Metal-driven anaerobic oxidation of methane as an important methane sink in methanic cold seep sediments

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23 **Abstract**

24 Anaerobic oxidation of methane (AOM) coupled with reduction of metal oxides is

25 supposed to be a globally important bioprocess in marine sediments. However, the

26 responsible microorganisms and their contributions to methane budget are not clear

27 in deep sea cold seep sediments. Here, we combined geochemistry, muti-omics and

28 numerical modeling to study metal-dependent AOM in methanic cold seep

29 sediments in the northern continental slope of the South China Sea. Geochemical

30 data based on methane concentrations, carbon stable isotope, solid-phase sediment

31 analysis and pore water measurements indicate the occurrence of anaerobic methane

32 oxidation coupled to metal oxides reduction in the methanic zone. The 16S rRNA

33 gene amplicons and transcripts, along with metagenomic and metatranscriptomic

34 data suggest that diverse ANME groups actively mediated methane oxidation in the

35 methanic zone either independently or in syntrophy with e.g. ETH-SRB1 as

36 potential metal reducers. Modeling results suggest that the estimated rates of

37 methane consumption via Fe-AOM and Mn-AOM were both 0.3 μmol cm^{-2} yr^{-1},

38 which account for ~3% of total CH$_4$ removal in sediments. Overall, our results

39 highlight metal-driven anaerobic oxidation of methane as an important methane sink

40 in methanic cold seep sediments.

41 **Introduction**

42 Marine cold seeps are not only an indicator of gas hydrate reservoirs, but also an

43 important methane source to the oceans, which has a significant impact on the global

44 carbon cycle and climate change[1]. It is estimated that 0.02 Gt methane is consumed

45 annually in the sediment with an additional 0.02 Gt methane releasing annually into

46 the overlying ocean by seafloor cold seeps[2]. Thus, the production and consumption

47 of methane is a key component of the carbon cycle for cold seeps[3]. Anaerobic

48 oxidation of methane (AOM) driven by microbial communities, plays a key role in

49 decreasing methane emissions to the atmosphere[4]. Anaerobic methanotrophic

50 archaea (ANME) mediate this process through the coupling of methane oxidation to

51 the reduction of nitrite, nitrate, manganese/iron oxides, and sulfate[5-8]. Sulfate-driven

52 AOM (S-AOM) by assemblages of ANME and sulfate-reducing bacteria (SRB)[6, 9] is

53 regarded as the major process for methane sink within cold seep sediments, reaching

54 highest activities within the sulfate-methane transition zone (SMTZ)[10].

55 Early studies show that methane oxidation possibly coupled with metal oxidation

56 can still occur at a considerable rate in the methanic zone below the SMTZ when

57 sulfate has been completely depleted or at very low levels[11]. A range of efforts have

58 been undertaken to demonstrate the occurrence of metal oxides-driven AOM (Metal-

59 AOM, including high-valence iron/manganese oxides)[5, 12, 13]. This microbial process

60 is mediated by ANME through the reverse methanogenesis pathway, typically in

61 syntrophy with dissimilatory iron/manganese reducing bacteria[5, 14]. Investigations of

62 enrichment cultures have also revealed that ANME-2a, ANME-2c, and ANME-2d

63 can perform AOM coupled to the extracellular dissimilatory reduction of iron and

64 manganese oxides independently using e.g. a unique set of multiheme cytochromes

65 (MHCs)[12, 13, 15-17].

66   The activity rates of Fe-AOM are efficiently estimated by incubation experiments or

67   geochemical modeling, but rarely for Mn-AOM (**Table 1**)[18-25]. Microbial culture

68   experiments from the Eel River Basin seep have found that manganese oxides can

69   drive AOM as electron acceptors more efficiently than ferrihydrite[5]. However,

70   microorganisms involved in the coupling between AOM and metal reduction in

71   marine environments are still largely unknown, especially for manganese reduction[26].

72   The contribution of Mn-AOM for CH_4 removal in situ marine environments is still

73   fully identified, neither. Without in-depth understanding of the role of Metal-AOM

74   in the biogeochemical cycle, the contribution of metal reduction to the global carbon

75   cycle, especially the methane sink, is likely to be undervalued[27].

76   Haima cold seep was firstly discovered as an active cold seep in the Qiongdongnan

77   basin on the northwest slope of the South China Sea by the dives of a remotely

78   operated vehicle named as Haima in 2015 (**Figure S1a**)[28]. A large number of

79   findings have since been emerging about the biogeochemistry of cold-seep

80   carbonates, benthos and sediments in Haima cold seep[29-32]. Massive amounts of

81   terrigenous metal oxides are supplied into the continental slope of the South China

82   Sea from rivers[33]. Consequently, iron/manganese-containing minerals are part of the

83   major components in the sediments of this region with high-flux methane seeps,

84   rendering it be such a natural laboratory to investigate the role of metal oxides in

85   methane cycle.

86   In this study, combining geochemical and microbial analyses of the Haima cold seep

87   sediments, we aimed to (1) reveal the occurrence of Metal-AOM in the methanic

88   zone; (2) identify microorganisms involved in the Metal-AOM and their key

89   mechanisms; (3) estimate the contribution of removal of methane by Fe/Mn-AOM.

90   Our findings provide insights into the coupling mechanism between iron/manganese

91   reduction and AOM as well as the role of Metal-AOM in the biogeochemical cycle.

92 **Results and Discussions**

93 **Geochemical data indicate anaerobic methane oxidation in the methanic zone**

94 At a water depth of 1375 m, a 4.3-meter-long piston core was retrieved from the

95 Haima cold seep in the South China Sea, where a gas chimney and bubble plumes

96 were observed indicative of ongoing seepage activities (**Figure S1b-d**). Massive gas

97 hydrates were also found at 347-352 cm below the seafloor (cmbsf) and 420 cmbsf

98 (**Figure S1e-g**). Methane (CH_4) was the dominant seeping hydrocarbon gas (0.13-

99 919.57 μM), along with ethane (C_2H_6) being detected (0.07-3.96 μM) below 130-

100 140 cmbsf (**Table S1**). Based on methane and sulfate profiles (**Figure 1a and Table**

101 **S1**), our sediment core samples were categorized into three biogeochemical zones[34,

102 35]: sulfate reduction zone (surface-130 cmbsf), sulfate–methane transition zone

103 (~130-250 cmbsf), and methanic zone (250-430 cmbsf).

104 In the methanic zone, methane concentrations fluctuated between 53 μM and 920

105 μM. A notable decrease was observed from 781 μM at 210-220 cmbsf to 53 μM at

106 350 cmbsf, with the enrichment of the stable carbon isotope ratios ($\delta^{13}C$) in residual

107 CH_4 from −68.77‰ to −64.33‰ (**Table S1**), pointing to occurrence of biological

108 unitization of methane[25]. In accordance with this, the dissolved inorganic carbon

109 (DIC) values increased from 250-260 cmbsf (10.38 mM) to 330-340 cmbsf (17.27

110 mM; **Figure 1b**), as also evidenced by the increase in total alkalinity[36] from 16.01

111 mM to 28.94 mM (**Table S1**). The measured $\delta^{13}C_{DIC}$ values (<−42.30‰; **Figure 1b**)

112 were much more ^{13}C-depleted than that of typical marine organic matters (~−20‰)

113 in this sea area[37]. This confirms that DIC increase was caused by microbial methane

114 oxidation rather than microbial degradation of other organic matters. Additionally,

115 low concentration profiles of phosphate (PO_4^{3-}) and ammonium (NH_4^+), lower than

116 41.65 μM and 56.72 μM respectively (**Table S1**) also support that organic matter

117   degradation was not the main reason for increased DIC concentrations in these

118   sediment samples[38].

Diverse ANME actively mediated methane oxidation in the methanic zone

120   Anaerobic methanotrophs are assigned to three distinct clades (ANME-1 with

121   subgroups a and b, ANME-2 with subgroups a, b, c and d, and ANME-3) within the

122   phylum Halobacteriota[39]. To identify potential ANME clades in the methanic zone,

123   we performed DNA and RNA sequencing of sediment samples. Taxonomy

124   classifications of archaeal 16S rRNA gene amplicons indicate that ANME accounted

125   for 69~87% of the whole archaeal community in the methanic zone (**Figure 2a and**

126   **Table S2**). These ANME populations are phylogenetically diverse, including

127   ANME-1a (up to 80% at 320-330 cmbsf), ANME-1b (up to 24% at 420-430 cmbsf),

128   ANME-2c (up to 16% at 280-290 cmbsf) and ANME-3 (up to 68% at 360-370

129   cmbsf). Relative abundances of 16S rRNA transcripts suggest that ANME-1a (up to

130   54% at 320-330 cmbsf), ANME-3 (up to 84% at 360-370 cmbsf), ANME-2c (up to

131   63% at 280-290 cmbsf), and ANME-1b (up to 19% at 400-410 cmbsf) were the

132   dominant and active players for AOM occurring in the methanic zone (**Figure 2b**

133   **and Table S2**).

134   Metagenomic assembly and binning yielded 17 metagenome-assembled genomes

135   (MAGs) taxonomically affiliated with methane-metabolizing lineages that either

136   produce or consume methane (**Table S3**). Among them, nine harbor sequences

137   encoding the catalytic subunit of methyl-coenzyme M reductases (McrA)[40] involved

138   in methyl reduction during methane oxidation (**Figure 2c and Table S4**). They

139   belong to clades of ANME-1 (n=3) and ANME-2 (n=4). Additionally, S11_12_1 and

140   S11_6_17 belonging to *Methanosarcinaceae* were predicted to have the capability to

141   perform the methanogenesis pathway. MAGs for ANME-3 lineages were not

142   recovered despite its highly relative abundance based on 16S rRNA genes and

143 transcripts (**Table S2**). Based on read mapping (**Table S3**), species represented by

144 ANME-1 (i.e., Ca. *Methanophagales*[41]) S11_3_46 (0.4-7.7 % of the whole microbial

145 community) and S11_6_2 (0.4-7.3%) were observed to be the most abundant in the

146 methanic zone, followed by S11_10_20 (0.1-1.2%) and S11_8_35 (0.1-0.6%) from

147 ANME-2c (Ca. *Methanogaster*[39, 40]). Metatranscriptomic analyses (**Table S5**)

148 showed that three ANME-1 genomes (Co_S11_862, S11_3_46 and S11_6_2) highly

149 expressed *mcrA* genes in methanic zone (up to 4382 TPM at 400-410 cmbsf). The

150 transcripts of *mcrB* genes from ANME-1 (Co_S11_862) and ANME 2c (S11_12_8)

151 genomes also had high expression levels (up to 2378 TPM at 400-410 cmbsf). These

152 results further suggest that ANME populations were actively responsible for the

153 observed anaerobic methane oxidation in the methanic zone.

154 **Methane oxidation is coupled to metal oxides reduction in the methanic zone**

155 In the methanogenic zone, dissolved ferrous iron (Fe^{2+}) and manganese (Mn^{2+})

156 concentrations in pore water were found to reach up to 148 μM at 370-380 cmbsf

157 and as high as 2289 μM at 330-340 cmbsf, respectively (**Figure 1c and Table S1**).

158 The Spearman correlations (**Figure 3a**) results further show that Fe^{2+} and Mn^{2+}

159 concentrations have a strong positive covariance with CH_4 (ρ = 0.874 and 0.699,

160 respectively), DIC (ρ = 0.891 and 0.818, respectively); Fe^{2+} has a strong negative

161 relationship with $\delta^{13}C_{DIC}$ (ρ = −0.655; $p < 0.05$). These data indicate that the high

162 amounts of dissolved Fe^{2+} and Mn^{2+} are associated with the fluctuation of methane

163 concentrations in methanic zone[15, 42]. Correspondingly, the solid-phase sediment

164 analysis revealed richer $Fe_2O_3^T$ (3.17–3.74%) and MnO_2^T (0.06%) in the methanic

165 zone than those of the SMTZ ($Fe_2O_3^T$: 2.31-3.60%; MnO_2^T: 0.04%) (**Figure 1d and**

166 **Table S6**). The reactive iron minerals, including carbonate-associated iron (Fe_{carb}; up

167 to 1.16%), amorphous iron (oxyhydr)oxides (Fe_{ox1}; up to 1.19%) and magnetite iron

168 (Fe_{mag}; up to 0.61%), were detected with the higher contents in the methanic zone

169 (**Figure 1e-1g and Table S7**). Therefore, these data implied sufficient supplies of

170   reactive Fe-oxides and the occurrence of Fe authigenic minerals as the products of

171   iron reduction[43]. Similar to that of iron, total manganese (MnO_2^T) is also elevated

172   from 0.04% in the SMTZ to 0.06% in the methanogenic zone (**Figure 1d and Table**

173   **S6**). Given the elevated MnO_2^T and extremely high dissolved Mn^{2+} (up to 2289 μM),

174   the contribution of manganese reduction to AOM cannot be ignored in this seep.

175   Overall, porewater and solid-phase profiles support metal-driven methane oxidation

176   in methanic sediments.

177   **Potential microorganisms involved in dissimilatory metal reduction**

178   For in-depth understanding of Fe(III)/Mn(IV)-dependent AOM in Haima cold seep,

179   it is critical to identify the indigenous microorganisms responsible for this process.

180   Members in different ANME clades are suggested to mediate metal-driven AOM by

181   extracellular electron transfer (EET) to Mn(IV)/Fe(III) (oxyhydr)oxides or metal-

182   reducing partners. In iron-reducer *Geobacter sulfurreducens*, the process of EET is

183   carried out via MHCs during metal reduction[44] [26]. For ANME-2d from freshwater

184   Fe-AOM enrichment, a set of MHCs for extracellular dissimilatory Fe(III) reduction

185   were highly expressed[12, 45, 46]. Here, all analyzed ANME genomes were found to

186   contain the genes encoding several c-type and periplasmic cytochromes (**Table S8**).

187   Among all ANME genomes, three MAGs, S11_2_24 and S11_6_25, belonging to

188   ANME-2c, also encode S-layer–associated multi-heme c-type cytochromes,

189   implying a role of ANME-2c archaea with an S-layer protein in conducting electron

190   derived from reverse methanogenesis shuttling from the archaeal membrane to the

191   outside of the cell[46]. S11_12_8 and S11_6_25 affiliated with the family ANME-2c,

192   also encode outer membrane cytochrome Z (*omcZ*) gene (**Table S9**) which plays an

193   important role in Fe(III) reduction[47]. Furthermore, S11_12_8 not only actively

194   expressed at zone C with the maximum of MAG's abundance and TPM values with

195   *mcrB* gene, but also had the significantly positive relation with the CH_4 (ρ=0.790,

196    $p<0.01$), Fe^{2+} (ρ=0.734, $p<0.01$), Mn^{2+} (ρ=0.601, $p<0.05$; **Figure 3b**).

8

197   Gene encoding decaheme c-type cytochrome (*mtrC*), was present in S11_6_22 and

198   Co_S11_566 affiliated with ETH-SRB1 from the order *Desulfobacterales* (**Table**

199   **S5**). The two MAGs (S11_6_22 and Co_S11_566) have the higher abundance in the

200   methanic zone (Mean: 0.20% and 0.12%) than the SMTZ (Mean: 0.18% and 0.07%;

201   **Table S3**). Besides, spearman's correlation results (**Figure 3b**) show that

202   Co_S11_566 closely related with concentrations of Fe^{2+} (ρ=0. 699) and Mn^{2+} (ρ=0.

203   650). Thus, ETH-SRB1 probably act as the role of metal reducing bacteria in the

204   methanic zone. We also found the presence of hypothetical proteins attributed to

205   porins, cytochrome *c* binding motif sites (CxxCH) and Geobacter-related gene

206   markers (*omc*) for iron reduction in Co_S11_933 (**Table S9**), belonging to

207   Zixibacteria, which was reported with pathways of either oxidation or reduction of

208   ferric/ferrous iron and arsenate/arenite and nitrate/nitrite[48]. Co_S11_933 also

209   displayed a higher abundance in the methanic zone (0.03~0.06%) than in other

210   zones (**Table S3**).

211   **Contribution of Metal-AOM to methane consumption**

212   Geochemical observations and microbiological analyses support that Fe and Mn

213   oxides reduction is coupled to methane oxidation in the methanic zone. We then

214   used numerical modeling to predict their contributions to methane consumption.

215   Constrained by the measured porewater data and Fe leaching experiments (**Figure 1;**

216   **Tables S1, S6 and S7**), the results of the reaction-transport modeling predict the

217   model-derived rates for Fe-AOM of up ~0.02 μmol CH_4 cm^{-3} d^{-1} in the methanic

218   zone (**Figure 4 and Table 1**). These rates are more than 20 times as big as the

219   estimated potential Fe-AOM rates from in situ marine methanic sediments with

220   much higher Fe^{2+} concentration (180~800 μM)[20-23] (**Table 1**), but much lower than

221   those derived from stimulated microbial communities in laboratory incubation

222   studies[5, 24, 25, 49].

223   As Mn speciation data were not available, we used the diffusive Mn^{2+} flux

224   calculated based on the quasi-linear concentration gradient at the depth interval of

225   ~250 cm and 350 cm to represent the rate of Mn-AOM. Based on the porewater

226   profiles, our estimated diffusive flux for Mn^{2+} is 1.276 µmol cm^{-2} yr^{-1}. Thus, taking

227   into account that only one molecule of CH_4 is needed to reduce four molecule of

228   MnO_2, CH_4 removal by Mn-AOM is estimated to 0.319 µmol CH_4 cm^{-2} yr^{-1} [19]. This

229   is a minimum estimate as the potential Mn^{2+} consumption by authigenic minerals is

230   not taken into account. Depth-integrated rates of Fe-AOM and Mn-AOM are both

231   0.3 µmol cm^{-2} yr^{-1} in the methanic zone, which are considerably lower than S-AOM

232   rate (~20 µmol cm^{-2} yr^{-1}) and account for ~1.5% of total CH_4 removal by microbial

233   metabolism, respectively (**Table 1**). The high S-AOM rate is caused by methane

234   bubble dissolution while its upward-ascending and enhanced sulfate supply from

235   seawater due to bubble irrigation. The estimated depth-integrated rate of Fe-AOM in

236   the Haima seep falls within the range of those reported in different environments

237   globally[20-23]. These data from Haima cold seep provide the first in situ evidence for

238   quantitatively significant manganese-dependent AOM in marine sediments. Given

239   an apparent elevated sedimentary manganese content in the methanic zone (from

240   0.04% to 0.06%) and high concentration of dissolved Mn^{2+} (up to 2289 µM), the

241   contribution for Mn-AOM consumed by authigenic minerals could have been

242   underestimated.

243

Conclusions

244

245   Methane oxidation occurs in methanic zones driven by sedimentary microbial

246   communities is an important mechanism that controls natural emissions of methane

247   from the gas hydrate–bearing area. It happens mainly due to the presence of

248   alternative electron acceptors other than sulfate to react with methane. Abundant

249   Fe/Mn-(oxyhydr)oxides preserved in the SCS shelf sediments might be migrated

250   into the study region due to the rapid increase of anomalous subsidence towards the

251   deep water areas in the Qiongdongnan basins (**Figure 5**). Therefore, high amounts

252   of buried reactive Fe(III)/Mn(IV) minerals seem to be an important available

253   electron acceptors for AOM in the methanic zone of the Haima methane seep,

254   accompanying by the generation of highly alkaline, extremely $\delta^{13}C_{DIC}$-depleted and

255   Fe(II)/Mn(II)-enriched pore waters, abundant Fe–Mn carbonates, along with

256   authigenic magnetite by microbial iron/manganese reduction. In methanic sediments,

257   abundant active ANME groups (ANME-1 and ANME-2c), and potential

258   dissimilatory iron reducers (e.g. ETH-SRB1) are potentially involved in Metal-

259   AOM *in situ*. Mechanistically, the apparent ability of ANME-2c to oxidize methane

260   via the release of single electrons in this study should also be able to respire solid

261   electron acceptors directly via extracellular metal reduction, which would confirm

262   the presence of previously reported methane oxidation coupled to insoluble

263   Fe(III)/Mn(IV) reduction. It is estimated that Metal-AOM at least contributes 3%

264   CH_4 removal from methanic sediments to the seep. Overall, Metal-AOM could

265   significantly impact the biogeochemical cycles in consuming CH_4 in modern marine

266   sediments.

267

Materials and methods

269   **Sampling and geochemical analyses**

270   Sediment core HM-S11 with the length of 430 cm was obtained by a gravity piston

271   sampler at the ROV1 site of Haima cold seeps[28] (**Figure S1**) during the

272   Haiyangdizhi10 cruise in June 2019 by the Guangzhou Marine Geological Survey,

273   China Geological Survey.

274 The pore water was extracted from each sediment segment with the interval of 40

275 cm except the top 60 cm (20-cm interval) on board at room temperature by a

276 vacuum apparatus. The concentrations of Fe^{2+} and Mn^{2+} in pore water were

277 immediately on board determined by UV/Vis spectrophotometer (Beijing Purkinje,

278 China) using the 1,10-phenanthroline colorimetric method and potassium periodate

279 oxidation spectrophotometry, respectively. Sulfate concentrations were measured by

280 an ICS-1100 ion chromatography (Thermofisher, USA) with an analytical error of

281 $\pm 1\%$. The $\delta^{13}C_{DIC}$ values in pore water were analyzed by a multiflow-isotope ratio

282 mass spectrometer (Thermofisher, Delta V Advantage, USA), with an analytical

283 error of $\pm 0.2‰$. The concentrations of PO_4^{3-} and NH_4^+ were photometrically

284 measured on board using a UV-Vis spectrophotometer (Hitachi U5100, Hitachi

285 Limited, Tokyo, Japan) with an analytical error of $\pm 3.0\%$. Porosity was determined

286 from the weight loss before and after freeze-drying of the wet sediments[37] using a

287 cutting ring with the volume of 5 ml on board, assuming a density of the porewater

288 of 1.0 g cm^{-3}.

289 The $C_1 \sim C_3$ concentrations of the headspace gas samples were determined using an

290 Agilent 6850 gas chromatograph (Thermo, USA) with a Porapark Q column. The

291 detection limit for all gases is 10 ppm, and the quantification limit is 30 ppm. The

292 $\delta^{13}C$ values of the methane were measured using gas chromatography-isotope ratio-

293 mass spectrometry (GC-IR-MS; Thermo, USA), and are reported relative to the

294 Vienna Peedee Belemnite standard (V-PDB), with an analytical error of $\pm 0.5‰$.

295 The major element composition of sediments was determined by an iCAP 7200

296 Inductively Coupled Plasma (ICP) Optical Emission Spectrometry (OES) (Thermo,

297 USA). The content of different iron phase mineral components was determined by

298 Infinite M200Pro multifunction enzyme marker (TECAN, USA) with a sequential

299 extraction method[50], and measured at the light absorption wavelength of 510 nm.

300   The accuracy and repeatability of the absorption wavelength of the instrument are <

301   ±0.5 nm (λ > 315 nm).

DNA and RNA extraction

303   Total genomic DNA and RNA of each sample was extracted using Soil DNA Kit and

304   Soil RNA Mini Kit (Omega Bio-Tek Inc., Norcross, GA) according to the

305   manufacturer's instructions, respectively. DNA concentration and purity were

306   measured by TBS-380 (Turner Biosystems, CA, USA) and Nanodrop ND-2000

307   (Thermo Fisher Scientific, Waltham, USA), respectively. DNA extract quality, RNA

308   degradation and contamination were monitored on 1% agarose gels. RNA quantity

309   was measured using Qubit 2.0 (Thermo Fisher Scientific, MA, USA) and Nanodrop

310   One (Thermo Fisher Scientific, MA, USA) at the same time. RNA integrity was

311   accurately detected using the Agilent 2100 system (Agilent Technologies, Waldbron,

312   Germany).

Amplicon analysis of 16S rRNA genes and transcripts

314   The DNA and RNA for each sample were amplified in triplicate using primers

315   338F/806R for bacteria and Arch344F/Arch915R for archaea. Their PCR products

316   were pooled and purified. The same amount of the PCR product from each sample

317   was mixed to construct a sequencing library. High-throughput sequencing was

318   carried out on the Illumina MiSeq sequencing platform using PE300 chemical at

319   Majorbio Bio-Pharm Technology Co. Ltd. (Shanghai, China).

320   After demultiplexing, the resulting sequences were merged with FLASH (v1.2.11)[51]

321   and quality filtered with fastp (v0.19.6)[52]. Then the high-quality sequences were de-

322   noised using DADA2[53] plugin in the Qiime2 (v2020.2)[54] pipeline with

323   recommended parameters, which obtains single nucleotide resolution based on error

324   profiles within samples. DADA2 denoised sequences are usually called amplicon

325   sequence variants (ASVs). To minimize the effects of sequencing depth on alpha and

326   beta diversity measure, the number of sequences from each sample was rarefied to

327   4000, which still yielded an average Good's coverage of 97.90%. Taxonomic

328   assignment of ASVs was performed using the Blast consensus taxonomy classifier

329   implemented in Qiime2 and the SILVA 16S rRNA database (v138).

330   **Metagenomic sequencing**

331   DNA extract was fragmented to an average size of about 400 bp using Covaris

332   M220 (Gene Company Limited, China) for paired-end library construction. Paired-

333   end library was constructed using NEXTFLEX® Rapid DNA-Seq (Bioo Scientific,

334   Austin, TX, USA). Adapters containing the full complement of sequencing primer

335   hybridization sites were ligated to the blunt-end of fragments. Paired-end

336   sequencing was performed on Illumina NovaSeq (Illumina Inc., San Diego, CA,

337   USA) at Majorbio Bio-Pharm Technology Co., Ltd. (Shanghai, China) using

338   NovaSeq Reagent Kits according to the manufacturer's instructions

339   (www.illumina.com).

340   **Assembly and binning of metagenomes**

341   Raw reads derived from the 13 metagenome libraries were quality-controlled by

342   clipping off primers and adapters then filtering out artefacts and low-quality reads

343   using Read_QC module within the metaWRAP pipeline v1.3.2 [55]. Filtered reads

344   were individually assembled using SPAdes v3.13.0 (metaSPAdes mode, default

345   parameters, for samples S11_2-3, S11_8-10, S11_12-13)[56] or MEGAHIT v1.1.3

346   (default parameters, for samples S11_1, S11_4-7, S11_11)[57]. Additionally, all

347   samples were co-assembled using MEGAHIT v1.2.9 (--kmin_1pass --k-min 31).

348   Each assembly was binned using the binning module within the metaWRAP pipeline

349   v1.3.2 (--metabat2 --maxbin2 --concoct for individual assembly; --metabat2 for co-

350   assembly). For each assembly except S11_10, the three bin sets (one for co-

351   assembly) were then consolidated into a final bin set with the bin_refinement

352   module of metaWRAP pipeline v1.3.2 (-c 50 -x 10). For S11_10, bin sets were

353   consolidated into a final bin set with DAS Tool v1.1.2 (default parameters)[58]. Finally,

354   638 bins were obtained from the 14 assemblies. They were then combined and

355   dereplicated using dRep v 2.6.2 (-comp 50 -con 10 -sa 0.95 --S_algorithm fastANI)

356   at 95% average nucleotide identity clustering (species level)[59]. After dereplication, a

357   total of 351 dereplicated MAGs were obtained. Each bin was taxonomically

358   assigned according to the Genome Taxonomy Database (GTDB) version r207 using

359   GTDB-tk v2.0.0 [60].

360   MAGs were annotated with FeGenie v2.0 and METABOLIC v4.0 [61, 62]. CoverM

361   v0.5.0 "genome" (https://github.com/wwood/CoverM) was used to obtain relative

362   abundance of each genome (parameters: –min-read-percent-identity 0.95 –min-read-

363   aligned-percent 0.75 –trim-min 0.10 –trim-max 0.90). For *mcr*A phylogenetic

364   analyses, amino acid sequences were aligned with MUSCLE algorithm[63] (-maxiters

365   16) and the maximum-likelihood phylogenetic tree was constructed in MEGA X[64].

366   **Metatranscriptomic sequencing**

367   Whole mRNAseq libraries were generated by Guangdong Magigene Biotechnology

368   Co.,Ltd. (Guangzhou, China) using NEB Next® Ultra™ Nondirectional RNA

369   Library Prep Kit for Illumina® (New England Biolabs, MA, USA) following

370   manufacturer's recommendations. Briefly, the bacterial and archaeal 16S rRNA

371   transcripts in total RNA samples were reduced by Ribo-zero rRNA Removal Kit.

372   Fragmentation was carried out using NEB Next First Strand Synthesis Reaction

373   Buffer. The first strand cDNA was synthesized using random hexamer primer and

374   M-MuLV Reverse Transcriptase (RNase H), and the second strand cDNA synthesis

375   was subsequently performed using DNA Polymerase I and RNase H. Remaining

376 overhangs were converted into blunt ends via exonuclease/polymerase activities.

377 After adenylation of 3' ends of DNA fragments, NEB Next Adaptor with hairpin

378 loop structure were ligated to prepare for hybridization. In order to select cDNA

379 fragments of preferentially 150~200 bp in length, the fragments were selected with

380 AMPure XP beads (Beckman Coulter, Beverly, USA). Then PCR was performed

381 with Phusion High-Fidelity DNA polymerase, Universal PCR primers and Index (X)

382 Primer. At last, PCR products were purified with AMPure XP beads and library

383 insert size was assessed on the Agilent 2100 system (Agilent Technologies,

384 Waldbron, Germany). The clustering of the index-coded samples was performed on

385 a cBot Cluster Generation System. After cluster generation, the library was

386 sequenced on an Illumina Hiseq Xten platform and 150 bp paired-end reads were

387 generated.

388 **Metatranscriptomic analysis**

389 Raw metatranscriptomic reads were quality filtered using Read_QC module within

390 the metaWRAP pipeline v1.3.2 [55]. Paired-end reads were merged using PEAR v0.9.8

391 [65]. The reads corresponding to ribosomal RNAs were removed using SortMeRNA

392 v4.2.0 [66]. Subsequently, mRNA reads were mapped to the predicted protein-coding

393 genes from MAGs using Salmon v1.5.0 in mapping-based mode (parameters: -

394 validateMappings -meta)[67]. The expression level for each gene was normalized to

395 transcript per million (TPM).

396 **Numerical modeling**

397 A one-dimensional, reaction-transport model[22, 68] was applied to simulate two solid

398 phases (reactive Fe oxides and Fe carbonate) and six dissolved species (SO_4^{2-}, CH_4,

399 DIC, Ca^{2+}, Mg^{2+}, and Fe^{2+}). The reactions and their kinetic rate expressions

400 considered in the model are listed in **Table S1**. Net reaction terms for all species and

401 model parameters are listed in **Tables S11 and S12**, respectively. Detailed model

402 construction can be found in Supplementary Materials.

403 **Flux calculations**

404 Diffusive fluxes J (μmol cm^{-2} yr^{-1}) of dissolved Mn^{2+} in the methanic zone were

405 calculated using its pore water concentration gradients according to Fick's first law

406 of diffusion. The algorithms are detailly described in Supplementary Materials.

407 **Statistical analysis**

408 The Spearman's linear correlation among the geochemical parameters and microbial

409 abundances of the subsamples was analyzed using the R package Vegan[69] via

410 RStudio (Ver. 1.3.1093).

411 **Data availability**

412 All metagenomes and metatranscriptomes are available at the Sequence Read

413 Archive under BioProject accession number PRJNA738468. Amplicon data were

414 deposited in Figshare (https://doi.org/10.6084/m9.figshare.21696296.v1).

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602

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Author contributions

QL and XD designed the study. XX, TZ, and XW performed porewater and sediment sampling, geochemical analyses. ML performed geochemical modeling estimates. XD and XX processed metagenome, metatranscriptome data analyses. JT and ZC provided the picture of seafloor observations. XW and JZ interpreted geophysical data. XX and TZ conducted amplicon sequencing and microbial diversity analyses. XX, XD, and ML wrote the manuscript. QL, CZ, and XY modified the manuscript. All authors reviewed the results and approved the submitted manuscript.

Competing interests

The authors declare no conflict of interest.

623   **Figure legends**

624   **Figure 1 Geochemical profiles of the sediment core HM-S11 in the Haima seep.**

625   (a) Profiles of methane (CH_4) and sulfate (SO_4^{2-}) contents in porewater; (b)

626   Concentrations of dissolved inorganic carbon (DIC) and stable carbon isotope ratios

627    $\delta^{13}C_{DIC}$ in porewater; (c) Concentrations of dissolved Fe^{2+} and Mn^{2+} in porewater; (d)

628   Contents of $Fe_2O_3^T$ and MnO_2^T in sediments; (e~h) Sequential extraction of iron

629   minerals in sediments. Fe_{carb} = carbonate-associated Fe; Fe_{ox1} = amorphous iron

630   (oxyhydr)oxides; Fe_{mag} = magnetite Fe; Fe_{py} = pyrite Fe. Zones A, B and C are

631   suggested as the sulfate zone, the sulfate-methane transition zone, and the methanic

632   zone.

633   **Figure 2 Relative abundances and phylogenetic tree of ANMEs.**

634   Relative abundances of ANME based on (a) 16S rRNA genes and (b) 16S rRNA

635   transcripts in the methanic zone. (c) Maximum-likelihood phylogenetic tree of *mcrA*

636   sequences.

637   **Figure 3 Spearman correlations of sediments in the core HM-S11.** (a) Spearman

638   correlation coefficients between depth-wise distribution of geochemical parameters;

639   (b) Correlation between geochemical parameters and abundances of MAGs belong

640   to ANMEs and metal reduction bacteria. The stars symbolize p values of correlation.

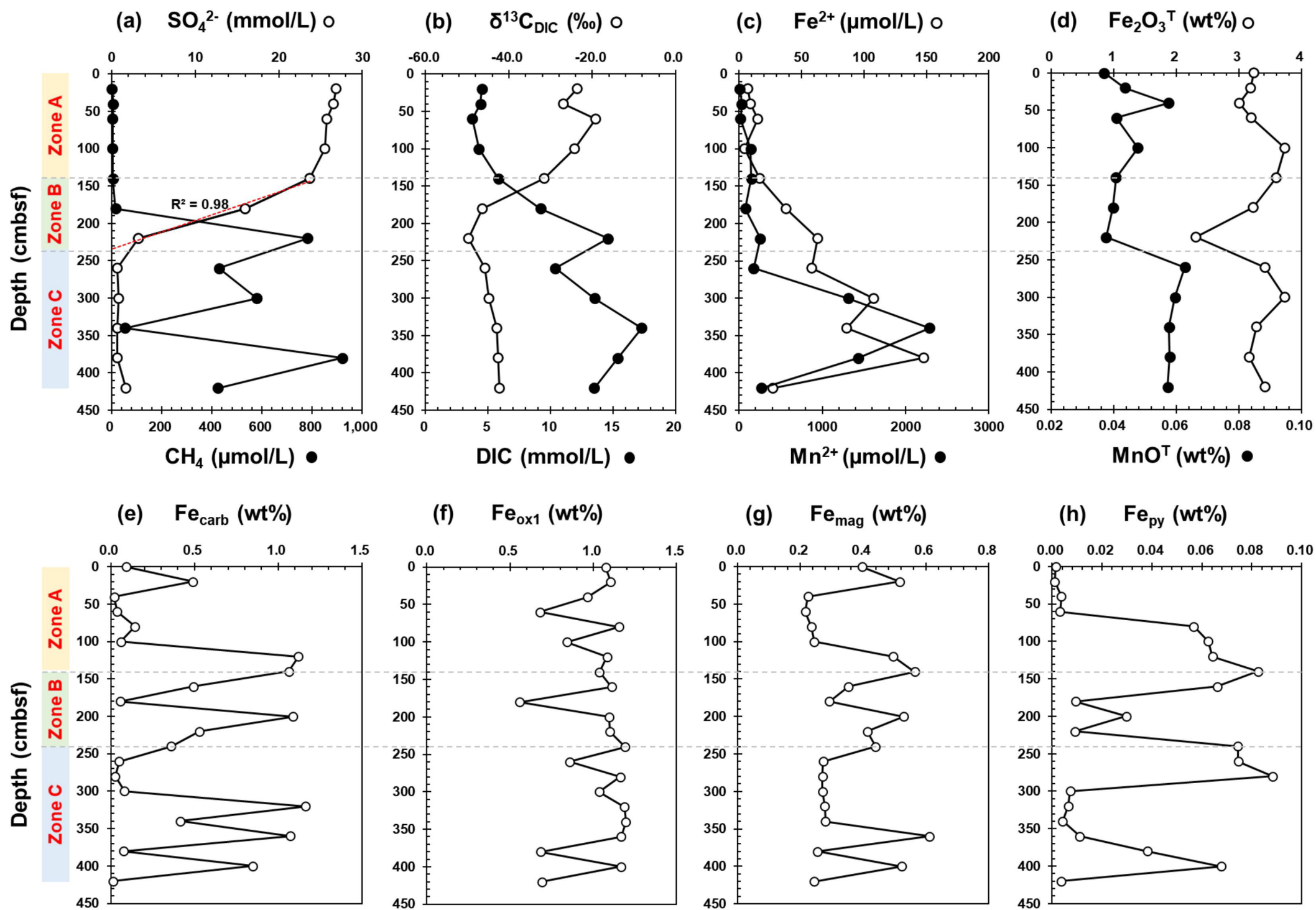641   *** means $p < 0.001$; ** means $p < 0.01$; *means $p < 0.05$.

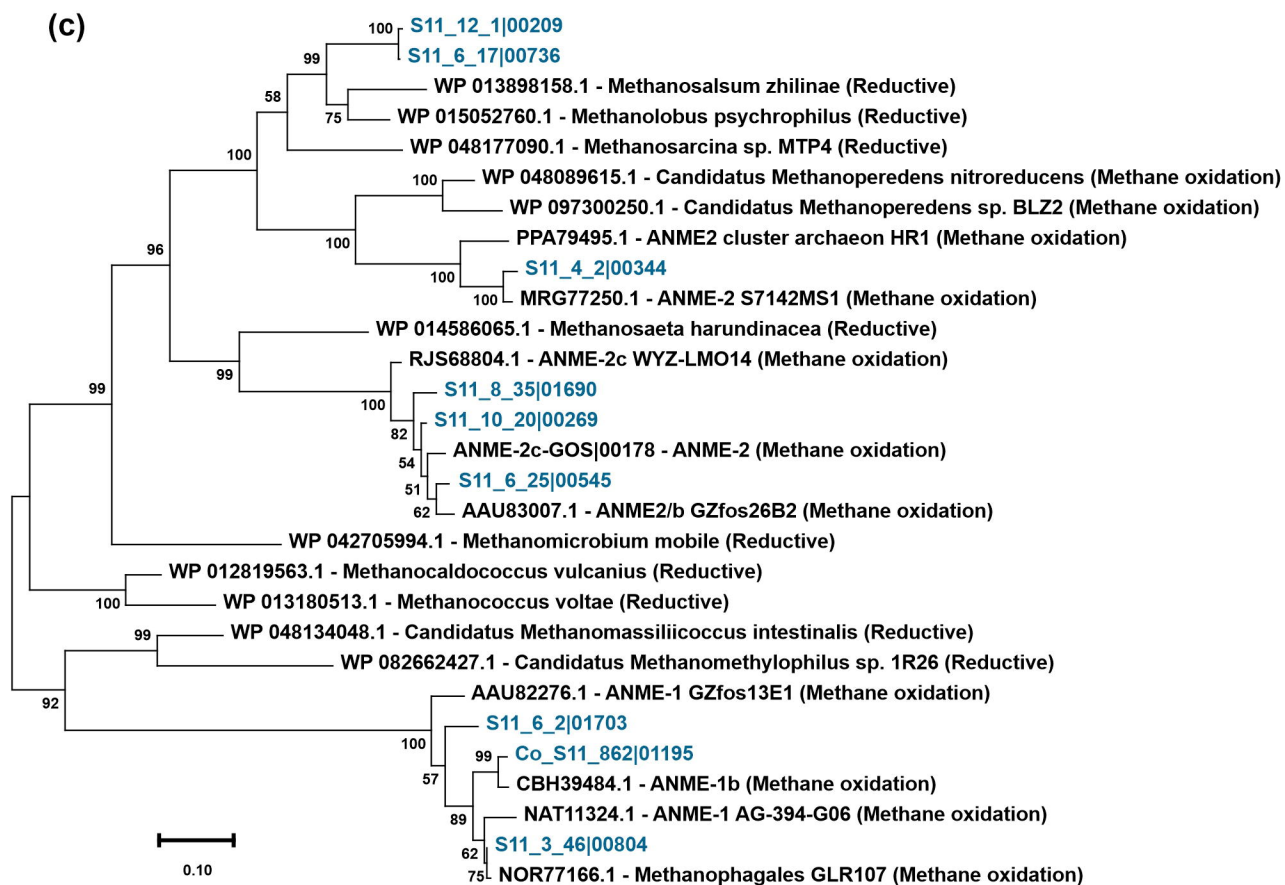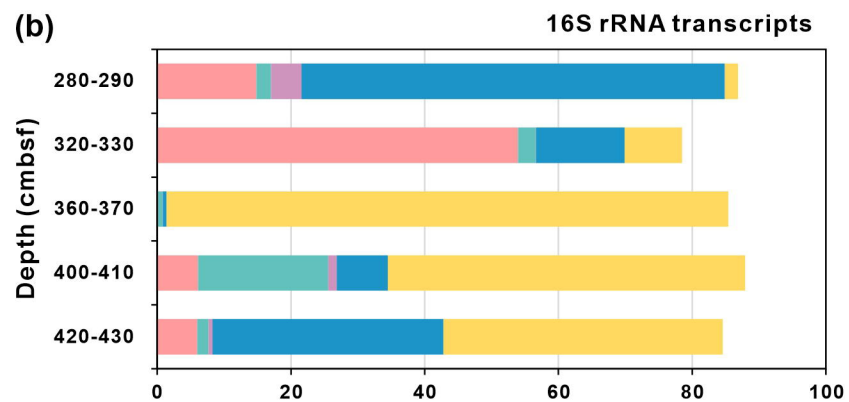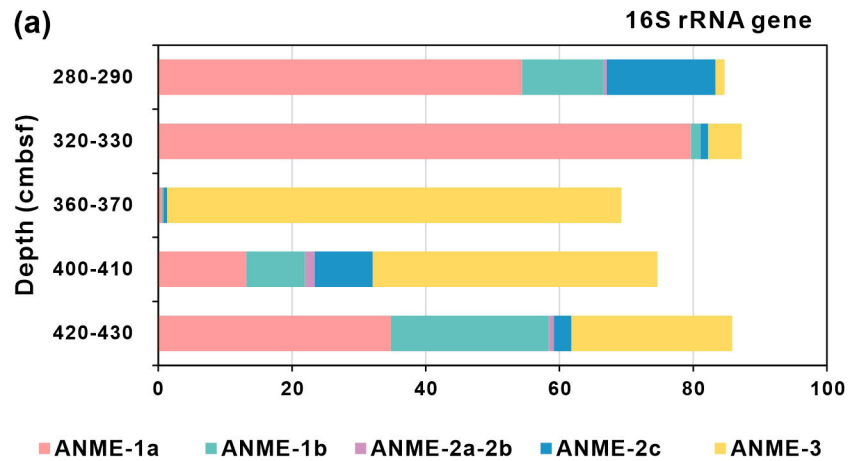642   **Figure 4 Modeled reaction rate profiles of S-AOM (green), and Fe-AOM (blue).**

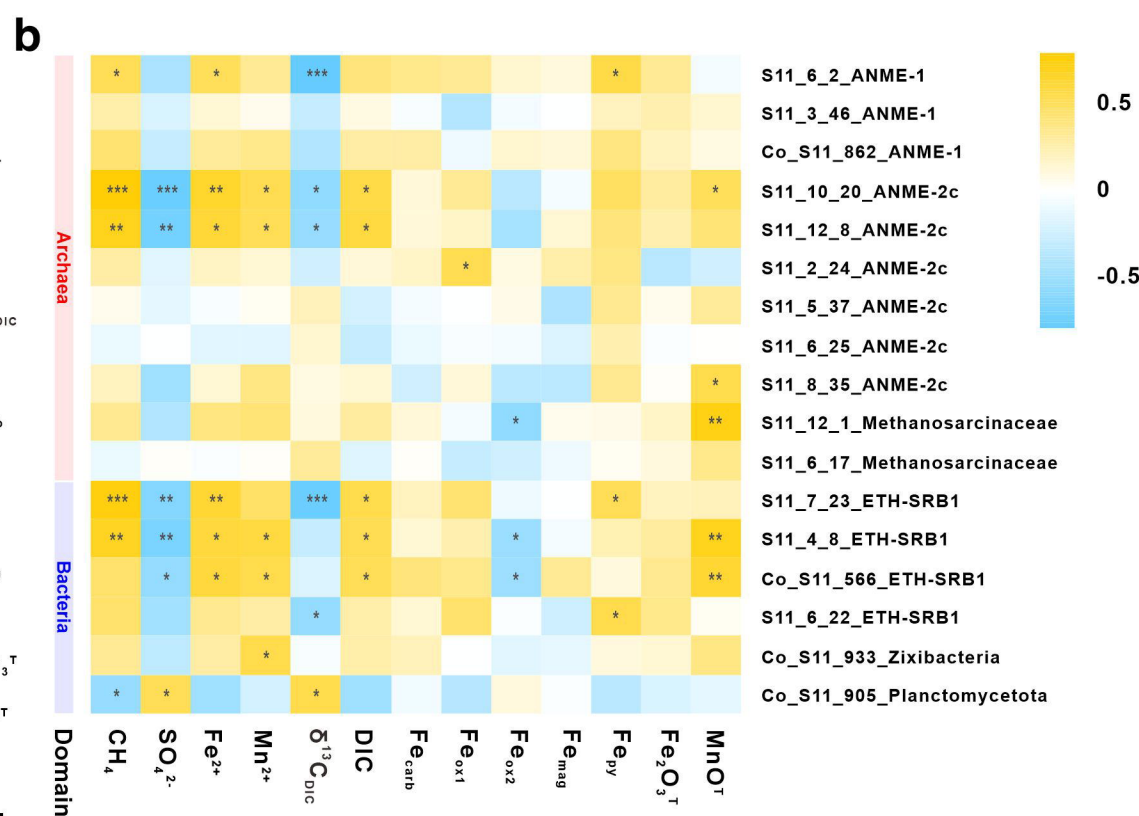643   **Figure 5 Model for AOM and methanogenesis in cold-seep marine sediments.**
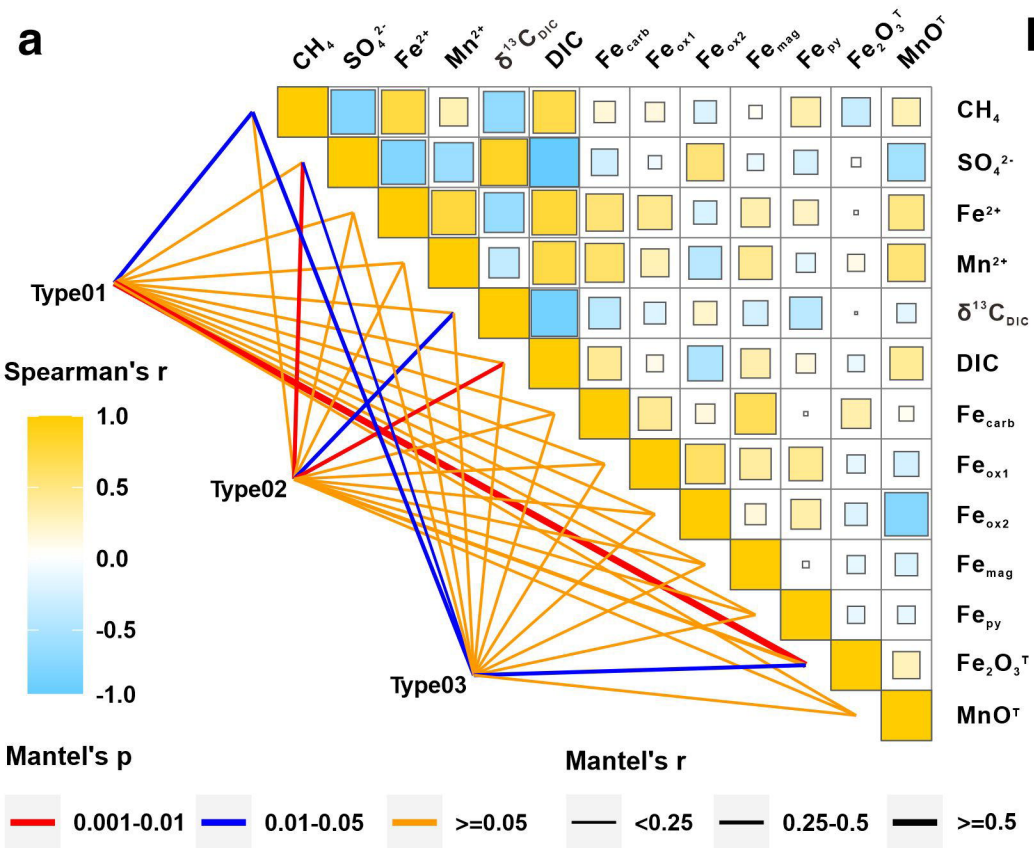
644 **Table 1 | Summary of the estimated rates of S-AOM, Fe-AOM and Mn-AOM in sediments from various freshwater and marine**

645 **environments**

Ecosystem	Environment	In situ concentration (μmol/L)		Model-derived rates (μmol CH₄ cm⁻³ yr⁻¹)			Depth-integrated rates (μmol CH₄ cm⁻² yr⁻¹)			The fraction in total CH₄ oxidation			Method	References
		Fe^{2+}	Mn^{2+}	S-AOM	Fe-AOM	Mn-AOM	S-AOM	Fe-AOM	Mn-AOM	S-AOM	Fe-AOM	Mn-AOM		
Marine	Haima cold seep	148	2289	0.66	0.02	N.A.	20.05	0.31	0.32	97%	1.5%	1.5%	Modeling	This study
	Hikurangi margin	184	N.A.	0.48	0.0005	N.A.	3	0.4	N.A.	88%	12%	N.A.	Modeling	Luo et al., 2020
	Black Sea	800	23	0.07	1.46E-05	N.A.	5.9	0.04	N.A.	99%	0.70%	N.A.	Modeling	Egger et al., 2016a
	Baltic Sea	600	N.A.	0.27	0.0011	N.A.	8.8	2.5	N.A.	78%	22%	N.A.	Modeling	Egger et al., 2017
	Bothnian Sea	1830	N.A.	N.A.	N.A.	N.A.	78.7	8	N.A.	90%	9%	N.A.	Modeling	Rooze et al., 2016
				6.94E-04	1.32	N.A.	58.38	1.63	N.A.	97%	3%	N.A.	¹³CH₄/Modeling	Egger et al., 2015
	Jiaolong cold seep	27	N.A.	328.57	13.87	N.A.	N.A.	N.A.	N.A.	N.A.	N.A.	N.A.	¹⁴CH₄	Li et al., 2020
	North Sea	380	40	2.04	0.03	N.A.	N.A.	N.A.	N.A.	98%	2%	N.A.	¹⁴CH₄	Aromokeye et al., 2020
	Eel River Basin seep	N.A.	N.A.	52	6	14	N.A.	N.A.	N.A.	N.A.	N.A.	N.A.	¹³CH₄	Beal et al., 2009
Freshwater	Lake Kinneret	70	N.A.	N.A.	1.26	N.A.	N.A.	N.A.	N.A.	N.A.	N.A.	N.A.	¹³CH₄	Sivan et al., 2011
	Dover Bluff salt marsh	30	80	2.41	1.42	0.876	N.A.	N.A.	N.A.	N.A.	N.A.	N.A.	¹⁴CH₄	Segarra et al., 2013
	Hammersmith Creek River	500	400	5.66	4.5	1.314	N.A.	N.A.	N.A.	N.A.	N.A.	N.A.	¹⁴CH₄	

646 Modeling: Geochemical modeling estimates; ¹³CH₄: ¹³CH₄ incubations; ¹⁴CH₄: ¹⁴CH₄ incubations; N.A., not available.

25

(a) 16S rRNA gene

(b) 16S rRNA transcripts

ANME-1a ANME-1b ANME-2a-2b ANME-2c ANME-3

(c)

$R_{S\text{-}AOM}$ (umol C cm^{-3} yr^{-1}) $R_{Fe\text{-}AOM}$ (umol C cm^{-3} yr^{-1})