

Inhibition of Microbial Methane Oxidation by 2-Chloro-6-Methylpyridine

Edward J. O'Loughlin ^{1*}, Dionysios A. Antonopoulos ¹, Kristin K. Arend ^{2§}, Theodore M. Flynn ^{1†}, Jason C. Koval ^{1‡}, and Sarah M. Owens ¹

¹ Biosciences Division, Argonne National Laboratory, Argonne, IL 60439-4843, USA; oloughlin@anl.gov, dantonop@anl.gov, sarah.owens@anl.gov

² Old Woman Creek National Estuarine Research Reserve, Huron, OH 44839, USA

[§] Current affiliation: United States Bureau of Reclamation, Science Division, Bay-Delta Office, Sacramento, CA 95814, USA; karend@usbr.gov

[†] Current affiliation: California Department of Water Resources, West Sacramento, CA 95814, USA; Theodore.Flynn@water.ca.gov

[‡] Current affiliation: Department of Medicine, Knapp Center for Biomedical Discovery, University of Chicago, Chicago, IL 60637, USA; jkoval@medicine.bsd.uchicago.edu

* Correspondence: oloughlin@anl.gov; Tel.: (1-630-252-9902)

Abstract: Several pyridine derivatives including the pesticide nitrapyrin [2-chloro-6-(trichloromethyl) pyridine] are strong inhibitors of methane monooxygenase, a key enzyme of aerobic methane (CH₄) oxidation. In this study we examined the effects of 2-chloro-6-methylpyridine (2C6MP) concentration on aerobic CH₄ oxidation and the development of populations of putative methanotrophs in sediment from Old Woman Creek, a freshwater estuary in Huron Co., Ohio. Experimental systems were prepared in serum bottles containing minimal medium with a headspace containing 20% O₂ and 10% CH₄. The microcosms were spiked with 2C6MP to achieve concentrations of 0, 0.1, 1, or 10 mM and inoculated with sediment. When headspace CH₄ concentrations decreased from 10% to < 2%, subsamples were taken for DNA extraction and sequencing of 16S rRNA gene amplicons. There was minimal effect of 2C6MP on CH₄ oxidation at concentrations of 0.1, and 1 mM, but complete inhibition for > 20 months was observed at 10 mM. ANOSIM of weighted UniFrac distances between groups of triplicate samples supported a primary distinction of the inoculum relative to the enrichments (R=0.999) and a secondary distinction between bottles containing 2C6MP versus those without (R=0.464 [0.1 mM]; R=0.894 [1 mM]). The inoculum was dominated by members of the *Proteobacteria* (49.9±1.5%), and to a lesser extent by *Bacteroidetes* (8.8±0.2%), *Acidobacteria* (8.9±0.4%), and *Verrucomicrobia* (4.4±0.3%). In enrichments with or without 2C6MP, *Proteobacteria* expanded to comprise 65–70% of the total. In the absence of inhibitor, members of the *Methylococcaceae* and *Methylophilaceae* increased in relative abundance from < 0.1% of the inoculum to 8.5±1.0% and 13.4±2.3%, of the total community respectively. At both 0.1 and 1 mM concentrations of the inhibitor, the *Methylococcaceae* were much less abundant, representing 3.3±0.5% and 2.8±3.3% respectively. No inhibition of the *Methylophilaceae* was seen at the lower concentration of 2C6MP, but at the higher concentration this taxon was only 7.8±1.1% of the total. In contrast, members of the *Crenotrichaceae*, another group of methane oxidizers, increased in relative abundance with greater amounts of inhibitor, representing 8.6±3.6% of the total at 0.1 mM and 12.0±4.5% at 1 mM, compared to only 4.1±0.4% when no inhibitor was present. These results clearly show changes in the populations of putative aerobic methanotrophs relative to the amount of 2C6MP present.

Keywords: methane oxidation, 2-chloro-6-methylpyridine, methanotrophs, nitrapyrin

1. Introduction

An improved understanding of the global terrestrial C cycle has become a policy imperative, both domestically and internationally, and is crucial in efforts to model, predict, and potentially mitigate the effects of increasing concentrations of greenhouse gasses on global climate. Although

not as prevalent in the atmosphere as carbon dioxide (CO₂), methane (CH₄) is an important greenhouse gas that accounts for ~20% of human-induced radiative forcing. Atmospheric CH₄ concentrations have increased by nearly 160% since 1850, largely due to human activities relating to large-scale land management and agricultural practices (e.g., wetland rice production, raising of ruminant livestock, and mining operations). Moreover, increased CH₄ emissions due to the warming of Arctic permafrost have been identified as a potentially significant factor resulting from (and contributing to) global climate change.

The formation (methanogenesis) and consumption of CH₄ (methane oxidation) in soils and sediments are the result of highly specialized microorganisms [1]. Methanogenesis is carried out by a group of archaea called methanogens, which are obligate anaerobes in the domain *Archaea*. In most anoxic environments (e.g., saturated soils, wetlands, lacustrine and marine sediments) methanogenesis is the final process in the anaerobic degradation of organic carbon. Most methanogens use CO₂ as the terminal electron acceptor in anaerobic respiration, reducing it to CH₄ (hydrogenotrophic methanogenesis) [2]. However, a limited number of methanogens can generate CH₄ from acetate fermentation (acetoclastic methanogenesis) [3]. Methane formed during methanogenesis can be oxidized to carbon dioxide via aerobic CH₄ oxidation by methanotrophic bacteria—obligate aerobes that use CH₄ as a sole C and energy source. Aerobic methanotrophs are found in a diverse range of aquatic and terrestrial environments including lacustrine, estuarine, and marine waters/sediments and soils ranging from arctic to tropical regions, and many are adapted to environments with extremes in pH (1–11), temperature (0–72 °C), and salinity (up to 30%) [4]. Despite the wide range in habitats, aerobic methanotrophic bacteria are not broadly distributed taxonomically, clustering primarily into two groups that differ with respect to phylogeny, ultrastructure, lipid composition, biochemistry, and physiology; type I methanotrophs are found within the *Gammaproteobacteria* in the family *Methylococcaceae* and type II methanotrophs are found within the *Alphaproteobacteria* in the family *Methylocystaceae*. Recently, a third group of aerobic methanotrophs have been identified comprising a single phylogenetic subcluster within the phylum *Verrucomicrobia* [5]. In addition to their role in C cycling via CH₄ oxidation, aerobic methanotrophs also play a role in N cycling via ammonium oxidation and N₂ fixation.

The pesticide nitrapyrin [2-chloro-6-(trichloromethyl) pyridine] is used to minimize the loss of nitrogen fertilizer (as ammonium) from soil by inhibiting ammonia monooxygenase (AMO), a key enzyme in the oxidation of ammonia to nitrate by nitrifying bacteria. However, AMO shares many similarities with methane monooxygenase (pMMO), a key enzyme of aerobic CH₄ oxidation, and nitrapyrin has been shown to directly inhibit aerobic CH₄ oxidation [6-8]. In this study we examined the effects of the nitrapyrin analog 2-chloro-6-methylpyridine (2C6MP) (Figure 1) on aerobic CH₄ oxidation and microbial community development in freshwater wetland sediment.



Figure 1. Molecular structure of nitrapyrin (left) and 2-chloro-6-methylpyridine (2C6MP) (right).

2. Materials and Methods

Overlying water and sediment (0-5 cm depth) were collected from Old Woman Creek, a freshwater estuary on the southern shore of Lake Erie in Huron Co., Ohio (41°22'35.472"N 82°30'29.999"W). The Old Woman Creek watershed encompasses 7,000 hectares, of which over 66%

is agricultural land. The sediment and water were combined to create a slurry, that was used to inoculate the experimental systems described below.

A defined minimal medium (DMM) consisting of 20 mM PIPES buffer, 20 mM HEPES buffer, 1 mM NH₄Cl, 1 mM KCl, 1 mM CaCl₂, 1 mM MgCl₂, 10 μM phosphate, 10 mL of trace metal solution [9], adjusted to pH 7.5, was sterilized by filtration through a 0.22 μm filter. The experimental systems were prepared in sterile 160-mL serum bottles containing 50 mL of DMM with a headspace consisting of 70% argon (Ar), 20% oxygen (O₂), and 10% CH₄, with 40 μmol of xenon (Xe) (as an internal standard), and sealed with butyl rubber plug stoppers and aluminum crimp caps. The microcosms were spiked with neat 2C6MP (99%, Sigma-Aldrich) to achieve concentrations of 0, 0.1, 1, or 10 mM (4 replicate microcosms at each concentration) and inoculated with 5 mL of sediment slurry. The microcosms were incubated at 30 °C in the dark and monitored for evidence of CH₄ oxidation using gas chromatography to measure changes in CH₄ and O₂ concentrations in the headspace over time. The concentrations of CH₄ and O₂ were measured by analyzing a 200 μL sample of headspace with an Agilent 7890A gas chromatograph equipped with a Supelco Carboxen 1010 Plot column (30 m × 0.35 mm) and a thermal conductivity detector, using Ar as the carrier gas (2.68 mL min⁻¹) and a temperature program of 60–100 °C at 5 °C min⁻¹, 100–230 °C at 40 °C min⁻¹, and held at 230 °C for 3.4 min. The system was calibrated by equilibrating known masses of analyte (CH₄ and O₂) and internal standard (Xe) in serum bottles having the same ratio of aqueous phase to vapor phase as the experimental systems, thereby accounting for water-vapor partitioning. When headspace CH₄ concentrations decreased from 10% to < 2%, a 10 mL sample of suspension was collected from each experimental system and frozen at -80 °C for subsequent DNA extraction and sequencing; samples for DNA extraction were not collected from the 10 mM 2C6MP treatments due to complete inhibition of CH₄ oxidation.

Samples were mixed prior to removing 0.5 mL of suspension for transfer to extraction tubes. DNA extraction was then performed using the MOBIO PowerSoil DNA Isolation kit according to the manufacturer's protocol. All DNA quantitation was performed using the Qubit assay (Invitrogen). The V4 region of the 16S rRNA gene (515F-806R) was PCR amplified as detailed in Caporaso et al. [10] to survey the total bacterial community in the extracted samples using the Earth Microbiome Project barcoded primer set adapted for the Illumina MiSeq <https://earthmicrobiome.org/protocols-and-standards/16s/>. Amplicons were then sequenced on a 151bp × 12bp × 151bp MiSeq run using customized sequencing primers and procedures as described by Caporaso et al. [11]. Triplicate libraries were prepared and sequenced for each sample as technical replicates. A total of 1.5 million paired sequences were generated (37,536 + 16,323 per amplicon library) and then merged followed by downstream processing using QIIME [12]. Briefly, singletons were removed, de novo OTU picking with uclust was used to cluster sequences at 97% similarity, and Greengenes (4feb2011) was used to assign taxonomies. For analyses of beta diversity, the total number of sequences in each library was normalized to the amount in the library containing the least number of sequences (7,168).

3. Results and Discussion

3.1. Inhibition of Methane Oxidation by 2C6MP

In the absence of 2C6MP, >80% of the CH₄ was oxidized within 11 days (Figure 2). The presence of 0.1 or 1 mM 2C6MP resulted in a limited inhibition of CH₄ consumption relative to the control without 2C6MP. However CH₄ oxidation was completely inhibited for > 20 months in the presence of 10 mM 2C6MP. In all systems, O₂ consumption tracked with CH₄ depletion, consistent with aerobic CH₄ oxidation. Our results are similar to those of Megraw and Knowles [13] showing ~60% inhibition of CH₄ oxidation by *Methylosinus trichosporium* OB3b in liquid culture containing 4 mM 2C6MP. However, Topp and Knowles [6] reported that concentrations of 2C6MP as low as 43 μM completely inhibited CH₄ oxidation by *M. trichosporium* OB3b. The higher concentration of 2C6MP needed for inhibition of CH₄ oxidation in our experimental systems compared to the pure

culture studies of Megraw and Knowles, and Topp and Knowles, may be due to uptake of 2C6MP by the wetland sediment, as sorption of nitrapyrin to soil has been shown to limit its effectiveness [14].

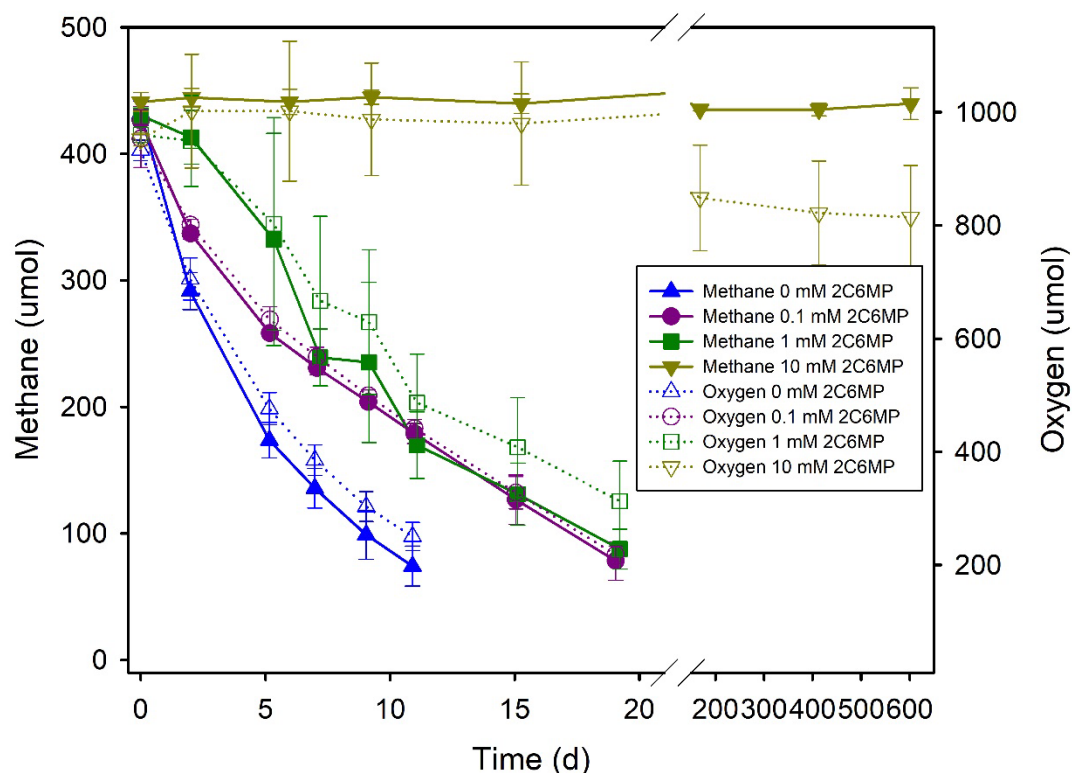


Figure 2. Changes in CH₄ and O₂ levels in the microcosm headspace over time.

Besides nitrapyrin and 2C6MP, other pyridine derivatives (as well as pyridine itself), have been studied as potential inhibitors of CH₄ oxidation. At a concentration of 1 mM, pyridine inhibited CH₄ oxidation by ~60% in pure cultures of *M. trichosporium* OB3b [15]. Picolinic acid (pyridine-2-carboxylic acid) inhibited > 90% of CH₄ oxidation by *M. trichosporium* OB3b at a concentration of 40 μM, but nicotinic acid (pyridine-3-carboxylic acid) and isonicotinic acid (pyridine-4-carboxylic acid) were largely ineffective (only 5.5% and 4.8% inhibition, respectively) [13]. These results highlight the significance of ring constituents and their location on the ring on the inhibition of microbial CH₄ oxidation. Although relatively high concentrations of 2C6MP were needed to completely inhibit CH₄ oxidation in our study, the effect was long lasting (> 20 months) compared to picolinic acid, which was readily degraded in soil, resulting in the loss of its inhibitory effects within 4 days [13], and nitrapyrin, which has a reported half-life of 5–42 days in soils [14].

In addition to pyridine, pyridine derivatives, and other aromatic *N*-heterocycles (e.g., 8-hydroxyquinoline) [15,16], a diverse range of compounds have been shown to inhibit CH₄ oxidation [17], including C₂ hydrocarbons (e.g., ethene and acetylene) [8,18,19], halogenated aliphatic hydrocarbons (e.g., fluoromethane, difluoromethane, and 1,3-dichloropropene) [18,20-23], and dimethyl sulfoxide [24].

3.2. Microbial Community Dynamics

Analysis of similarity (ANOSIM) of weighted UniFrac distances between groups of triplicate samples supported a primary distinction of the inoculum relative to the enrichments ($R=0.999$) and a secondary distinction between microcosms containing 2C6MP versus those without ($R=0.464$ [0.1 mM]; $R=0.894$ [1 mM]) (Figure 3).

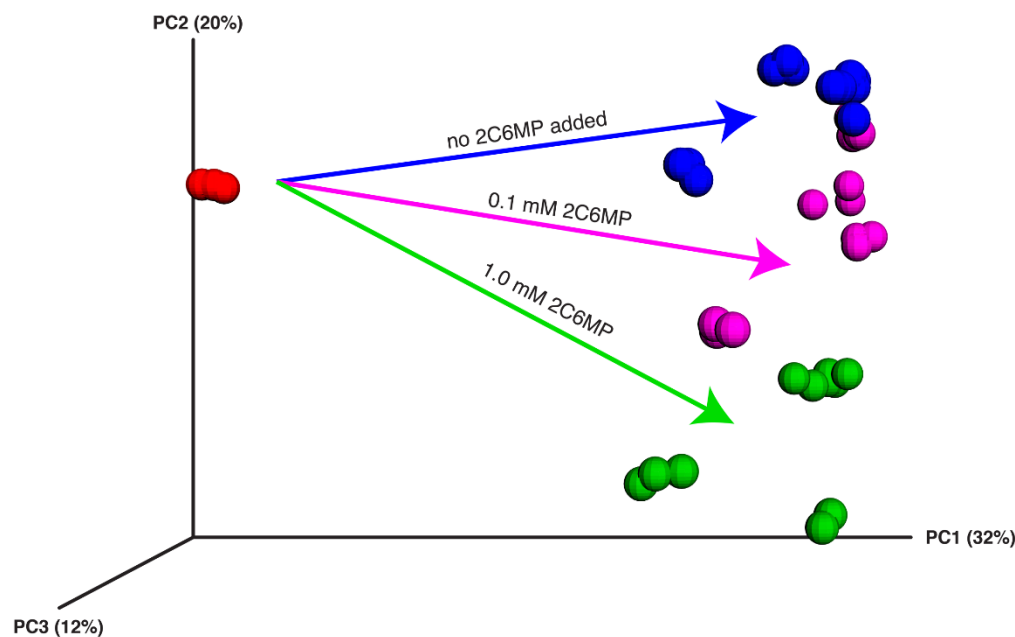


Figure 3. Principal coordinates analysis (PCoA) of microbial communities in Old Woman Creek inoculum and microcosms based on 16S rRNA amplicon libraries using the Illumina MiSeq platform. Distance between each data point is proportional to the weighted UniFrac distance. Triplicate DNA extractions were taken from each sample and are plotted separately.

The inoculum was dominated by members of the *Proteobacteria* (49.9%), and to a lesser extent by *Bacteroidetes* (8.8%), *Acidobacteria* (8.9%), and *Verrucomicrobia* (4.4%). In all microcosms, with or without 2C6MP, *Proteobacteria* expanded to comprise 65–70% of the total. In the absence of inhibitor, members of the *Methylococcaceae* and *Methylophilaceae* increased in relative abundance from < 0.1% in the inoculum to 10.8% and 18.1%, of the total community respectively (Figure 4). At both 0.1 and 1 mM concentrations of the inhibitor, the *Methylococcaceae* were less abundant, representing 7.9% and 8.4% respectively. No inhibition of the *Methylophilaceae* was seen at the lowest concentration of 2C6MP, but at 1 mM this taxon was only 11.4% of the total. In contrast, members of the *Crenotrichaceae*, which includes methanotrophs [25] that possess an ‘unusual’ pMMO [26], increased in relative abundance with greater amounts of inhibitor, representing 8.6% of the total at 0.1 mM and 12.3% at 1 mM, compared to only 4.1% when no inhibitor was present. Interestingly, members of *Rhodobacteraceae* were also more abundant in the presence of 0.1 (9.6%) and 1 mM (8.6%) 2C6MP than in the control (< 0.2%). Further differentiation in the relative abundance of dominant populations as a function of 2C6MP concentration is apparent at the genus level (Figure 5).

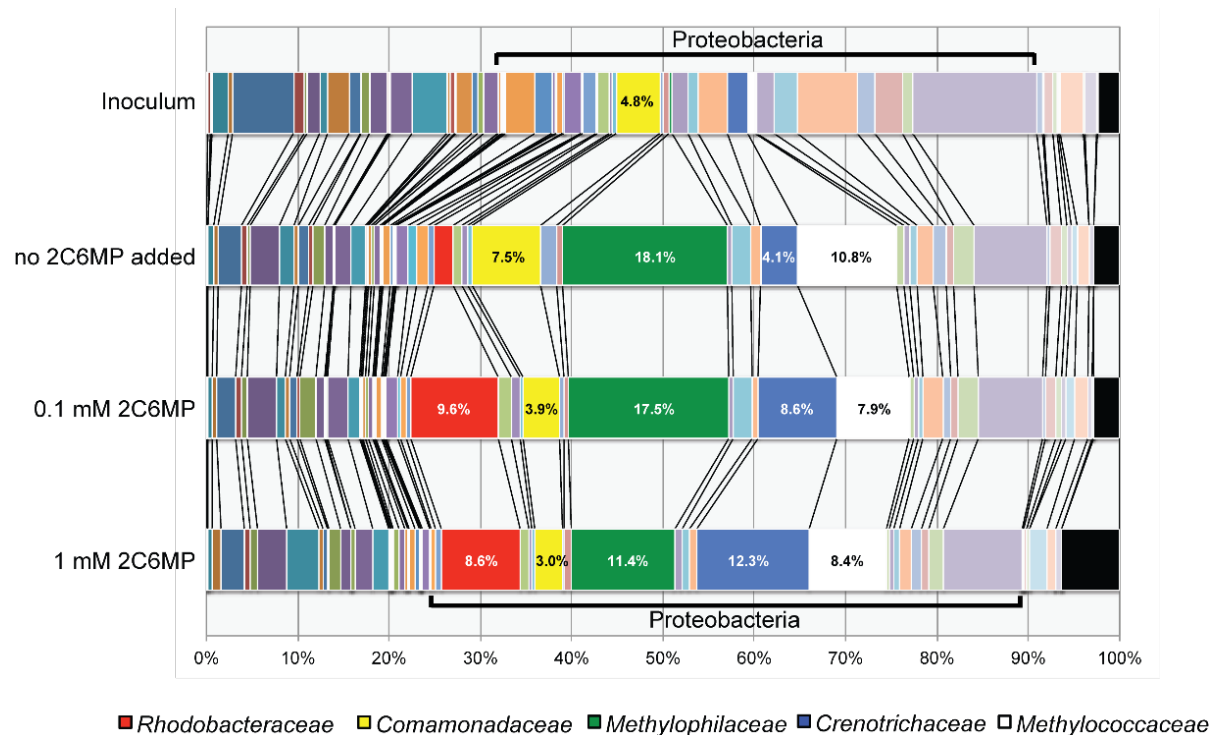


Figure 4. Bar graphs represent average family-level compositions for each treatment. Highlighted are key Proteobacteria families that increase in abundance relative to the inoculum.

Taxon

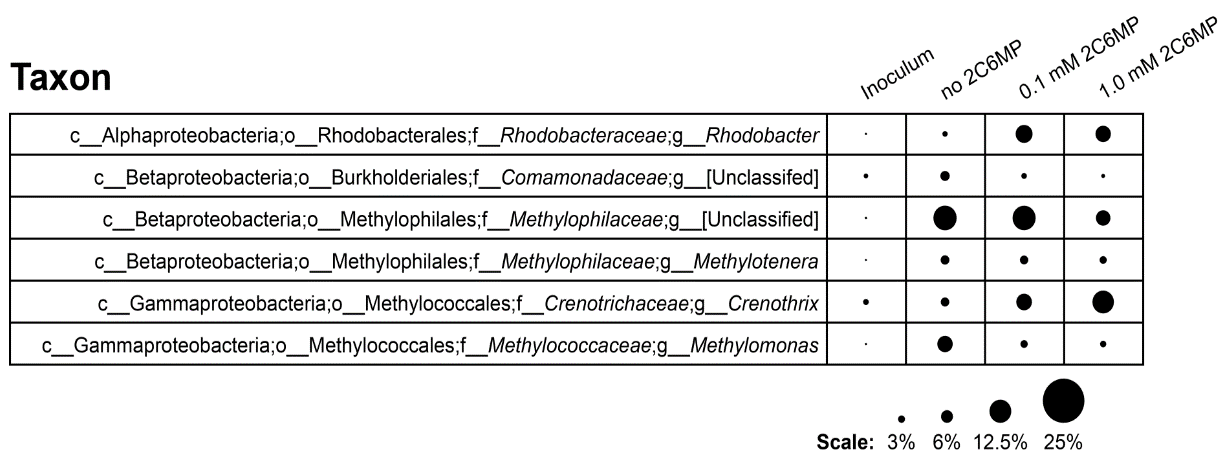


Figure 5. Taxonomy and average relative abundance of selected taxa in original Old Woman Creek inoculum and microcosms. Sequences were classified to the genus level of the Greengenes taxonomy using QIIME. The area of each circle is proportional to the average relative abundance of each genus.

3.3. Conclusions

At the highest 2C6MP concentration tested (10 mM), CH₄ oxidation was completely inhibited for >20 months. Although lower concentrations of 2C6MP (0.1 mM and 1 mM) exhibited similar levels of inhibition of CH₄ oxidation, there were distinct changes in the populations of putative aerobic methanotrophs relative to the amount of 2C6MP present. Though not as toxic as some other inhibitors of microbial CH₄ oxidation (i.e., higher concentrations of 2C6MP are needed for complete inhibition), 2C6MP may provide long-term inhibition and may therefore be useful as a CH₄ oxidation inhibitor in laboratory studies.

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