phenotype, N<sub>2</sub>O emission potential, microbiome, rainfall,  
16 carbon availability, *nosZ*

## 17 Abstract

18 Soil denitrification produces the potent greenhouse gas nitrous oxide ( $N_2O$ ) and by further  
19 reduction of  $N_2O$ , the harmless inert gas  $N_2$ .  $N_2O$  emission is determined by rate and timing  
20 of the  $N_2O$  producing and reducing steps which are sensitive to a series of proximal and  
21 distal regulators such as pH and microbial community composition. Microbial community  
22 associations to  $N_2O$  emission potential ( $N_2O/(N_2O+N_2)$ ) are commonly entangled with pH  
23 leaving the true role of community composition unclear. Here, we leverage a set of soil  
24 microbiomes strongly linked to rainfall above pH to test the hypothesis that microbiome vs.  
25  $N_2O$  emission potential ( $N_2O/(N_2O+N_2)$ ) correlations will be maintained across alternative  
26 distal drivers.  $N_2O$  emission potential ( $N_2O/(N_2O+N_2)$ ) and denitrification gas (NO,  $N_2O$ ,  
27  $N_2$ ) kinetics were assessed by automated gas chromatography while community  
28 composition was assessed by 16S rRNA gene sequencing and qPCR of *nosZI* and *II* genes.  
29 Analyses revealed a sustained correlation between microbiome and  $N_2O$  emission potential  
30 ( $N_2O/(N_2O+N_2)$ ) in the absence of a pH effect. Further, a continuum of gas accumulation  
31 phenotypes linked to NO accumulation and sensitive to carbon addition are identified.  
32 Separate phenotypes carried out  $N_2O$  production and reduction steps more concurrently or  
33 sequentially and thus determined  $N_2O$  accumulation and emission potential  
34 ( $N_2O/(N_2O+N_2)$ ). Concurrent  $N_2O$  producing/reducing soils typically contained NO  
35 accumulation to a low steady state, while carbon addition manipulations which increased  
36 NO accumulation also increased sequentiality of  $N_2O$  production/reduction and thus  
37 emission potential ( $N_2O/(N_2O+N_2)$ ). These features may indicate a conserved NO  
38 inhibitory mechanism across multiple effectors (rainfall, community composition, carbon  
39 availability).

## 40 Introduction

41 Production and emission of nitrous oxide ( $\text{N}_2\text{O}$ ) represents a significant climate  
42 concern due to its high global warming potential (298 times that of  $\text{CO}_2$  over a 100 year  
43 time span on a mass to mass basis) (Myhre *et al.*, 2013) and ozone depleting activity  
44 (Ravishankara *et al.*, 2009). The most recent IPCC report ranks  $\text{N}_2\text{O}$  as the third most  
45 significant greenhouse gas, accounting for 6.2% of global climate forcing  
46 (Intergovernmental Panel on Climate Change, 2013). Atmospheric concentrations of  $\text{N}_2\text{O}$   
47 have risen dramatically over the past century to a current concentration of greater than 333  
48 ppb (Jan, 2020; 2° Institute, 2016), much of which is attributed to anthropogenic soil  
49 emissions (Davidson, 2009). Global  $\text{N}_2\text{O}$  budgets suggest that around 45% of the emitted  
50  $\text{N}_2\text{O}$  is produced anthropogenically with the majority (60%) coming from agricultural  
51 sources (Syakila and Kroeze, 2011). In an agricultural setting,  $\text{N}_2\text{O}$  production is  
52 traditionally attributed to denitrification and nitrification (Bremner, 1997) of N in animal  
53 excreta or applied fertilizers (Davidson, 2009; Syakila and Kroeze, 2011; Oenema *et al.*,  
54 2005) but a number of other biological processes are also relevant (Baggs, 2011).

55 Denitrification occurs under anoxic conditions when microbial populations switch  
56 from  $\text{O}_2$  based respiration to reduction of nitrogenous molecules ( $\text{NO}_3^- \rightarrow \text{NO}_2^- \rightarrow \text{NO} \rightarrow$   
57  $\text{N}_2\text{O} \rightarrow \text{N}_2$ ). In each step, reduction of the nitrogenous molecule as a terminal electron  
58 acceptor is catalyzed by an independent reductase enzyme (nitrate reductase-Nar or Nap,  
59 nitrite reductase-Nir, nitric oxide reductase-Nor, and nitrous oxide reductase-Nos) (Zumft,  
60 1997). The last step in the process,  $\text{N}_2\text{O}$  reduction, is an important focus in denitrification  
61 and greenhouse gas research (Jones *et al.*, 2014; Liu *et al.*, 2014; Richardson *et al.*, 2009)  
62 as it determines whether the final gaseous product of denitrification is the greenhouse gas  
63  $\text{N}_2\text{O}$  or the harmless inert gas  $\text{N}_2$ . In fact,  $\text{N}_2\text{O}$  reductase is the only known biological sink  
64 of  $\text{N}_2\text{O}$  (Thomson *et al.*, 2012), therefore, encouraging complete denitrification at the time

65 of  $\text{N}_2\text{O}$  production represents an important strategy to preventing further rise in  
66 atmospheric concentrations (Richardson *et al.*, 2009). In reality,  $\text{N}_2\text{O}$  vs.  $\text{N}_2$  production is  
67 not binary (only  $\text{N}_2\text{O}$  or  $\text{N}_2$  produced) and  $\text{N}_2\text{O}$  to  $\text{N}_2$  product ratios depend on a great  
68 number of factors including pH (Simek and Cooper, 2002), carbon and nitrate ( $\text{NO}_3^-$ )  
69 availability (Senbayram *et al.*, 2012), as well as nitrite ( $\text{NO}_2^-$ ) (Firestone *et al.*, 1979;  
70 Gaskell *et al.*, 1981).

71 Conceptually, factors affecting  $\text{N}_2\text{O}$  emission ratios can be separated into i) microbial  
72 community genetic potential for each denitrification step, ii) distal factors, determining that  
73 genetic potential in the long term, and iii) proximal factors, acting within genetic potential  
74 on short term time scales to impact instantaneous denitrification rates (e.g. carbon and  $\text{NO}_3^-$   
75 concentrations), (Wallenstein *et al.*, 2006; Groffman *et al.*, 1988). There has been some  
76 debate over the relative importance of these factors and disentangling their effects can be  
77 difficult when factors such as pH have both immediate effects on enzymatic activity during  
78 denitrification and distal effects on denitrification potential (Samad *et al.*, 2016b).

79 The effect of pH on soil  $\text{N}_2\text{O}$  emissions is well documented (Simek and Cooper,  
80 2002). Low soil pH results in higher soil  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  ratios, most clearly demonstrated  
81 in pH manipulations of soils from the same site (Čuhel *et al.*, 2010; Liu *et al.*, 2010; Simek  
82 and Cooper, 2002), but also manifests in differences in  $\text{N}_2\text{O}$  product ratios between sites  
83 (Samad *et al.*, 2016b). Bergaust *et al.* (2010) showed evidence for a post-transcriptional  
84 effect on the formation of functional  $\text{N}_2\text{O}$  reductase at low pH in pure culture experiments,  
85 possibly due to impeded assembly of this periplasmic enzyme at low pH. A similar post  
86 transcriptional phenomenon was supported using microbial consortia extracted from soils  
87 with different native pH (Liu *et al.*, 2014). Contrastingly, some studies have demonstrated  
88 that pH effects on  $\text{N}_2\text{O}$  reduction are dependent on the concentration of  $\text{NO}_3^-$  or  $\text{NO}_2^-$   
89 (Blackmer and Bremner, 1978; Firestone *et al.*, 1979; Gaskell *et al.*, 1981). Blackmer and

90 Bremner (1978) showed that pH had negligible effects on  $\text{N}_2\text{O}$  reduction activity of soils in  
91 the absence of supplied  $\text{NO}_3^-$  while other studies observed an increased inhibitory impact  
92 of  $\text{NO}_3^-$  or  $\text{NO}_2^-$  under decreasing pH (Firestone *et al.*, 1979; Gaskell *et al.*, 1981). To  
93 confuse matters further, pH probably also has long term distal effects on denitrification  
94 potential due to its well-known impact on microbial community structuring (Lauber *et al.*,  
95 2008; Kaminsky *et al.*, 2017) and probably more specifically the abundance and ratios of  
96 denitrification genes (e.g. Samad *et al.*, 2016b; Domeignoz-Horta *et al.*, 2015; Jones *et al.*,  
97 2014).

98 The functional and taxonomic composition of denitrifier communities has become an  
99 important focus of denitrification research due to advances in molecular tools. Denitrifying  
100 microbes may carry all or only some of the full denitrification gene repertoire and therefore  
101 changes in the phylogenetic composition of denitrifier communities can affect the ratio of  
102 genes coding for  $\text{N}_2\text{O}$  reductase to those coding for  $\text{N}_2\text{O}$  producing enzymes, thus  
103 determining the genetic potential for  $\text{N}_2\text{O}$  emission (Graf *et al.*, 2014; Roco *et al.*, 2017).  
104 Graf *et al.* (2014) showed that the organisms carrying the *nosZII* gene encoding nitrous  
105 oxide reductase clade II commonly had a truncated denitrification pathway without the  
106 genes encoding the preceding denitrification steps. Implicit to this finding is the suggestion  
107 they may reduce  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  ratios by acting as  $\text{N}_2\text{O}$  sinks. Jones *et al.* (2014) showed  
108 evidence for  $\text{N}_2\text{O}$  sink capacity related to low *nosZI/nosZII* ratios, though controversial  
109 because the results could also be explained as a direct (proximal) effect of soil pH (Bakken  
110 *et al.*, 2015).  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  product ratios have been linked to differences in overall  
111 microbial community structure as measured by 16S rRNA gene sequencing, suggesting that  
112 this could be used as a predictor of  $\text{N}_2\text{O}$  emission potential (Morales *et al.*, 2014; Samad *et*  
113 *al.*, 2016b). However, it remains unclear whether these correlations indicate a true causal  
114 relationship. For example Samad *et al.* (2016b) linked 16S community composition to soil  
115  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  product ratios but both these measures were also correlated to soil pH.

116 Therefore, the results are possibly explained by the well documented but separate effects of  
117 pH on  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  (Simek and Cooper, 2002) and on microbial community composition  
118 (Lauber *et al.*, 2009; Kaminsky *et al.*, 2017).

119  $\text{N}_2\text{O}$  accumulation during denitrification may also be caused by differential flow of  
120 electrons to the separate N-reductases (Pan *et al.*, 2013). In wastewater treatment, low  
121 carbon (reductant) availability enhances competition for electrons between  $\text{N}_2\text{O}$  and  
122 upstream N-reductases which can result in transient  $\text{N}_2\text{O}$  accumulation (Pan *et al.*, 2013;  
123 Ribera-Guardia *et al.*, 2014). Indeed carbon and substrate availability can affect  
124  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  product ratios in many contexts including soils, though the direction of the  
125 effect is not always consistent (Gillam *et al.*, 2008; Senbayram *et al.*, 2012; Weier *et al.*,  
126 1993).

127 Here, we aimed to re-assess the consistency of previously outlined (Samad *et al.*,  
128 2016b, 2016a) linkages between  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$ , microbial community composition (as  
129 measured by 16S rRNA gene sequencing and qPCR of *nosZ* genes) and pH in a larger  
130 alternative cohort (20 soils) of pasture soils. Previous investigations indicated community  
131 composition in the present soil set was correlated to changes in long term rainfall above pH  
132 and we hypothesized microbiome to  $\text{N}_2\text{O}$  emission potential associations would be  
133 maintained across this alternate distal driver. Anoxic soil incubations revealed contrasting  
134  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  and denitrification phenotypes based on the timing of  $\text{N}_2\text{O}$  reduction which  
135 determined the propensity for soil  $\text{N}_2\text{O}$  emission. We further assessed the phenotypes and  
136  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  ratios in relation to potential proximal controls ( $\text{NO}_3^- + \text{NO}_2^-$  concentration,  
137 carbon availability) based on 2 alternate hypotheses: 1)  $\text{N}_2\text{O}$  reduction activity was  
138 impaired by higher  $\text{NO}_3^- + \text{NO}_2^-$  concentrations. 2) Impaired  $\text{N}_2\text{O}$  reduction in high  
139  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  ratio soils is caused by limited carbon availability and thus electron supply  
140 to  $\text{N}_2\text{O}$  reductase.

141 **Materials and methods**

142 **2.1 Soil collection**

143 Soils were sampled at 20 sites representing sheep, dairy, beef and goat farms across  
144 multiple regions of New Zealand's South Island. Sampling begun on the 2<sup>nd</sup> of September  
145 2016 and continued through until the 8<sup>th</sup> of September. At each site, soil cores (10 cm  
146 depth, 2.5 cm diameter) were sampled at 2.5 m intervals across a 7.5 m transect (0 m, 2.5  
147 m, 5 m, 7.5 m) using a stainless steel auger. Triplicate cores were collected for each  
148 distance and composited in a bag (12 pooled cores) while a fourth core from each distance  
149 was kept separate for molecular analyses (4 cores). If topsoil was less than 10cm deep,  
150 additional cores were taken to make up the volume. Soil cores were stored in partially open  
151 ziplock bags (to prevent anoxia) on ice until sampling was completed. Pooled cores were  
152 homogenized, and worms, insects, grass and large roots removed before field moist storage  
153 at 4°C. Core samples for molecular analysis were immediately frozen at -80°C until DNA  
154 extraction. See Table S1 for basic soil descriptors based on current analyses and data  
155 collected from separate sampling in 2011 (Wakelin *et al.*, 2013).

156 Pooled site cores were transported to the Norwegian University of Life Sciences  
157 (NMBU, Ås, Akershus, Norway), where they were sieved (2mm) and stored at 4°C before  
158 initiating kinetic experiments.

159 **2.2 Nitrate + nitrite measurements**

160 Endogenous soil nitrate + nitrite ( $\text{NO}_3^- + \text{NO}_2^-$ ) content was measured in sieved  
161 pooled soils using 0.2g of each soil with 1mL of 2M KCl extractant in 1.5 mL microfuge  
162 tubes (Lim *et al.*, 2018). Slurries were shaken and spun down at 16000G for 2 minutes  
163 before recovering the supernatant into fresh 1.5mL microfuge tubes.  $\text{NO}_3^- + \text{NO}_2^-$   
164 concentration was quantified by chemical reduction to nitric oxide (NO) followed by

165 chemiluminescent detection as detailed by (Braman and Hendrix, 1989; Lim *et al.*, 2018). In  
166 brief, 10 $\mu$ L of supernatant was introduced into a sealed glass piping system containing a  
167 heated (95°C) acid vanadium chloride solution (50mM VCl<sub>3</sub> in 1M HCl). VCl<sub>3</sub> reacts to  
168 reduce NO<sub>3</sub><sup>-</sup> quantitatively to NO<sub>2</sub><sup>-</sup> before converting it to NO gas. The NO gas was  
169 captured and carried in an N<sub>2</sub> stream to a Sievers Nitric Oxide Analyzer 280i system (GE  
170 Analytical Instruments, Boulder, CO, USA) for quantification. Standard KNO<sub>3</sub> solutions  
171 (100 to 0.01mM) were used to calibrate the area under detection signal peaks allowing  
172 calculation of NO<sub>2</sub><sup>-</sup>+NO<sub>3</sub><sup>-</sup> concentration in soil supernatants. NO<sub>2</sub><sup>-</sup>+NO<sub>3</sub><sup>-</sup> per gram soil or  
173 per  $\mu$ L porewater was back calculated based on KCl dilution factor, and soil dry  
174 weight/moisture weight (as determined gravimetrically below) .

175 **2.3 Nitrate adjustment**

176 Soil moisture was determined gravimetrically by drying 5-10g soil at 60°C for  
177 minimum 24h (ASTM D2216-10, 2010). Prior to incubation, soils were supplemented with  
178 NO<sub>3</sub><sup>-</sup> and ammonium by flooding and draining with a 2mM NH<sub>4</sub>NO<sub>3</sub> solution e.g. (Samad  
179 *et al.*, 2016a; Liu *et al.*, 2010; Qu *et al.*, 2014). NO<sub>3</sub><sup>-</sup> supplied N for denitrification while  
180 ammonium acted as a preferential assimilatory N source. For this, an 80g dry weight  
181 equivalent of soil was placed in a 500mL Sterafil Filter Holder (Merck, Burlington, MA,  
182 USA) and flooded with 300mL of 2mM NH<sub>4</sub>NO<sub>3</sub> (sufficient volume to dilute endogenous  
183 NO<sub>3</sub><sup>-</sup>) . After 15min the solution was drained through a 0.2 $\mu$ M cellulose filter (Merck) with  
184 1.2 $\mu$ M glass-fibre pre filter (Merck) using a vacuum manifold. Soils were mixed and a  
185 subsample was taken for overnight moisture content analysis (5g, as above). The remainder  
186 of the soils was stored overnight in funnels covered with aluminum foil before use in  
187 incubation experiments the next day.

188 **2.4 Gas kinetics measurements**

189           Respiration and denitrification activity of soils was measured by gas chromatography  
190           of headspace gases (O<sub>2</sub>, CO<sub>2</sub>, NO, N<sub>2</sub>O, N<sub>2</sub>) in oxic and anoxic batch incubations  
191           commencing between 39 and 56 days post sampling. The temperature controlled robotic  
192           autosampler, gas chromatograph (Agilent GC -7890A equipped with ECD, TCD, FID) and  
193           chemiluminescence NO<sub>x</sub> analyzer (Model 200A, Advanced Pollution Instrumentation, San  
194           Diego, USA) used here are described in detail by (Molstad *et al.*, 2007; Qu *et al.*, 2014).  
195           The device holds up to 44 sealed 120 mL serum vials in a temperature controlled water  
196           bath. A robotic arm equipped with a hypodermic needle and a peristaltic pump takes  
197           headspace gas samples periodically and pumps them through dedicated sample loops in the  
198           GC. The GC uses helium as carrier gas while subsequent back pumping replaces sampled  
199           gas with helium thus maintaining the pressure in the serum vials at ~1 atm. Dilution and  
200           leakage are back calculated following experiment completion to allow estimation of true  
201           gas production. Here, 20g dry weight equivalent of NH<sub>4</sub>NO<sub>3</sub> adjusted soil was placed in  
202           triplicate 120mL serum vials and crimp sealed with butyl rubber septa. A 15g subsample of  
203           each soil was frozen at the start of each incubation for subsequent measurement of NO<sub>2</sub><sup>-</sup>  
204           +NO<sub>3</sub><sup>-</sup> concentrations as described above. The serum vials were placed in the water bath  
205           (20°C) under the autosampler and allowed to equilibrate before releasing overpressure  
206           through a water filled syringe without piston. Additional vials (duplicates) were filled with  
207           premixed standard gases (including 400ppm CO<sub>2</sub>, 10,000ppm CO<sub>2</sub>, 500ppb N<sub>2</sub>O, 151ppm  
208           N<sub>2</sub>O, 25ppm NO) supplied by AGA industrial gases (Oslo, Akershus, Norway),  
209           compressed air (781,000ppm N<sub>2</sub>, 200900ppm O<sub>2</sub>) or helium. The autosampler was  
210           programmed to take headspace samples every ~5hrs. After 7 rounds of sampling, at ~40hrs,  
211           anoxia was induced by helium flushing the vials (3 cycles of evacuation for 180sec and He-  
212           filling for 20sec) and incubation was continued until most soils had converted all  
213           denitrification products to N<sub>2</sub>. Two separate experiments were required 10 days apart to  
214           process all 20 soils.

215 **2.5 Measures of N<sub>2</sub>O emission potential/kinetics**

216 The N<sub>2</sub>O hypothetically emitted (%) metric was calculated as max  $\mu\text{mol N}_2\text{O-N}$   
217 accumulated in vial expressed as a percentage of final cumulative N (N<sub>2</sub>-N). N<sub>2</sub>O  
218 hypothetically emitted (%) was used to evaluate the sequentiality of N<sub>2</sub>O production and  
219 reduction steps over the course of an incubation and the relative N<sub>2</sub>O emission potential of  
220 each soil. It is equivalent to typical N<sub>2</sub>O product ratios (N<sub>2</sub>O/N<sub>2</sub>O + N<sub>2</sub>) but expressed  
221 relative to total final accumulated N rather than just gaseous N<sub>2</sub>O+N<sub>2</sub> at the timepoint of  
222 calculation. It is our preferred metric because:

223 1) It is event based (i.e. calculated at peak in vial N<sub>2</sub>O). This allows comparison of  
224 soils with divergent denitrification timescales/rates.

225 2) It is a relative measure (i.e. expressed as a percentage of total N finally  
226 accumulated). This allows comparison of soils with contrasting initial NO<sub>3</sub><sup>-</sup> supply  
227 and net denitrification rates.

228 3) It describes N<sub>2</sub>O emission potential (i.e. soils accumulating greater peak N<sub>2</sub>O in  
229 headspace are likely to be higher emitters in situ). This N<sub>2</sub>O is likely to be emitted  
230 in an unsealed environment therefore this metric ignores net N<sub>2</sub>O reconsumption  
231 from the headspace.

232 4) It directly describes the sequentiality of N<sub>2</sub>O production/reduction (i.e. to what  
233 degree N<sub>2</sub>O production and reduction to N<sub>2</sub> were carried out at the same time and to  
234 the same magnitude, thus mitigating N<sub>2</sub>O accumulation). This was highly relevant  
235 to the soil kinetics observed here (see results).

236 Additional emission potential metrics were evaluated to allow comparison with previous  
237 studies. N<sub>2</sub>O/(N<sub>2</sub>O+N<sub>2</sub>) (N<sub>2</sub>O ratio) calculated as  $\mu\text{mol cumulative N}_2\text{O-N}$  (per vial) over  
238 max  $\mu\text{mol cumulative N}_2\text{O-N}$  plus  $\mu\text{mol cumulative N}_2\text{-N}$  at various timepoints

239 (N<sub>2</sub>O/(N<sub>2</sub>O+N<sub>2</sub>) (max N<sub>2</sub>O), N<sub>2</sub>O/(N<sub>2</sub>O+N<sub>2</sub>) (50hrs)) (Samad *et al.*, 2016a). N<sub>2</sub>O index  
240 (I<sub>N<sub>2</sub>O</sub>) as used previously (Liu *et al.*, 2010; Qu *et al.*, 2014; Samad *et al.*, 2016a), calculated  
241 as the area under a N<sub>2</sub>O curve over the area under an N<sub>2</sub>O curve + N<sub>2</sub> curve using the  
242 following formula:  $I_{N_2O} = \int_0^T N_2O(t)dt / \int_0^T [N_2O(t) + N_2(t)]dt$

243 It is useful because it allows a time-integrated view of N<sub>2</sub>O vs. N<sub>2</sub> stoichiometry. Areas are  
244 calculated for each time period between two sampling points (~5hrs) and summed up to an  
245 arbitrary time point (T). Here, we used 50hrs (I<sub>N<sub>2</sub>O</sub> (50hrs)) and once all denitrification gas  
246 was accumulated as N<sub>2</sub> (I<sub>N<sub>2</sub>O</sub> (N<sub>2</sub> plateau)).

247 N<sub>2</sub>O hypothetically emitted (%) value cut-offs were assigned to separate N-gas  
248 accumulation patterns into discrete phenotypes based on N<sub>2</sub>O, NO, and N<sub>2</sub> accumulation  
249 patterns. Sequential soils (88% to 100%) had close to 0 accumulation of N<sub>2</sub> before peak  
250 N<sub>2</sub>O, a sudden increase in N<sub>2</sub> rate was observed after peak N<sub>2</sub>O, and a single often high NO  
251 peak was usually observed. Concurrent soils (0 to 80%) developed N<sub>2</sub> production early  
252 which was either sustained or gradually increased. NO levels were typically controlled  
253 after a brief peak. Intermediate soils (81% to 87%) had features of both phenotypes. NO  
254 was typically poorly controlled, some N<sub>2</sub> production was observed before peak N<sub>2</sub>O but  
255 sharp increases in N<sub>2</sub> rate were observed after peak N<sub>2</sub>O.

## 256 **2.6 pH measurement**

257 For each soil, a 10mL subsample was placed in a plastic container using a volumetric  
258 spoon and 20mL of 10mM CaCl<sub>2</sub> was added. Containers were capped and shaken until the  
259 soil was dispersed in the solution. Soils were left overnight at room temperature. Soils were  
260 re-dispersed by shaking and left to settle for 10min before the pH was measured in the  
261 supernatant using a H170 pH meter (Hach, Loveland, CO, USA).

## 262 **2.7 Rainfall**

263        Average daily rainfall at the sample sites (mm day<sup>-1</sup>) was estimated at various  
264        timescales (month, year, 10 years) using rainfall data from New Zealand's national climate  
265        database. Data were assessed through the CliFlo web system (NIWA, 2017). Collected data  
266        spanned from 08/09/96 to 07/09/2016. We also included rainfall and potential  
267        evapotranspiration estimates from a previous study (Wakelin *et al.*, 2013) on the same soils  
268        (rainfall historical) which used average daily measurements for 5 years prior to sampling.  
269        Values were calculated through interpolations (Cichota *et al.*, 2008; Tait *et al.*, 2005) using  
270        the Virtual Climate Station from NIWA (Wellington, Wellington, New Zealand). Drainage  
271        class (1: very poor to 5: well) was collected from the New Zealand Fundamental soil Layer  
272        (LRIS, 2020).

273        **2.8 DNA extraction**

274        Distance-specific site cores (0m, 2.5m, 5m, 7.5m) were defrosted for DNA extraction.  
275        To test if pooling and sieving was necessary for future sampling, a subsample of 15g fresh  
276        weight from each core was pooled into a mixed sample (mixed-m) and a subsample of this  
277        sample was sieved through a 2mm sieve (mixed and sieved-ms). ~0.25g of each distance  
278        specific sample per site and the two additional m and ms samples per site (6 samples per  
279        site times 20 sites = 120 samples) were extracted using a Powersoil DNA Isolation kit  
280        (Mobio, Carlsbad, CA, USA) according to the standard instructions. Bead beating was  
281        carried out at 1500rpm in two 15sec steps with intermittent cooling using a 1600 MiniG  
282        cell-lyser (SPEXSamplePrep, Metuchen, NJ, USA). DNA extracts were quantified and  
283        quality checked using a Qubit fluorometer (Invitrogen, Carlsbad, CA, USA) with Qubit  
284        dsDNA HS assay (Invitrogen) and NanoDrop One (Thermo Scientific, Waltham,  
285        Massachusetts, USA). pH-CaCl<sub>2</sub> and pH-H<sub>2</sub>O of ms soils was measured as described earlier  
286        (section 2.2.2, pH measurement) but using a MP220 pH meter with Inlab 413 electrode  
287        (Mettler Toledo, Columbus, Ohio, USA).

288 **2.9 16S amplicon sequencing**

289 16S amplicon sequencing was carried out on a single lane of an Illumina HiSeq using  
290 Version 4\_13 of the Earth Microbiome Project standard protocol (Caporaso *et al.*, 2012).  
291 Open reference OTU picking (97% similarity, UCLUST (Edgar, 2010) and taxonomy  
292 assignment (BLAST (Altschul *et al.*, 1990) was carried out in QIIME 1.9.1 (Caporaso *et*  
293 *al.*, 2010) using version 128 of the SILVA database (Quast *et al.*, 2013). Site-specific  
294 sequence pools were then subsampled 10 times to a depth of 37120 sequences. Subsampled  
295 pools were averaged using basic R functions (R Core Team, 2016). NMDS ordinations  
296 (Bray Curtis dissimilarity) were carried out using Phyloseq (McMurdie and Holmes, 2013).  
297 Mantel tests were carried out in Vegan (Dixon, 2003) using a Pearson correlation method.  
298 Sequences have been submitted to the NCBI Sequence Read Archive under accession  
299 numbers SRR11650167 to SRR11650286 and BioProject ID PRJNA629050.

300 **2.10 qPCR**

301 Total prokaryotic abundance, and nitrous oxide reductase gene abundance for Clade I  
302 and II were measured by targeting the 16S rRNA gene and *nosZ* gene respectively, using  
303 the following primer pairs: 16S UNIV F&R (Hartman *et al.*, 2009), nosZ2F & nosZ2R  
304 (Henry *et al.*, 2006) , 1153\_nosZ8F & 1888\_nosZ29R (Jones *et al.*, 2013). Reactions (10 $\mu$ L  
305 total volume) consisted of 10ng soil DNA, forward and reverse primers at a final  
306 concentration of 0.5 $\mu$ M (except for *nosZ* II reactions which included 1 $\mu$ M), 5 $\mu$ L of  
307 Luminaris HiGreen low Rox qPCR Master Mix (Thermo Scientific) and nuclease free  
308 water (Thermo Scientific) to make up the 10 $\mu$ L volume. Minimum triplicate reactions per  
309 sample were performed using a QuantStudio 6 flex qPCR machine (Applied Biosystems,  
310 Foster City, CA, USA) according to the following thermal cycling conditions. 16S: 2 min  
311 UDG pre-treatment at 50°C, 10min initial denaturation at 95°C, 40 cycles of 15sec  
312 denaturation at 95°C, 30sec annealing at 65°C and 30sec extension at 72°C. *nosZ*I: 2 min

313 UDG pre-treatment at 50°C, 10min initial denaturation at 95°C, 40 cycles of 15sec  
314 denaturation at 95°C, 30sec annealing at 58.5°C and 30sec extension at 72°C. *nosZII*  
315 touchdown: 2 min UDG pre-treatment at 50°C, 10min initial denaturation at 95°C, 6 cycles  
316 of amplification decreasing annealing temperature by 1°C per cycle consisting of 15sec  
317 denaturation at 95°C, 30sec annealing at 60-55°C and 30sec extension at 72°C, 44 cycles of  
318 amplification consisting of 15sec denaturation at 95°C, 30sec annealing at 54°C, 30sec  
319 extension at 72°C and 30sec at 80°C for signal detection. All reaction plates included  
320 minimum triplicate no-template controls and a 10-fold dilution series of pGEM-t-easy  
321 (Promega, Madison, WI, USA) cloned standards for the relevant amplicon, encompassing  
322 the sample quantification range. Measurement of the desired amplicon was confirmed by a  
323 melt curve analyses (15sec denaturation at 95°C, 1min 60°C, 30 sec 95°C) following target  
324 amplification.

325 **2.11 Predicted  $\text{NO}_3^- + \text{NO}_2^-$  accumulation and predicted nitrification**

326 Predicted  $\text{NO}_3^- + \text{NO}_2^-$  at the start of anoxia was calculated as the final accumulated  
327  $\mu\text{mol-N}$  accumulated in each vial at the end of the incubation. Predicted nitrification during  
328 the oxic period was calculated as the difference between measured  $\text{NO}_3^- + \text{NO}_2^-$  at the  
329 beginning of the experiment and predicted  $\text{NO}_3^- + \text{NO}_2^-$  at the start of anoxia.

330 **2.12 Carbon amendment experiment**

331 An independent experiment was set up as described above with some modifications  
332 to test the influence of carbon availability on gas kinetics. Incubations commenced 3  
333 months after initial sampling. Five soils were selected based on covering a range of  $\text{N}_2\text{O}$   
334 ratios (40-Fairlie Geraldine, 20-Waitaha, 1-Woodend, 33-Rae's Junction, 5-Waipapa). In  
335 these incubations, an initial period of oxic storage and incubation was not carried out and  
336 the concentration of  $\text{NH}_4\text{NO}_3$  in flooding solutions was increased (4mM) to account for  
337 extra  $\text{NO}_3^-$  accumulated during the oxic period in the original incubations. Incubations were

338 monitored under two treatment conditions: 4mM NH<sub>4</sub>NO<sub>3</sub>, ± 10mM sodium glutamate as a  
339 carbon source. Glutamate can be utilised by most bacteria and in addition could provide a  
340 preferential organic N source preventing NO<sub>3</sub><sup>-</sup> assimilation (Halvorson, 1972). Sodium  
341 glutamate solutions were pH adjusted to the soils' native pH using HCl.

342 **2.13 Statistical analyses**

343 Incubation kinetics variables are presented and used in correlations as the mean of  
344 triplicate incubations or duplicate incubations in cases where a replicate had to be dropped  
345 due to gas leakage. Spearman's ranked correlations between incubation kinetic measures  
346 and soil variables were used based on non-normaly distributed data. Discrete phenotypic  
347 groups were compared to incubation variables using a Wilcoxon rank sum test (differences  
348 of medians) with chi-squared approximation (but are also supported with continuous/ranked  
349 analyses). NMDS plots were evaluated based on stress <0.2. NMDS plots are presented for  
350 a single pooled sample per soil for appropriate statistical comparison to gas kinetic and  
351 environmental variables but ordinations with full distance specific replicates are available  
352 (Figure S3). Bray Curtis dissimilarity matrixes were compared against incubation and  
353 environmental variables by Mantel test using a pearson correlation co-efficient.. NMDS  
354 axis 1 and 2 co-ordinates were also extracted and tested against incubation and  
355 environmental variables using a spearman's ranked correlation to identify variables  
356 associated with a particular axis. Multiple linear regression for prediction of N<sub>2</sub>O  
357 hypothetically emitted (%) using rainfall, soil drainage class and potential  
358 evapotranspiration was performed using standard least squares.

359 **Results**

360 **3.1 Denitrification gas kinetics**

361 We monitored gas production ( $\text{CO}_2$ ,  $\text{NO}$ ,  $\text{N}_2\text{O}$ ,  $\text{N}_2$ ) from  $\text{NH}_4\text{NO}_3$  amended soil  
362 incubations to identify soils with contrasting denitrification gas production kinetics and  
363 potential for  $\text{N}_2\text{O}$  emission (key incubation variables available in Table S2). Soils were  
364 initially incubated under oxic conditions (40hrs) to identify their aerobic respiratory  
365 potential. Soil  $\text{CO}_2$  production was in the range of 1 to  $5\mu\text{mol hr}^{-1} \text{ vial}^{-1}$  (mean  $\pm$  SD = 3.43  
366  $\pm$  2.89) with the exception of one soil (18-Kumara, a flipped pasture) that had a production  
367 rate of  $15.21\mu\text{mol hr}^{-1} \text{ vial}^{-1}$ .

368 During the subsequent anoxic incubation period, soils varied greatly in the timing of  
369  $\text{N}_2\text{O}$  production and further reduction to  $\text{N}_2$ : while some soils carried out concurrent  $\text{N}_2\text{O}$   
370 production/reduction from the onset of anoxia, others carried out each step sequentially,  
371 accumulating most N as  $\text{N}_2\text{O}$  before converting it stoichiometrically to  $\text{N}_2$  (Figure 1). We  
372 evaluated the sequentiality of  $\text{N}_2\text{O}$  production/reduction on a continuous scale using  $\text{N}_2\text{O}$   
373 hypothetically emitted (%) and applied somewhat arbitrary cutoffs (see section 2.5.) to  
374 place each soil in discrete phenotypic groups: concurrent (0 to 80%), intermediate (81% to  
375 87%) and sequential (88% to 100%). In addition to timing of  $\text{N}_2\text{O}$  production/reduction,  
376 alternative phenotypic groups had contrasting NO accumulation patterns: more concurrent  
377  $\text{N}_2\text{O}$  producing/reducing soils accumulated far less NO (Spearman's correlation, average  
378  $\mu\text{mol NO vial}^{-1}$  vs.  $\text{N}_2\text{O}$  hypothetically emitted %,  $\rho=0.80$ ,  $p<0.0001$ ), and most displayed a  
379 very low pseudo steady state NO level after a brief peak in accumulation (Figure 1A).

380  $\text{N}_2\text{O}$  hypothetically emitted (%) was also used to evaluate soil  $\text{N}_2\text{O}$  emission potential  
381 as sequentiality of  $\text{N}_2\text{O}$  production/reduction determines  $\text{N}_2\text{O}$  accumulation, while omitting  
382 the reconsumption of headspace  $\text{N}_2\text{O}$  predominant late in sequential soils which is more

383 likely to be emitted in an unsealed environment. Alternative measures of N<sub>2</sub>O emission  
384 potential were also evaluated to maintain comparability with previous studies and gave  
385 similar soil rankings (Spearman's correlation vs. N<sub>2</sub>O hypothetically emitted %,  
386 N<sub>2</sub>O/(N<sub>2</sub>O+N<sub>2</sub>) (max N<sub>2</sub>O):  $\rho=0.94$ ,  $p<0.0001$ , N<sub>2</sub>O/(N<sub>2</sub>O+N<sub>2</sub>) (50hrs):  $\rho=0.71$ ,  $p<0.001$ ,  
387 I<sub>N2O</sub> (N<sub>2</sub> plateau):  $\rho=0.79$ ,  $p<0.0001$ , I<sub>N2O</sub> (50hrs):  $\rho=0.53$ ,  $p<0.05$ ).

388 **3.2 Interaction between N<sub>2</sub>O emission potential, pH and community composition**

389 We hypothesized that the observed kinetic patterns/N<sub>2</sub>O emission potentials were  
390 linked to differences in pH and community differences based on previously observed  
391 linkages between N<sub>2</sub>O product ratios, pH and 16S microbial community composition using  
392 the same incubation methodology (Samad *et al.*, 2016b). N<sub>2</sub>O hypothetically emitted (%)  
393 was not correlated with soil pH (Spearman's correlation, measures of N<sub>2</sub>O emission vs. pH  
394 (CaCl<sub>2</sub> or H<sub>2</sub>O),  $p > 0.05$ ), but did map to differences in 16S community composition across  
395 axis 1 in NMDS plots (Table 1, Spearman's correlation NMDS axis 1, Figure 2A).  
396 However, significant correlation to the full dissimilarity matrix based on Mantel tests  
397 (Table 1) was not achieved unless all distance specific replications were included in  
398 analyses (Figure S3). Similar trends were observed for alternative emission potential  
399 metrics (Table 1). pH and long-term average daily rainfall were identified as potential  
400 drivers of differences in microbial community composition. Both were significantly  
401 correlated to overall changes in the dissimilarity matrix, and mapped primarily to NMDS  
402 axis 1 and 2 respectively (Table 1, Figure 2B)

403 We also measured *nosZII* gene copy numbers as a community related functional  
404 metric as it has been suggested *nosZII* carrying organisms are important for soil N<sub>2</sub>O  
405 reduction activity. *nosZII* copy number expressed in various forms (numbers per ng soil  
406 DNA, per gram of soil, normalized to 16S copy numbers and relative to *nosZI* copy  
407 numbers) were not significantly correlated to N<sub>2</sub>O hypothetically emitted % or other

408 measures of emission potential ( $p>0.05$ , Spearman's correlation). *nosZII* copy numbers  
409 were most strongly correlated with soil pH CaCl<sub>2</sub> (Spearman's correlation,  $\rho=0.60$ ,  $p<0.01$ )  
410 and interestingly, were ~10 fold higher in abundance than *nosZI* (Figure 2C).

411 **3.3 Rainfall**

412 In addition to community associations, we assessed the direct relationship between rainfall  
413 and soil phenotypes/N<sub>2</sub>O emission potential. More concurrent N<sub>2</sub>O production/reduction  
414 was associated with higher long-term average daily rainfall (Figure 3A). The strength of the  
415 correlation was best (and highly significant;  $p<0.01$ ) when rainfall regime was averaged  
416 over a prior year or decade, and was not significant when averaged over shorter time span  
417 (Table S3). Linear regression of rainfall (10 years) and hypothetically emitted N<sub>2</sub>O (%)  
418 poorly recapitulated the trends observed in non-parametric and non-continuous analyses  
419 (Figure 3B), probably due to the high variability in average daily rainfall among low N<sub>2</sub>O  
420 emitting soils. Correlations between all rainfall and N<sub>2</sub>O emission potential metrics are  
421 found in Table S3. Multiple linear regression indicated drainage class and potential  
422 evapotranspiration did not aid prediction of N<sub>2</sub>O hypothetically emitted (%) with only  
423 rainfall producing a significant parameter effect ( $p = 0.03$ , Table S4).

424 **3.4 Nitrate + nitrite**

425 Final cumulated N<sub>2</sub> levels per vial were inconsistent between soils suggesting  
426 significant variation in NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> concentrations upon initiation of the anoxic incubation  
427 period (Figure 4A). Further, comparison of measured soil NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> before incubation  
428 and at the start of the anoxic incubation estimated from cumulative denitrified N (Figure  
429 4A) suggested NO<sub>3</sub><sup>-</sup> or NO<sub>2</sub><sup>-</sup> was accumulated during the oxic incubation period,  
430 presumably due to nitrification of added ammonium. We investigated the potential impact  
431 of these variable initial NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> concentrations as it has previously been demonstrated  
432 that N<sub>2</sub>O reduction activity is sensitive to NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> concentration. Predicted NO<sub>3</sub><sup>-</sup> +

433 NO<sub>2</sub><sup>-</sup> porewater concentrations at the start of the anoxic period were not significantly  
434 correlated to N<sub>2</sub>O hypothetically emitted %, however, significant positive correlations were  
435 observed for some alternative measures of N<sub>2</sub>O emission potential (Table 2).

436 Normalized N<sub>2</sub> production rates (% of maximum) were plotted against residual NO<sub>3</sub><sup>-</sup>  
437 + NO<sub>2</sub><sup>-</sup> concentrations estimated from denitrification progress at different stages during the  
438 anoxic incubation to allow comparison of soil N<sub>2</sub>O reduction rates at similar NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup>  
439 concentrations (Figure 4B, C). Concurrent soils showed greater % N<sub>2</sub> production rate at  
440 greater NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> levels while most sequential phenotype soils maintained near zero N<sub>2</sub>  
441 production rates until NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> fell below 20-10 μmol (Figure 4C). However, it should  
442 not necessarily be concluded that different sensitivities of N<sub>2</sub>O reduction to NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup>  
443 explain the variation in soil N<sub>2</sub>O reduction/production phenotype, due to confounding by  
444 time, N denitrified, the natural progression of denitrification and other unknown factors.  
445 Sensitivity of N<sub>2</sub>O reduction to NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> has previously been explained by pH,  
446 therefore we overlayed pH onto plots but did not find an explanatory pattern (Figure S4).

447 **3.5 Carbon supplementation**

448 We hypothesized that differences in apparent denitrification phenotypes resulted from  
449 electron competition under carbon-limited conditions between earlier steps of  
450 denitrification and N<sub>2</sub>O reduction. In further incubations, soils representing a range of  
451 phenotypes/N<sub>2</sub>O emission potentials were amended with both glutamate (to relieve  
452 potential carbon limitation) and NH<sub>4</sub>NO<sub>3</sub> (to provide NO<sub>3</sub><sup>-</sup> for denitrification) or control  
453 vials with NH<sub>4</sub>NO<sub>3</sub> alone. In most cases, carbon negative controls (Figure 5, left panel)  
454 recapitulated the general phenotypic trends observed in initial incubations (Figure 1) but  
455 there were some large observable differences, possibly caused by changes to the incubation  
456 preparation methodology (omitted oxic preincubation, increased added NH<sub>4</sub>NO<sub>3</sub> to 4mM).  
457 In particular, soil 40-Fairlie-Geraldine (Figure 5A, top-left) showed a “weakened”

458 concurrent phenotype, compared with original incubations ( $\text{N}_2\text{O}$  hypothetically emitted %,  
459 0.41 in original vs. 0.80 in second incubation). Differences for other soils were much less  
460 dramatic (Table 3). All +N treatments also had higher  $\text{CO}_2$  production rates and  
461 denitrification process rates, on average 1.44 (average  $\text{CO}_2$  production rate) and 1.38 (max  
462  $\text{N}_2\text{O}$  production rate) times higher respectively, compared with original incubations. This  
463 may suggest changes made to incubation methodology resulted in higher respiration and  
464 probably more available soil carbon.

465 Carbon amendment had variable impacts on denitrification phenotype, respiration and  
466 denitrification rates depending on the soil (Figure 5, Table 3). Added carbon clearly  
467 relieved some limitation as we observed increased  $\text{CO}_2$  production (Table 3) and reduced  
468 time to complete denitrification in all soils (Figure 5), but impacts on denitrification  
469 phenotype were not in line with our original hypothesis. Carbon amendments drove  
470 concurrent soils (40-Fairlie-Geraldine, 20-Waitaha Valley) towards a more sequential  
471 phenotype (increased  $\text{N}_2\text{O}$  hypothetically emitted % and max NO accumulation compared  
472 with +N controls, Table 3) while sequential soils (1-Woodend, 33-Rae's Junction)  
473 maintained their sequential phenotype (similar  $\text{N}_2\text{O}$  hypothetically emitted % and max NO  
474 accumulation compared with +N controls, Table 3). Our C amended intermediate soil 5-  
475 Waipapa did not appear to respond in the same way as other soils. The soil accumulated  
476 less NO than the N amended control (Difference max NO, 3.26 $\mu\text{mol}$ ) and showed a  
477 variable response in measures of  $\text{N}_2\text{O}$  emission potential (Table 3).

478 **Discussion**

479 The initial incubation experiment unexpectedly revealed a continuum of soil  
480 denitrification phenotypes based on the timing of  $\text{N}_2\text{O}$  reduction/production. The most  
481 striking soils (Figure 1C) carried out  $\text{N}_2\text{O}$  production and reduction steps almost entirely  
482 sequentially, accumulating most N as  $\text{N}_2\text{O}$  in vial headspace before initiating rapid  $\text{N}_2\text{O}$   
483 reduction. In an open vial or pasture soil, this emission pattern is predicted to result in up to  
484 100% emission of produced  $\text{N}_2\text{O}$ , depending on soil physical properties (e.g. depth, water  
485 filled porosity) determining the ability of delayed  $\text{N}_2\text{O}$  reduction to transform  $\text{N}_2\text{O}$  before  
486 emission. In addition to our initial aim of re-assessing previously observed links between  
487  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$ , pH and microbial community composition (Samad *et al.*, 2016b, 2016a),  
488 we explored the potential causes of these contrasting  $\text{N}_2\text{O}$  production/reduction phenotypes  
489 which are hypothesized to be due to a transient mechanism of action, potentially a  
490 reversible inhibition or regulatory process.

491 **4.1 The role of pH**

492 The correlation between low pH and high  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  ratio is well documented and has  
493 been demonstrated in a variety of experimental systems (Bergaust *et al.*, 2010; Liu *et al.*,  
494 2014; Samad *et al.*, 2016a). As such, we were surprised to find that pH was not correlated  
495 to measures of  $\text{N}_2\text{O}$  emission potential in the present study. It is possible that variability of  
496 other factors influencing  $\text{N}_2\text{O}$  emission ratios overshadowed a pH effect in this particular  
497 data. pH correlations with  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  ratios are most often observed through variable  
498 pH manipulation within a single site or soil e.g. (Čuhel *et al.*, 2010; Liu *et al.*, 2010; Simek  
499 and Cooper, 2002), while the studied soils were from varying geographical locations with  
500 variable management.

501        Alternatively, unobserved pH changes before the anoxic incubation period could  
502        obscure a true correlation. Comparisons between measured soil  $\text{NO}_3^- + \text{NO}_2^-$  before oxic  
503        incubation and predicted  $\text{NO}_3^- + \text{NO}_2^-$  at the beginning of the anoxic period (based on final  
504         $\text{N}_2$  accumulated) showed nitrification must have occurred in most soils while they were  
505        under oxic conditions (Figure 4A). Nitrification of ammonium results in the release of two  
506         $\text{H}^+$  ions per molecule of ammonium oxidized (Rowell and Wild, 1985; Zhao *et al.*, 2014)  
507        and therefore could have caused significant acidification of incubated soils in the present  
508        study, making initial pH measurements irrelevant. Various lines of evidence seem to  
509        counter this hypothesis:

510        1. There was no correlation between nitrification activity, predicted from the  
511        difference in initial measured  $\text{NO}_3^- + \text{NO}_2^-$  vs. estimated at the start of the anoxic  
512        incubation and  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  suggesting that pH changes due to nitrification were  
513        negligible or at least too minor to completely define soil pH trends (Spearman's  
514        correlation  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  vs.  $\text{NO}_3^- + \text{NO}_2^-$  accumulation during oxic incubation  
515         $\mu\text{mol}$ ,  $\rho=0.15$ ,  $p > 0.05$ ).

516        2. Samad *et al.* (2016b, 2016a) observed a significant correlation between initial soil  
517        pH and  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  using the same incubation methodology used here.  
518        Therefore, variable acidification was not an issue, even though high gaseous N  
519        accumulation suggested substantial nitrification occurred.

520        3. pH was still not correlated to  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  in repeated soil incubations without an  
521        oxic period (and presumably minimal nitrification) (Spearman's correlation,  
522         $\rho=0.10$ ,  $p > 0.05$ ).

523 4. Omission of the oxic incubation period usually resulted in increased hypothetically  
524 emitted  $\text{N}_2\text{O}$  (%) compared with initial incubations (Table 3). The opposite would  
525 be expected if significant amounts of acidification occurred during oxic periods.

526 Based on these arguments, we tentatively conclude that factors other than pH were the most  
527 important drivers of  $\text{N}_2\text{O}$  production/reduction phenotypes and  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  in the  
528 current study but do not doubt that soil pH could exert effects on the observed phenotype,  
529 associated  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  ratios and NO accumulation patterns based on retrospective  
530 analysis of (Samad *et al.*, 2016b, 2016a). Previous evidence suggests pH based control of  
531  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  is due to a post-transcriptional impairment of enzyme maturation in the  
532 periplasm (Bergaust *et al.*, 2010; Liu *et al.*, 2014), however, we suggest this does not  
533 explain well the delayed nature of  $\text{N}_2\text{O}$  reduction observed in (Samad *et al.*, 2016a) or  
534 elsewhere (Liu *et al.*, 2010).

535 **4.2 The role of microbial community composition and distal regulators in  
536 determining observed phenotypes**

537 Samad *et al.* (2016b, 2016a) previously linked  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$ , pH, and 16S microbial  
538 community composition using the same methodology used here. However, it remained  
539 unclear whether correlations between  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  and 16S microbial community  
540 composition indicated a true causal link. Based on the well-known impacts of pH on both  
541  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  (Simek and Cooper, 2002) and microbial community structuring (Lauber *et*  
542 *al.*, 2009; Kaminsky *et al.*, 2017; Samad *et al.*, 2016b), a plausible explanation was that pH  
543 independently determined  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  and microbial community composition. A similar  
544 3-way correlation emerged here but with average daily rainfall at the sample sites in place  
545 of pH. Again, it is plausible that long term rainfall patterns or a linked variable separately  
546 influenced  $\text{N}_2\text{O}/(\text{N}_2\text{O}+\text{N}_2)$  and community composition, however, taken together Samad *et*  
547 *al.* (2016a, 2016b) and the present study may indicate an alternate story: a consistent link

548 (though less strong here than Samad *et al.* (2016a)) between N<sub>2</sub>O emission potential and  
549 community composition and a consistent continuum of N<sub>2</sub>O production/reduction  
550 phenotypes (though less obvious in (Samad *et al.*, 2016a)) both occurring across different  
551 soil sets among alternate potential confounding drivers (rainfall patterns here, pH in  
552 (Samad *et al.*, 2016b). Thus, there is some increased support for a true link between 16S  
553 community composition and N<sub>2</sub>O emission potential. Soil phenotypes were clearly  
554 sensitive to manipulations i.e. carbon addition (Figure 5 Left side vs. right side) and  
555 methodology changes altered N<sub>2</sub>O emission potential/phenotypes (Figure 1 vs. Figure 5),  
556 but different communities could hypothetically display a greater propensity for more  
557 sequential or concurrent denitrification under consistent proximal regulators. This could be  
558 due to, for example, alternate denitrification regulatory phenotypes (e.g. early *nosZ*  
559 expression) of community members between soils (Liu *et al.*, 2013; Lycus *et al.*, 2017;  
560 Bergaust *et al.*, 2011 ).

561 It remains unclear how exactly rainfall patterns shape the denitrifying community  
562 and N<sub>2</sub>O emission potential, though past hydrological experience has previously been  
563 linked to the timing of soil N<sub>2</sub>O reduction (Zhu *et al.*, 2013). Short term rainfalls prior to  
564 sampling, soil storage moisture and soil moisture content at the time of experimentation  
565 were irrelevant to N<sub>2</sub>O emission potential, therefore a longterm effect on chemistry or  
566 community selection is implied. Selection could involve recruitment of successful  
567 organisms from the available biosphere and, over longer periods, evolutionary adaption e.g.  
568 (Lynch and Neufeld, 2015; Parkin *et al.*, 1985). We hypothesize long periods of soil  
569 saturation, ensuing anoxia and slowed diffusion of N<sub>2</sub>O provide a niche in which complete  
570 and concurrent denitrifiers are more successful. Under more transient and less complete  
571 anoxia (low rainfall), denitrifiers showing short term prioritization of NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> may be  
572 selected due to a greater energy yield of NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> (Giles *et al.*, 2012; Simon and Klotz,  
573 2013), their immediacy in the denitrification pathway, a higher availability of nitrate from

574 nitrification coupled denitrification (due to semi-oxic conditions) (Wrage *et al.*, 2001), or  
575 the relatively poor (in comparison to prior reductases) activity of N<sub>2</sub>O reductase in the  
576 presence of O<sub>2</sub> (Morley *et al.*, 2008).

577 The abundance and diversity of Clade II nitrous oxide reductase genes is previously  
578 predicted to control the N<sub>2</sub>O sink capacity of soils (Jones *et al.*, 2014) as nosZII carrying  
579 denitrifiers are more likely to lack N<sub>2</sub>O producing steps (Graf *et al.*, 2014). Although  
580 nosZII gene abundances and nosZI/nosZII gene abundance ratios here were related to pH  
581 differences (Spearman correlation N<sub>2</sub>O/(N<sub>2</sub>O+N<sub>2</sub>) vs. nosZI/nosZII  $\rho=-0.47$ ,  $p < 0.05$ , vs.  
582 nosZII copy numbers  $\rho=-0.60$ ,  $p < 0.05$ ), as in previous studies (Jones *et al.*, 2014; Samad  
583 *et al.*, 2016b), they did not show any correlation to N<sub>2</sub>O/(N<sub>2</sub>O+N<sub>2</sub>) product ratios  
584 (Spearman's correlation,  $p > 0.05$ ) in this study, suggesting that they probably did not  
585 determine N<sub>2</sub>O sink capacity. Indeed, the decoupling of N<sub>2</sub>O/(N<sub>2</sub>O+N<sub>2</sub>) product ratios and  
586 nosZII abundances/ratios in this study under circumstances where pH was not found to  
587 drive N<sub>2</sub>O/(N<sub>2</sub>O+N<sub>2</sub>) product ratios may weaken prior claims (Jones *et al.*, 2014; Samad *et*  
588 *al.*, 2016b) that nosZII abundances affected N<sub>2</sub>O/(N<sub>2</sub>O+N<sub>2</sub>) product ratios rather than  
589 simply varying with the shared driver of pH.

590 Higher N<sub>2</sub>O emission potential for sequential soils occurred due to poor timing of  
591 N<sub>2</sub>O reduction rather than a deficit in actual N<sub>2</sub>O reduction capability (Figure 1), therefore  
592 a genotype based explanation (lack of nosZ containing denitrifiers vs. high presence of non-  
593 denitrifying nitrous oxide reducers only containing nosZ) for the differing N<sub>2</sub>O  
594 production/reduction phenotypes seems unlikely. Some of the lowest nosZII gene copy  
595 numbers were actually seen in concurrent soils with lower N<sub>2</sub>O emission potential (Figure  
596 2C, e.g. 39-Lake Heron, 40-Fairlie-Geraldine, 27-Lumsden). Quantification of nosZ  
597 transcripts may have been more informative in the current study as it remains unclear  
598 whether nosZ expression was just delayed in sequential N<sub>2</sub>O producing/reducing soils or

599 early function was impaired by post-transcriptional effects, which are previously observed  
600 to occlude transcription based effects (e.g. Liu *et al.*, 2010, 2014).

601 **4.3 The role of proximal regulators in determining observed phenotypes**

602 **4.3.1 The effect of carbon availability**

603 Enhanced N<sub>2</sub>O accumulation in response to carbon limitation has been attributed to  
604 competition for electrons between the different denitrification enzymes (Pan *et al.*, 2013;  
605 Ribera-Guardia *et al.*, 2014; Dendooven *et al.*, 1994). Here, carbon additions were made to  
606 denitrifying soil incubations to test the hypothesis that sequential phenotype soils have  
607 limited electron supply and thus direct electrons preferentially towards the earlier steps of  
608 denitrification. This mechanism would also explain why impaired N<sub>2</sub>O reduction activity  
609 was transient i.e. as prior electron acceptors deplete, competition would be relieved. Under  
610 these circumstances, addition of carbon should lead to increased electron availability (as  
611 long as regeneration of the electron carrier pool was not already maximal) and presumably  
612 increased early N<sub>2</sub>O reduction. Experimental evidence here mostly contradicted that  
613 hypothesis. Carbon addition to the hypothesized “electron limited” sequential soils did not  
614 result in a consistent shift towards a concurrent phenotype, though increases in  
615 denitrification process rates do suggest that those soils were indeed somewhat carbon  
616 limited (Table 3). The overall trend observed was that carbon availability, substrate  
617 type/quality, C/N ratios or some other related effect sustained or drove soils toward a  
618 sequential phenotype with increased N<sub>2</sub>O hypothetically emitted % and NO accumulation,  
619 excepting soil 20-Waitaha (Figure 5). A possible explanation is that carbon addition  
620 preferentially stimulated NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> reduction leading to accumulation of NO which may  
621 in turn have an inhibitory impact on N<sub>2</sub>O reductase (see 4.3.2). Though it remains unclear  
622 why the initial denitrification steps would be preferentially enhanced.

623 Comparisons to initial soil incubations may also be informative about the role of  
624 carbon, though differences in methodology and initial  $\text{NO}_3^-$  concentrations should be taken  
625 into consideration. These initial soil incubations probably had less carbon available for  
626 denitrification due carbon consumption during oxic pre-incubations as evidenced by lower  
627  $\text{CO}_2$  production rates during denitrification (Table 3). If the crude assumption is made that  
628 average  $\text{CO}_2$  production during denitrification was proportional to carbon availability then  
629 hypothetical  $\text{N}_2\text{O}$  emission potential in many of these soils (40-Fairlie-Geraldine, 20-  
630 Waitaha Valley, 1-Woodend) appear to exhibit a positive correlation to carbon availability  
631 (Table 3).

632 Based on the above observations it seems plausible that differences in carbon  
633 accounted for some of the phenotypic variation observed between soils in the original  
634 incubations. Direct measurement of starting carbon concentrations (e.g. total C, dissolved  
635 organic C) or substrates by mass spectrometry in soils would be beneficial in future  
636 investigations of the observed denitrification phenotypes.

637 **4.3.2 Nitric oxide accumulation, nitrite accumulation and nitrate concentration**

638 Accumulation of prior N oxyanions/oxides ( $\text{NO}_3^-$ ,  $\text{NO}_2^-$ , NO) can impair  $\text{N}_2\text{O}$   
639 reduction activity (Blackmer and Bremner, 1978; Firestone *et al.*, 1979; Gaskell *et al.*,  
640 1981; Senbayram *et al.*, 2012; Ha *et al.*, 2015; Pan *et al.*, 2013; Zhou *et al.*, 2008; Frunzke  
641 and Zumft, 1986) due to N reductase competition for electrons e.g. (Pan *et al.*, 2013;  
642 Dendooven *et al.*, 1994) or alternate mechanisms such as direct inhibitory interaction  
643 between NO and  $\text{N}_2\text{O}$  reductase (Frunzke and Zumft, 1986),  $\text{NO}_2^-$  protonation to inhibitory  
644 nitrous acid (Zhou *et al.*, 2008) or  $\text{NO}_2^-$  based enhancement of obligate  $\text{N}_2\text{O}$  endproduct  
645 producing fungi (Maeda *et al.*, 2015). Again, transient accumulation of N oxyanions/oxides  
646 is in line with transient impairment of  $\text{N}_2\text{O}$  reductase in sequential  $\text{N}_2\text{O}$  producing/reducing  
647 soils. The NO accumulation patterns and timing are particularly conspicuous: NO

648 accumulation was higher in sequential N<sub>2</sub>O producing/reducing soils (Spearman's  
649 correlation, average  $\mu\text{mol NO vial}^{-1}$  vs. N<sub>2</sub>O hypothetically emitted %,  $\rho=0.80$ ,  $p<0.0001$ ),  
650 increased N<sub>2</sub>O reduction coincided with rapid depletion of NO (Figure 1C), concurrent N<sub>2</sub>O  
651 producing/reducing soils eventually stabilized NO levels to a low steady state (Figure 1A),  
652 and C amendments increasing or decreasing N<sub>2</sub>O hypothetically emitted % also  
653 increased/decreased max NO accumulation (Table 3). Based on these observations we  
654 hypothesize that sequential type soils were unable to maintain NO concentrations below an  
655 inhibitory level, resulting in impaired N<sub>2</sub>O reduction until NO production ceased.  
656 Alternatively, NO accumulation may be indicative of a significant NO<sub>2</sub><sup>-</sup> pool stimulating  
657 NO production by both abiotic and biotic processes (Lim *et al.*, 2018). NO<sub>2</sub><sup>-</sup> reductase is  
658 proposed to be particularly competitive with N<sub>2</sub>O reductase for electrons due to a shared  
659 use of the electron carrier cytochrome C550 (Richardson *et al.*, 2009; Pan *et al.*, 2013),  
660 therefore, tracking of endogenous NO<sub>2</sub><sup>-</sup> and evaluating responses to exogenous NO<sub>2</sub><sup>-</sup> in  
661 future experimentation is highly desirable.

662 Correlations between some measures of N<sub>2</sub>O emission potential and predicted NO<sub>3</sub><sup>-</sup> +  
663 NO<sub>2</sub><sup>-</sup> at the start of anoxia (Table 2) suggest initial N supply may have impacted the  
664 observed gas kinetic patterns. However, we are skeptical based on a lack of correlation with  
665 the most pertinent variables (N<sub>2</sub>O hypothetically emitted % and N<sub>2</sub>O/(N<sub>2</sub>O+N<sub>2</sub>) (max N<sub>2</sub>O))  
666 and potential biases in the other measures. For instance, measures taken at the 50hr anoxia  
667 timepoint will capture higher ratios in soils with high NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> because they typically  
668 take longer to denitrify. Further, soils showed diverse relative N<sub>2</sub>O reduction rates at the  
669 same level of remaining NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> (Figure 4B, C).

670 NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> concentration effects could hypothetically be occluded in correlations  
671 between separate soils if individual soils had dramatically differing sensitivities to the  
672 similar NO<sub>3</sub><sup>-</sup> or NO<sub>2</sub><sup>-</sup> concentrations. Differing sensitivity of N<sub>2</sub>O reduction to NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup>

673 concentration in different soils has been previously reported, with higher sensitivity in  
674 lower pH soils (Blackmer and Bremner, 1978; Firestone *et al.*, 1979; Gaskell *et al.*, 1981).  
675 However, we did not observe pH based ranking of soils once  $\text{NO}_3^- + \text{NO}_2^-$  availability was  
676 accounted for (Figure S4), and this analysis cannot be considered conclusive due to the  
677 potential bias of denitrification progress; because of differences in initial  $\text{NO}_3^- + \text{NO}_2^-$ ,  
678 different soils reached the same remaining  $\text{NO}_3^- + \text{NO}_2^-$  at different times and different  
679 amounts of  $\text{NO}_3^- + \text{NO}_2^-$  were already utilized. Experiments applying varying  
680 concentrations of  $\text{NO}_3^-$  or  $\text{NO}_2^-$  to pH manipulated soils or using temporally constant  $\text{NO}_3^-$   
681 concentrations (chemostats) would be necessary to understand the true impact of  $\text{NO}_3^-$  and  
682  $\text{NO}_2^-$  on the observed denitrification phenotypes and  $\text{N}_2\text{O}$  emission potential.

683 **4.4 Conclusion**

684 Here, we demonstrate considerable variation in  $\text{N}_2\text{O}$  emission potential for New  
685 Zealand pasture soils based on the timing and activity of  $\text{N}_2\text{O}$  reduction and associated with  
686 the accumulation of NO gas. We show an association between  $\text{N}_2\text{O}$  production/reduction  
687 phenotypes and microbial communities in the absence of a pH effect and in conjunction  
688 with results from Samad *et al.* (2016b, 2016a) suggest this improves the plausibility of a  
689 true link between community composition and the observed phenotypes/ $\text{N}_2\text{O}$  emission  
690 potential. Additional correlates of  $\text{N}_2\text{O}$  emission potential/emission phenotypes are  
691 identified at both distal (long term rainfall) and proximal levels (carbon availability) which  
692 may be linked by a common mechanism of NO accumulation and inhibition. Further  
693 research on the phenomena described here should focus on directly testing the impact of  
694 NO concentrations on the observed phenotypes, the potential accumulation of  $\text{NO}_2^-$  in  
695 sequential type soils, and the potential for regulatory effects such as delayed transcription  
696 of *nosZ*.

**Acknowledgements**

697 This work was funded by the New Zealand Government through the New Zealand Fund for  
698 Global Partnerships in Livestock Emissions Research to support the objectives of the  
699 Livestock Research Group of the Global Research Alliance on Agricultural Greenhouse  
700 Gases (Agreement number: 16084 and SOW12-GPLER-OU-SM) awarded to SEM and the  
701 University of Otago, New Zealand. MH was funded by a University of Otago Postgraduate  
702 Scholarship. PD P received funding from the FACCE-ERA-GAS project MAGGE-pH  
703 under the grant agreement no. 696356. We would like to thank the Nitrogen group at the  
704 Norwegian University of Life Sciences NMBU for access to lab, robotic autosamplers,  
705 experimental and technical assistance. We also thank Steve Wakelin and AgResearch for  
706 providing historic physicochemical data and preliminary DNA samples for the analysed  
707 soils.

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907 **Figure Legends**

908 **Figure 1.** N<sub>2</sub> gas kinetics of 20 NZ pasture soils amended with 2mM NH<sub>4</sub>NO<sub>3</sub> and incubated  
909 in a helium atmosphere. The figure shows the wide variation in timing of N<sub>2</sub>O reduction  
910 (N<sub>2</sub> production) leading to variable N<sub>2</sub>O accumulation. N<sub>2</sub>O hypothetically emitted (%)  
911 (Table S2) was used to evaluate sequentiality of N<sub>2</sub>O production/reduction on a continuous  
912 scale (soils ordered top left to bottom right) and define soil phenotypes (concurrent (A),  
913 intermediate (B) and sequential (C)) based on discrete arbitrary cutoffs. Circles, squares,  
914 triangles represent three replicate vials. Concurrent N<sub>2</sub>O production/reduction, is associated  
915 with specific NO emission pattern: Lower NO accumulation eventually stabilising at a  
916 pseudo steady state. Sequential soils usually accumulate higher max NO. Note that the  
917 panels have different scaling of N<sub>2</sub>O (Orange), N<sub>2</sub> (Black) and NO (Blue) and values are  
918 reported as  $\mu$ mol-N per vial. Scaled version available (Figure S1). Version with CO<sub>2</sub>  
919 available (Figure S2).

920 **Figure 2.** Microbial community analyses reveal links between 16S community composition  
921 and N<sub>2</sub>O emission potential/phenotypes (A), average daily rainfall over 10 years and pH  
922 (B). qPCR reveals greater abundance of nosZII relative to nosZI (C). NMDS ordination  
923 plots (A, B) compare prokaryotic dissimilarities (Bray Curtis) of a single pooled mixed  
924 sieved soil sample per site. Full distance specific replications are presented in Figure S3.  
925 Stress values for ordinations were 0.14. Correlations between variables and NMDS axes or  
926 the Bray Curtis dissimilarity matrix are presented in Table 1.

927 **Figure 3.** Relationship between average daily rainfall (average daily mm rainfall over 10  
928 years prior to sampling) and N<sub>2</sub>O production/reduction phenotypes (A) or N<sub>2</sub>O  
929 hypothetically emitted % (B). p-value presented is for difference of medians using

930 Wilcoxon rank sum test with chi-squared approximation. Correlations between all rainfall  
931 and N<sub>2</sub>O emission potential metrics are presented in **Table S3**.

932 **Figure 4.** Normalized soil N<sub>2</sub> production rates (N<sub>2</sub> rate over max N<sub>2</sub> rate in same soil)  
933 increase as available NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> is depleted in anoxic soil incubations amended with 2mM  
934 NH<sub>4</sub>NO<sub>3</sub>. In most soils, predicted NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> at beginning of anoxic period is greater than  
935 measured NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> immediately following NO<sub>3</sub><sup>-</sup> amendment (A) indicating NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup>  
936 accumulation, most likely due to nitrification during oxic pre-incubations. Plots (C, D)  
937 allow comparison of N<sub>2</sub>O reduction activity at similar NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> concentration for each  
938 soil. Ranking of soils across Y-axis could indicate potential variation in soil N<sub>2</sub>O reduction  
939 sensitivity to NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> concentration, however this ranking simply describes the  
940 aforementioned N<sub>2</sub>O production/reduction phenotypes (C) and a potential effect cannot be  
941 separated from the natural progression of denitrification or other biases.

942 **Figure 5.** Effect of carbon (10mM Na-glutamate + 4mM NH<sub>4</sub>NO<sub>3</sub> by flooding and  
943 draining) on soil denitrification kinetics in representative soils ranging in N<sub>2</sub>O  
944 hypothetically emitted (%)/phenotypes: concurrent (A), sequential (B) and intermediate  
945 (C). Triplicate incubations per treatment (dots, squares, triangles) were carried out under  
946 anoxia without oxic preincubation. Carbon amended treatments right, C negative controls  
947 left. Single leaky reps excluded for 20 +N and 40 +C+N. N<sub>2</sub>O (Orange), N<sub>2</sub> (Black) and NO  
948 (Blue) are reported as  $\mu$ mol-N per vial. Carbon additions shift kinetics in the tested  
949 concurrent soils towards sequential N<sub>2</sub>O production/reduction and greater NO accumulation  
950 while no dramatic change is observed for the sequential or intermediate soils. Graphs with  
951 axes scaled to same maximum available (Figure S5)

952 **Tables**

953 **Table 1.** N<sub>2</sub>O emission potential and other variable correlations to community dissimilarity

| Variable vs. community composition  | NMDS axis 1<br>co-ords<br>(Spearman's) |         | NMDS axis 2<br>co-ords<br>(Spearman's) |         | Bray curtis<br>dissimilarity<br>matrix (Mantel) |         |
|---|--|---------|--|---------|---|---------|
|   | rho (ρ)                                | Sig (p) | rho (ρ)                                | Sig (p) | r<br>statistic                                  | Sig (p) |
| N <sub>2</sub> O hypothetically emitted (%)                               | -0.51                                  | 0.022*  | -0.34                                  | 0.137   | 0.18  | 0.139   |
| IN <sub>2</sub> O (50hrs)   | -0.64                                  | 0.002*  | -0.29                                  | 0.215   | 0.15  | 0.208   |
| IN <sub>2</sub> O (N <sub>2</sub> plateau)                                | -0.40                                  | 0.093   | -0.34                                  | 0.154   | 0.24  | 0.073   |
| N <sub>2</sub> O/(N <sub>2</sub> O+N <sub>2</sub> ) (50hrs)               | -0.70                                  | 0.001*  | -0.28                                  | 0.232   | 0.26  | 0.079   |
| N <sub>2</sub> O/(N <sub>2</sub> O+N <sub>2</sub> )(max N <sub>2</sub> O) | -0.61                                  | 0.004*  | -0.26                                  | 0.262   | 0.19  | 0.134   |
| pH H <sub>2</sub> O   | -0.32                                  | 0.165   | 0.50                                   | 0.025*  | 0.35  | 0.005*  |
| pH CaCl <sub>2</sub>  | -0.37                                  | 0.110   | 0.35                                   | 0.128   | 0.32  | 0.008*  |
| Daily average rainfall (10 years)   | 0.67                                   | 0.001*  | 0.55                                   | 0.012*  | 0.44  | 0.009*  |

954

955 \*p<0.05

956 **Table 2.** Correlations between predicted NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> at the start of anoxia and measures of  
957 N<sub>2</sub>O emission potential

| Variable  | Spearman's ρ | Sig (p) |
|---|--------------|---------|
| IN <sub>2</sub> O (50hrs)   | 0.70         | 0.001*  |
| IN <sub>2</sub> O (N <sub>2</sub> plateau)                                | 0.53         | 0.019*  |
| N <sub>2</sub> O/(N <sub>2</sub> O+N <sub>2</sub> ) (50hrs)               | 0.54         | 0.013*  |
| N <sub>2</sub> O/(N <sub>2</sub> O+N <sub>2</sub> )(max N <sub>2</sub> O) | 0.30         | 0.200   |
| N <sub>2</sub> O hypothetically emitted (%)                               | 0.09         | 0.701   |

958

959 \*p<0.05

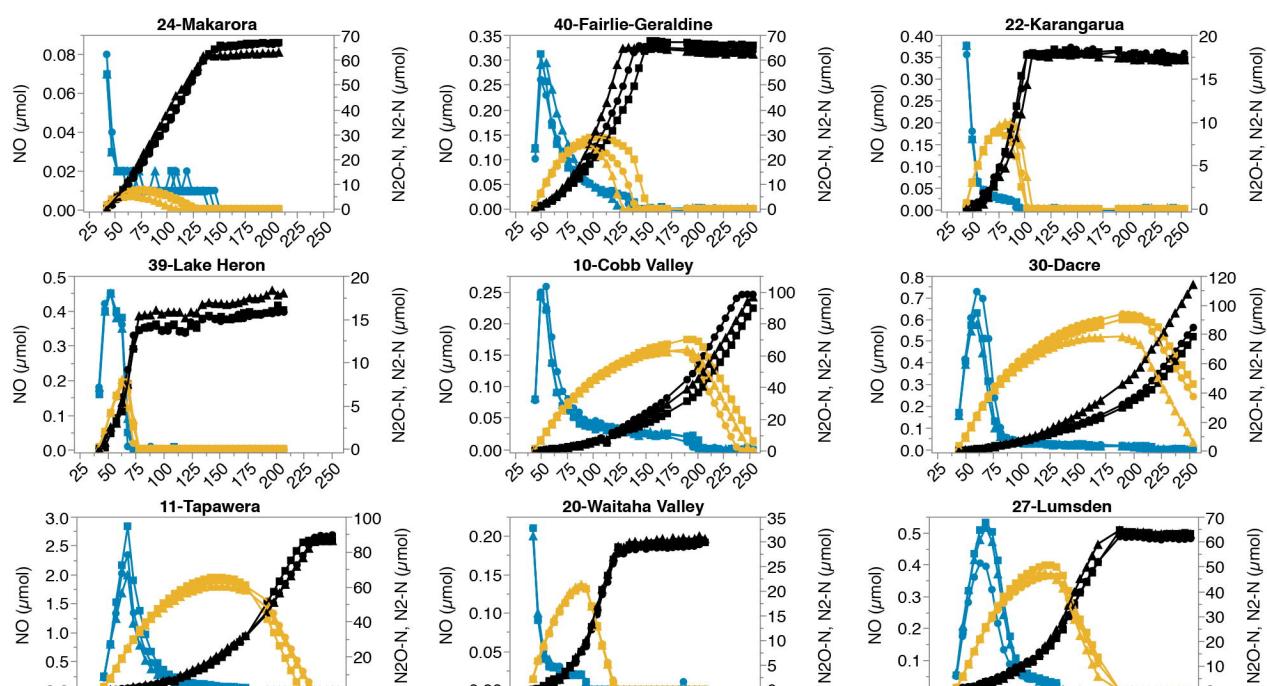
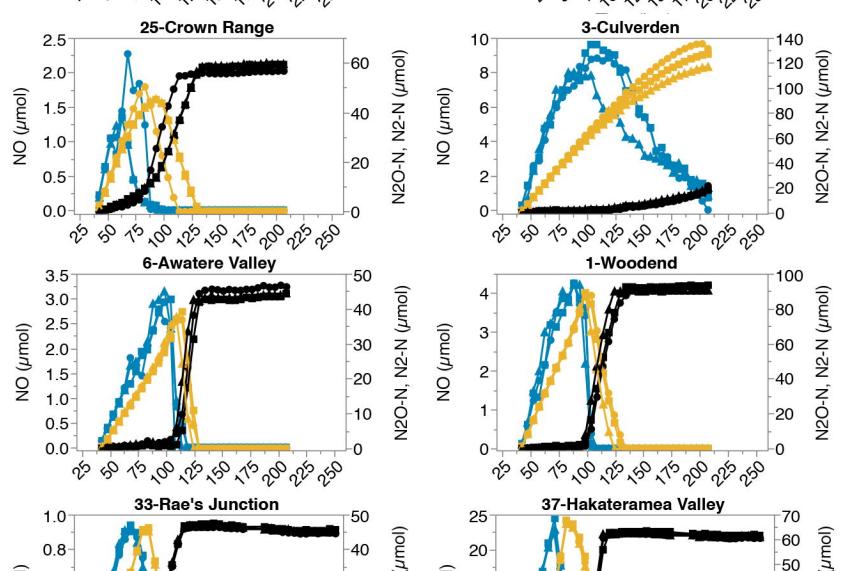
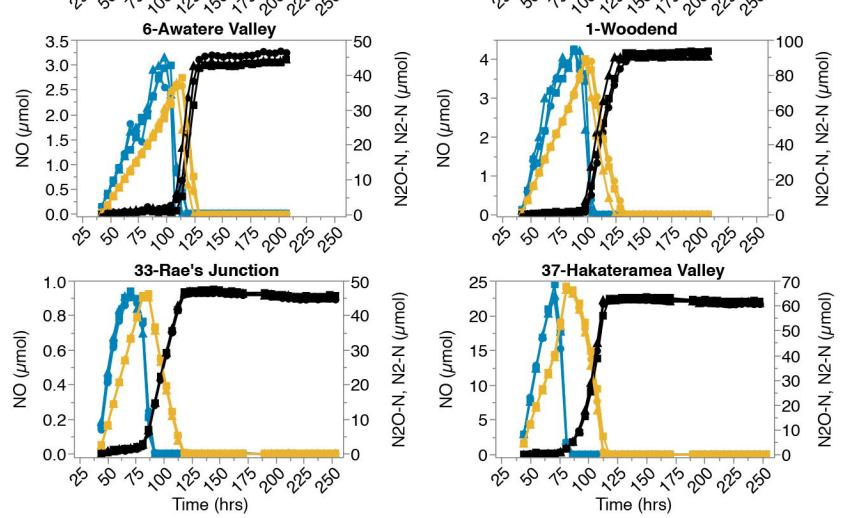
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961 **Table 3.** Comparison of gas kinetics features across all incubation treatments for five soils

| Soil   | Concurrent       |      |      |              |      |      | Sequential       |      |      |             |      |      | Intermediate |      |      |
|--|------------------|------|------|--------------|------|------|------------------|------|------|-------------|------|------|--------------|------|------|
|  | Fairlie-Ger (40) |      |      | Waitaha (20) |      |      | Rae's Junct (33) |      |      | Woodend (1) |      |      | Waipapa (5)  |      |      |
|  | 1st              | +N   | +C+N | 1st          | +N   | +C+N | 1st              | +N   | +C+N | 1st         | +N   | +C+N | 1st          | +N   | +C+N |
| Av N <sub>2</sub> rate before peak N <sub>2</sub> O (μmol/hr)  | 0.38             | 0.36 | 0.10 | 0.17         | 0.14 | 0.06 | 0.16             | 0.08 | 0.10 | 0.08        | 0.18 | 0.22 | 0.11         | 0.26 | 0.54 |
| Av N <sub>2</sub> rate after peak N <sub>2</sub> O (μmol/hr)   | 0.98             | 2.20 | 3.08 | 0.58         | 0.52 | 1.87 | 1.26             | 1.57 | 4.65 | 2.62        | 2.77 | 4.22 | 0.84         | 1.30 | 2.40 |
| Max N <sub>2</sub> rate (μmol/hr)  | 1.39             | 2.54 | 4.26 | 0.76         | 0.64 | 3.84 | 1.51             | 2.10 | 6.89 | 3.65        | 3.53 | 5.01 | 1.04         | 1.50 | 3.14 |
| Max N <sub>2</sub> O rate (μmol/hr)  | 0.91             | 1.60 | 2.46 | 0.71         | 0.84 | 1.15 | 1.25             | 1.59 | 2.77 | 2.24        | 2.81 | 2.83 | 0.79         | 1.11 | 1.87 |
| Estimated Av total N turnover rate (μmol NO <sub>2</sub> <sup>-</sup> , NO, N <sub>2</sub> O, N <sub>2</sub> N/hr) | 2.78             | 4.40 | 5.46 | 1.42         | 1.50 | 3.29 | 2.73             | 3.27 | 6.09 | 4.44        | 5.59 | 6.90 | 1.77         | 3.10 | 4.99 |
| Av CO <sub>2</sub> rate (μmol/hr)  | 1.15             | 1.92 | 2.68 | 0.90         | 0.92 | 1.64 | 1.34             | 1.51 | 2.55 | 1.99        | 2.91 | 3.61 | 0.74         | 1.27 | 2.28 |
| Av NO (μmol)   | 0.10             | 0.67 | 1.15 | 0.04         | 0.04 | 0.56 | 0.61             | 0.63 | 0.62 | 1.54        | 0.27 | 0.23 | 1.34         | 1.50 | 0.51 |
| Max NO (μmol)  | 0.29             | 2.21 | 4.64 | 0.21         | 0.27 | 1.27 | 0.92             | 2.31 | 2.51 | 4.25        | 1.53 | 1.23 | 3.25         | 6.06 | 2.79 |
| N <sub>2</sub> OI (N <sub>2</sub> plateau)   | 0.44             | 0.61 | 0.69 | 0.57         | 0.58 | 0.80 | 0.65             | 0.65 | 0.69 | 0.66        | 0.63 | 0.70 | 0.64         | 0.53 | 0.55 |
| N <sub>2</sub> O/(N <sub>2</sub> O+N <sub>2</sub> )(max N <sub>2</sub> O)  | 0.59             | 0.79 | 0.95 | 0.71         | 0.80 | 0.97 | 0.87             | 0.94 | 0.95 | 0.96        | 0.93 | 0.92 | 0.84         | 0.79 | 0.71 |
| Hypothetically emitted N <sub>2</sub> O (%)  | 0.41             | 0.80 | 0.92 | 0.74         | 0.83 | 1.00 | 0.98             | 0.99 | 1.00 | 0.96        | 0.90 | 0.96 | 0.87         | 0.70 | 0.74 |

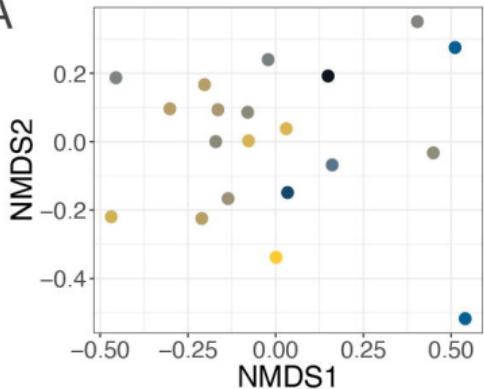
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963

**A****B****C**

— NO — N2O — N2

A

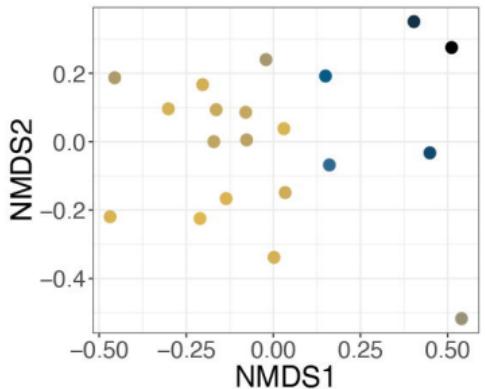


N2O hypothetically emitted (%)



25 50 75 100

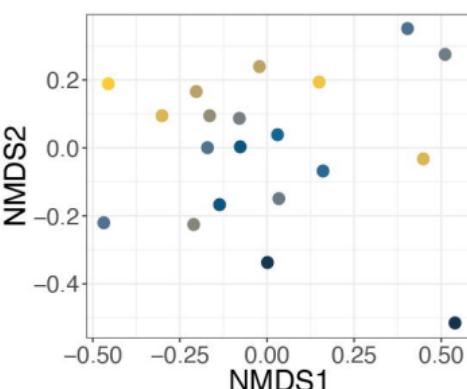
B



Average Daily Rainfall (mm)



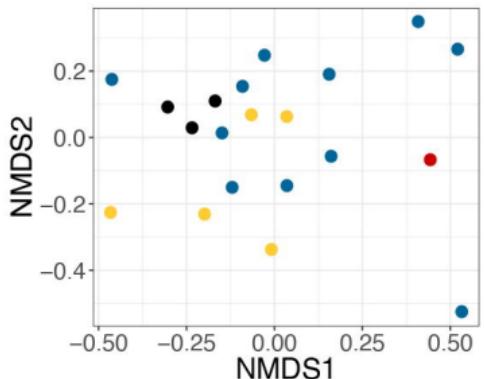
3 6 9



pH H<sub>2</sub>O

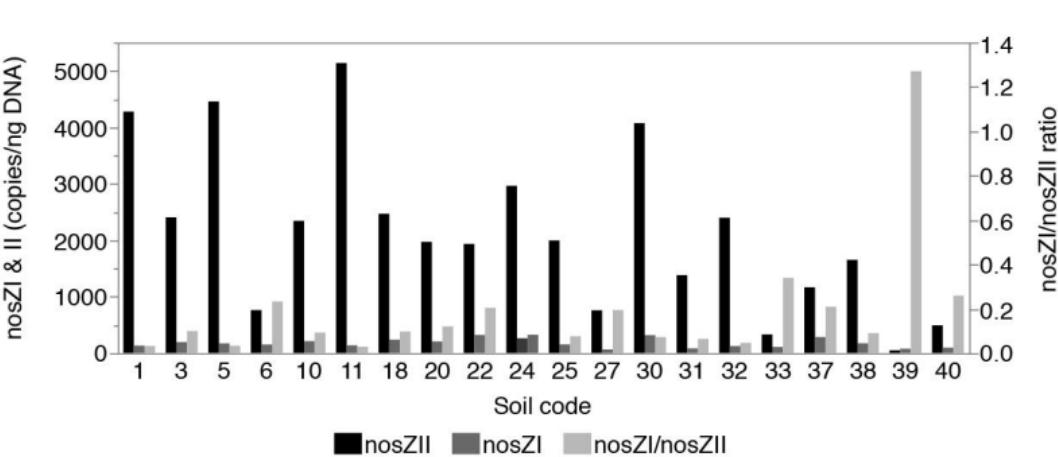


5.6 6.0 6.4 6.8

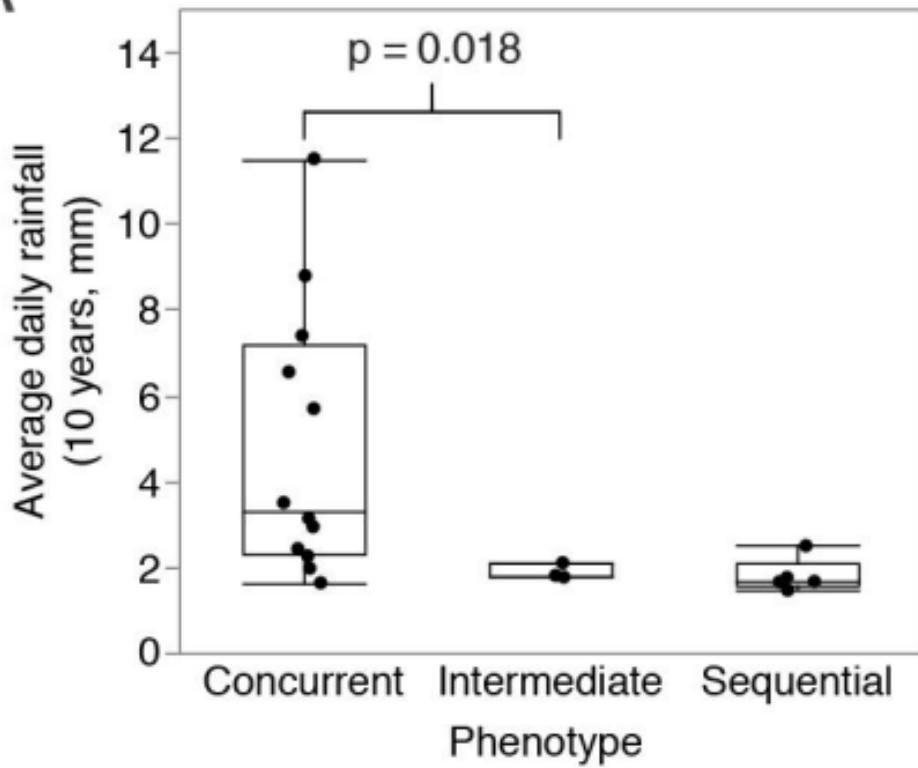


Sequential  
N-A  
Intermediate  
Concurrent

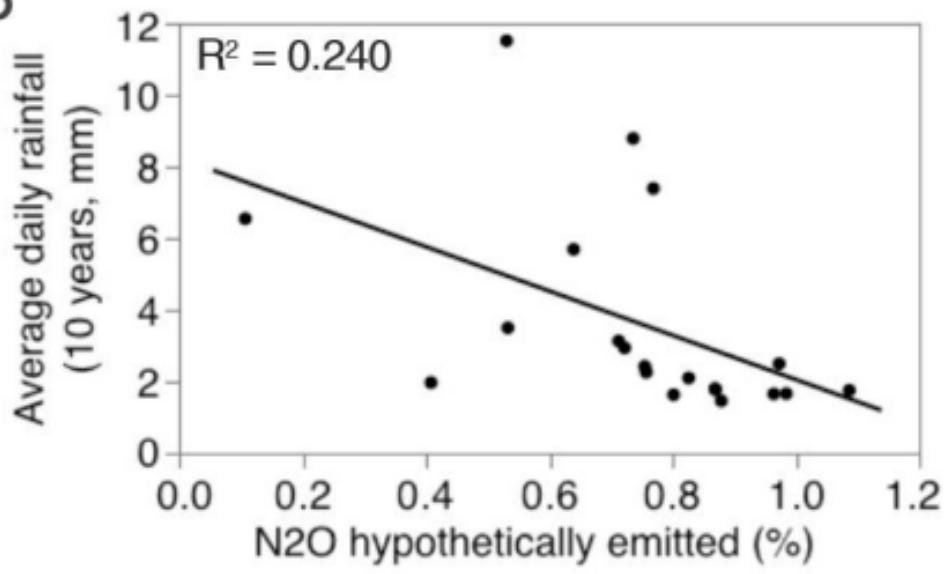
C

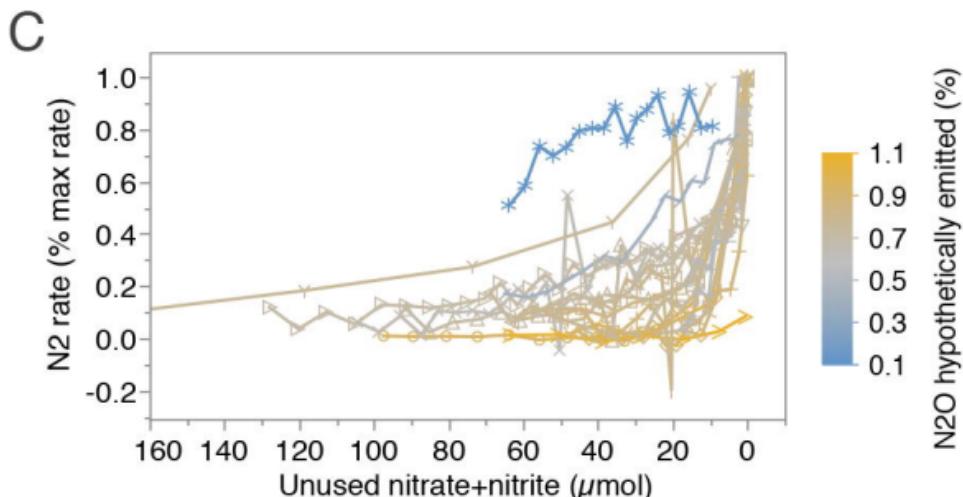
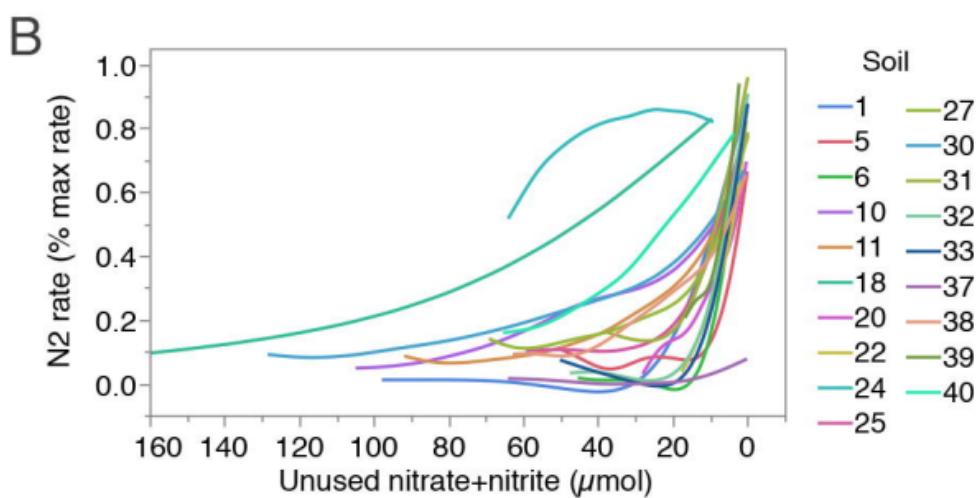
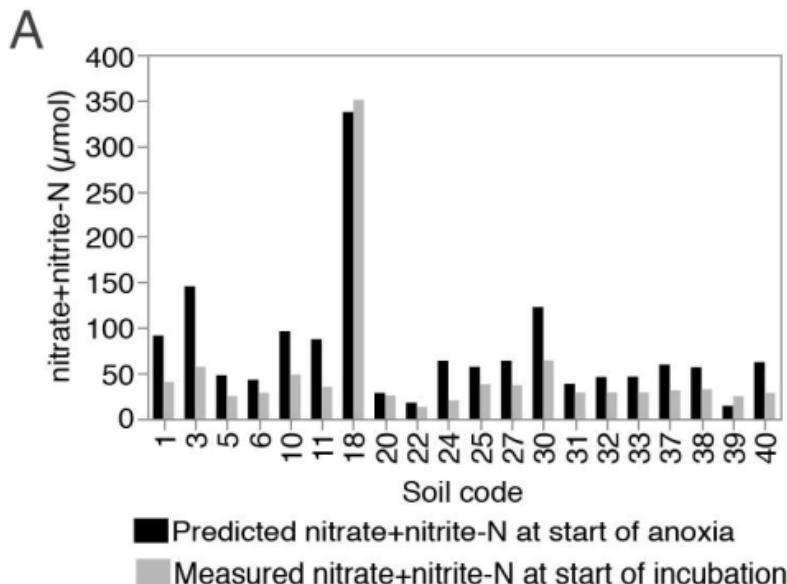


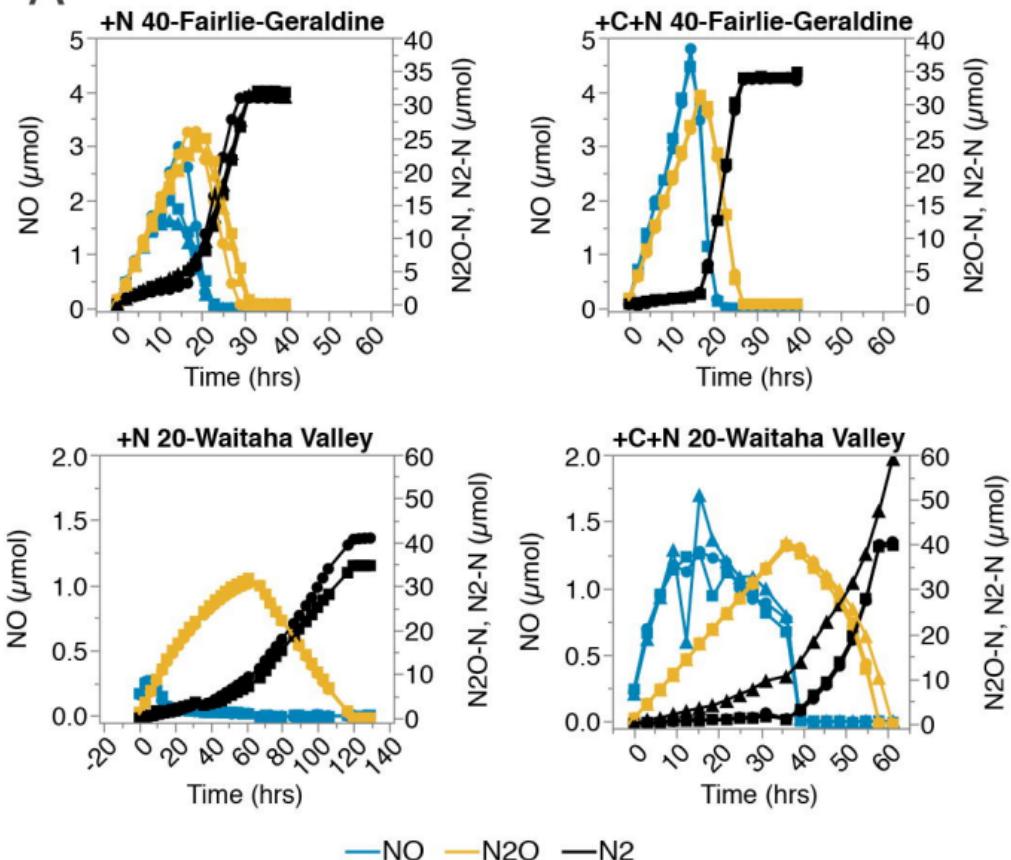
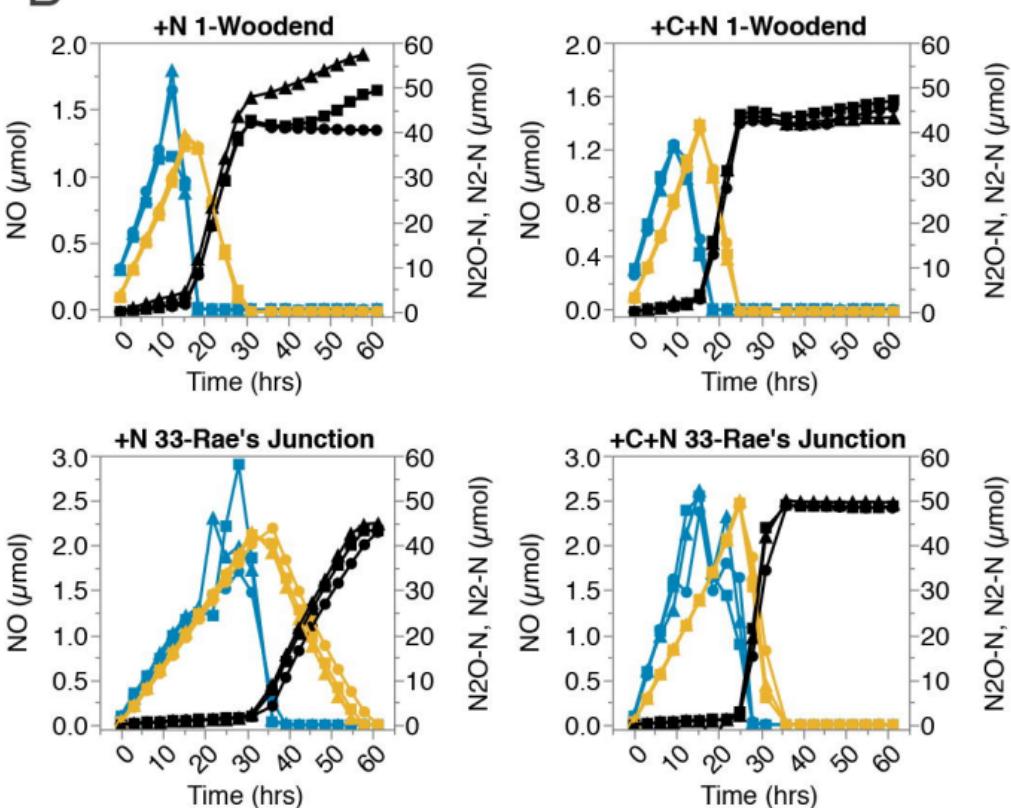
A



B





**A****B****C**