

1 **Hot spots drive uptake and short-term processing of organic and inorganic carbon and nitrogen**
2 **in intertidal sediments**

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4 Philip M. Riekenberg^{1*}, Bradley D. Eyre², Marcel T.J. van der Meer¹, Joanne M. Oakes²

5 1) Department of Marine Microbiology & Biogeochemistry, NIOZ Royal Netherlands

6 Institute for Sea Research, PO Box 59, Den Hoorn, 1790AB, The Netherlands

7

8 2) Centre for Coastal Biogeochemistry, Faculty of Science and Engineering, Southern Cross
9 University, Lismore, New South Wales, Australia

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12 *Corresponding author: phrieken@gmail.com

13 Running Head: Hot spots drive variability in mudflat uptake

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20 (equal)

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23 Abstract:

24 This study uses dual-labelled (^{13}C and ^{15}N) stable isotope applications to examine uptake and short-
25 term processing of carbon (C) and nitrogen (N) by microbial communities in intertidal sediment from
26 three subtropical estuarine sites. We examine differences in microbial uptake and retention that arise
27 due to domination of microbial processing by either microphytobenthos or heterotrophic bacteria. We
28 compare amino acids and algal dissolved organic matter (Algal DOM) and glucose and NH_4^+ versus
29 newly fixed microphytobenthos C (MPB-C) and NH_4^+ using *in situ* applications across 24 h to
30 identify uptake into the microbial community and sediment OM. Algal DOM had preferential C
31 uptake and more retention across 24 h indicating precursors incorporated into biosynthetic pathways
32 for biomass. Conversely, amino acid C was not incorporated or rapidly respired to DIC but displayed
33 clear preferential uptake and retention of ^{15}N . Short-term (24 h) retention of glucose was higher than
34 MPB-C, while uptake of ^{15}N from NH_4^+ was similar between treatments, potentially indicating
35 glucose-stimulated export of ^{15}N via coupled nitrification-denitrification. Despite careful selection of
36 similar sites and sediment types, we found substantial variability between replicates and sites in the
37 uptake and processing of labeled substrate that challenged traditional statistical analysis due to non-
38 homogenous variance. Uptake variability across orders of magnitude is likely due to disproportionate
39 processing of substrates occurring in hotspots of microbial processing within sediment. Development
40 of analytical techniques to provide robust strategies to handle variability caused by abiotic and biotic
41 factors will allow greater clarity surrounding *in situ* biogeochemical processing in intertidal
42 environments.

43

44 Introduction:

45 Intertidal sediments are important sites for intercepting and processing organic and inorganic
46 matter prior to its export to shallow coastal seas (Bauer et al., 2013). Within euphotic intertidal
47 sediments microbial communities dominated by microphytobenthos (diatom and cyanobacteria) are
48 important mediators for both carbon (C) and nitrogen (N) derived from inorganic and organic
49 substrates (Middelburg et al., 2000). Fixation of inorganic C via primary production coincides with
50 high uptake and competition for N by autotrophs that can limit N for heterotrophs, depending on
51 nutrient availability (Cook et al., 2007; Evrard et al., 2012; Riekenberg et al., 2020). Sediment C and
52 N uptake and processing pathways have been relatively well characterized using stable isotope
53 labelling techniques, (see below), but there is less evidence for how substrate quality, and small-scale
54 variability in uptake influences these processes. Previous work has often had minimal replication
55 primarily due to cost and the labour-intensive processing for samples supporting this work despite
56 finding considerable differences between uptake rates for ^{13}C and ^{15}N within a variety of single site
57 intertidal studies (coefficient of variation (%CV); Table 1).

58 Short term inorganic C uptake and processing is dominated by quick uptake and short-term
59 retention of C in MPB biomass (Moerdijk-Poortvliet et al., 2018) followed by longer term retention of
60 microphytobenthos derived carbon (MPB-C) in the sediments (30 d+) as extracellular polymeric
61 substances (EPS) and reworked detrital MPB biomass (Oakes and Eyre, 2014; Oakes et al., 2013;
62 Riekenberg et al., 2018). Newly fixed MPB-C is primarily exported via respiration and efflux of
63 dissolved inorganic carbon (DIC) to the water column. MPB-C that is retained over longer periods is
64 generally transferred to heterotrophic microbes and higher consumers via MPB exudates, which
65 comprise a snot-like complex pool of substrates MPB exudates are sugar rich (primarily glucose) with
66 labile components that are quickly (<24 h) used by heterotrophic bacteria (Oakes et al., 2012). More
67 refractory components (e.g., carbohydrates in EPS) are processed more slowly (>24 h) as they require
68 enzymatic break down into simpler components prior to use by heterotrophic bacteria (Miyatake et
69 al., 2014). EPS can account for a considerable (29%; Cook et al. 2007) to marginal (10%; Oakes et al.
70 2010) portion of sediment organic carbon that can have implications for the function and support of
71 intertidal food webs (Christianen et al., 2017; Riekenberg et al., 2022). Fixation of carbon, and
72 production of EPS, is stimulated under high light environments (Goto et al., 1999; McMinn and Lee,
73 2018) as diatoms use EPS to make their surrounding environmental conditions more favorable. EPS
74 helps to protect diatoms from excessive UV exposure, whilst also stabilizing sediments (minimizing
75 diatom resuspension), and facilitating migration by motile diatoms (Hoagland et al., 1993).

76 Short term nitrogen uptake and processing via MPB-mediated pathways are less well
77 characterized, but tend to be dominated by short term uptake of inorganic and organic materials from
78 the overlying water column (Eyre et al., 2016a; Veugel and Middelburg, 2007) followed by limited
79 transfer of N to heterotrophic bacteria (Evrard et al., 2008). In nutrient limiting settings MPB can
80 outcompete heterotrophic bacteria for nitrogen substrates. N limitation of the microbial community

81 results in strong coupling between MPB and heterotrophic bacteria that causes efficient recycling and
82 strong retention of N at the sediment-water interface (Cook et al., 2007; Risgaard-Petersen et al.,
83 2004). Increased nitrogen availability from the overlying water column makes the coupling between
84 heterotrophic bacteria and MPB less efficient (Oakes et al., 2020), causing an increased accumulation
85 of newly fixed N into the uncharacterized sediment pool (Riekenberg et al., 2020). Intertidal uptake of
86 N is dominated by incorporation into the microbial community followed by loss of N as N₂ via
87 coupled nitrification-denitrification (Eyre et al., 2016b) or via physical loss (Oakes et al., 2020).
88 Denitrification can be stimulated by access to a labile C source (Oakes et al., 2011) and is a vital
89 pathway for nitrogen removal in estuarine sediments (Slater and Capone, 1987) that varies
90 considerably within intertidal and subtidal ranges of subtropical estuaries (Eyre et al., 2016a; Oakes et
91 al., 2020) and coastal shelf seas (Chua et al. 2021), ranging from a minimal to considerable pathway
92 for N export (<1 to >100 $\mu\text{molN}_2\text{m}^{-2}\text{h}^{-1}$; Douglas et al., 2022).

93 An array of potential C and N substrates are available to microbial communities ranging in
94 complexity from glucose and ammonium to lignin (Bronk et al., 2007; See et al., 2006). However,
95 substrate bioavailability and oxygen availability largely determine whether these sources are used by
96 microbes (Berman and Bronk, 2003). Substrate use by the microbial community varies as uptake can
97 be limited by secondary limitations (e.g. C, N or phosphorous) or an inability to access the substrate
98 through enzymatic action (Crawshaw et al., 2019; Riekenberg et al., 2017). Thus, substrate
99 availability does not necessarily indicate microbial usage. The net role of oxygen, enzyme, and
100 secondary nutrient availability will influence benthic N metabolism, especially when organic matter is
101 the predominate source supporting the microbial community. The stoichiometry (C:N) and
102 complexity of organic matter govern microbial uptake, with the relative lability of OM contributing to
103 denitrification efficiency within systems (Albert et al., 2021; Eyre et al., 2013; Mayali et al., 2013;
104 Oakes et al., 2011). The complexity of carbon substrates can determine which heterotrophs thrive
105 within benthic microbial communities (Abell et al., 2013; Carlson et al., 2020) and marine biomass is
106 an important source of dissolved proteins and peptides supporting heterotrophic bacteria in the
107 intertidal zone (Schmidt et al., 2017; Seidel et al., 2015). Further examination of uptake and short-
108 term processing using labeled organic and inorganic substrates will be useful in directly tracking
109 substrate use and identifying rate differences in the processing pathways that potentially limit or drive
110 benthic N metabolism (Eyre et al., 2016a; Oakes et al., 2020). The variability and relative efficiencies
111 of uptake for carbon and nitrogen from organic and inorganic substrates within *in situ* intertidal
112 sediments largely remain a knowledge gap. Application of combined isotope-labelled substrates may
113 provide further insight into the short-term processing of organic matter and indicate which scenarios
114 are likely to stimulate short-term denitrification (e.g. lability of OM, and competition between
115 heterotrophs and microphytobenthos).

116 Here, we applied inorganic and organic substrates labelled with the rare stable isotopes ¹³C and
117 ¹⁵N to *in situ* intertidal sediments to examine uptake and short-term retention (24 h) by the sediment

118 microbial community within three subtropical estuaries. This work compares uptake and short-term
119 processing across a range of bioavailable compounds (e.g. glucose to dissolved organic matter) with
120 emphasis on 1) comparing the microbial processing of carbon and nitrogen from organic versus
121 inorganic substrates 2) comparing the microbial uptake and processing of carbon and nitrogen from
122 organic compounds across a range of lability and bioavailability and 3) quantifying the variability in
123 substrate use both within and between subtropical estuaries from sites with similar characteristics. We
124 expected 1) relatively labile substrates such as amino acids would have greater uptake and processing
125 than more complex pools such as algal derived DON, 2) uptake of ^{13}C and ^{15}N to be dominated by
126 MPB, and 3) the subsequent processing pathways for newly fixed OM to be dominated by support of
127 primary production pathways.

128

129 **Methods**

130 *Site description*

131 The study was undertaken in the austral spring (August and September) of 2017 in three river-
132 dominated subtropical estuaries: the Richmond, Tweed and Brunswick River estuaries in New South
133 Wales, Australia (Fig. 1A). Across the study period these sites had a semidiurnal tidal range of ~2 m
134 with site water salinity ranging from 25.2 to 28.6. This range is typical of Australian subtropical rivers
135 sampled close to the river mouth during low flow conditions which occur in the winter/spring
136 transition (Ferguson et al., 2004). Water column dissolved inorganic nitrogen (DIN) values typical of
137 this period are 1.4, 1.9, and 2.4 $\mu\text{mol L}^{-1}$ with DON values of 8.3, 9.6 and 11.8 $\mu\text{mol L}^{-1}$ for the
138 Brunswick, Tweed and Richmond, respectively (Eyre, 2000), with DON comprising the largest
139 portion of total nitrogen, which is typical in subtropical estuaries during low flow conditions (McKee
140 et al., 2000). Flow rates across the year can be considerably larger and quite variable due to episodic
141 flushing usually associated with low pressure systems or tropical cyclones during the wet season
142 summer/autumn (Ferguson et al., 2004) and this variability in flow can impact sediment processing
143 (Oakes et al. 2020; Oakes and Eyre 2014).

144 *^{13}C and ^{15}N labeling additions*

145 There are four treatment applications in this study: 1) algal DOM extracted from diatoms
146 raised on labelled NaHCO_3 and NH_4^+ resulting in atom% values of 37.1 for ^{15}N and 5 for ^{13}C , 2) an
147 amino acid mixture labelled 99 atom% for both ^{13}C and ^{15}N , 3) glucose and NH_4^+ , both labelled 99
148 atom% for ^{13}C and ^{15}N , respectively, and 4) NaHCO_3 and NH_4^+ labelled 98 and 99 atom% for ^{13}C and
149 ^{15}N , respectively. Further detail about preparation and concentration of each treatment is described
150 below.

151 The diatom-derived dissolved organic matter treatment (algal DOM; treatment 1) was
152 prepared following the method in Riekenberg et al. (2017) using an axenic culture of the diatom
153 *Thalassiosira pseudonana* grown in media containing 99% $^{15}\text{NH}_4^+$ and 98% $\text{NaH}^{13}\text{CO}_3$ (Cambridge
154 Isotope Laboratories). The extracted DOM had a ^{15}N atom% of 37.1 and a ^{13}C atom% of 5

155 (corresponding to ~160,000‰ and 3,800‰, respectively). A small concentration of remaining DIN
156 represented 2% of the total N in the treatment as measured via flow injection analysis after dilution
157 and persulfate digestion (Lachat, 1994). Any remaining dissolved inorganic carbon was removed
158 through acidification with 2N HCl prior to filtration of algal extract through a 0.45 µm syringe filter
159 (Minisart, Sartorius). The diatom-derived DOM was divided into aliquots prior to freeze drying that
160 were determined to be sufficient to produce a target concentration of 60 µmol L⁻¹ for DON and 162
161 µmol L⁻¹ for DOC when resuspended with 20 ml of filtered site water (0.2 µm, Sartorius).

162 The amino acid treatment (amino acid mixture; treatment 2) was prepared from a
163 commercially available 99 atom% dual-labelled (¹³C and ¹⁵N) mixture of 20 amino acids (aspartic
164 acid, glutamic acid, asparagine, serine, glutamine, histidine, glycine, threonine, alanine, arginine,
165 tyrosine, cystine, valine, methionine, tryptophan, phenylalanine, isoleucine, lysine, and proline;
166 Cambridge Isotope Laboratories). The amino acid mixture was suspended in deionized water and
167 divided into aliquots prior to freeze drying that were determined to be sufficient to produce a target
168 concentration of 60 µmol L⁻¹ for DON and 211 µmol L⁻¹ for DOC when resuspended with 20 ml of
169 filtered site water (0.2 µm, Sartorius).

170 Mixtures of glucose and NH₄⁺ (treatment 3) and NaHCO₃ and NH₄⁺ (treatment 4) mixtures
171 were prepared from 99 atom% ¹³C-labelled glucose, 99 atom% ¹⁵N-labelled NH₄⁺, and 98 atom% ¹³C-
172 labeled NaHCO₃. These two treatments were suspended in deionized water and divided into aliquots
173 prior to freeze drying that were determined to be sufficient to produce a target concentration of 60 and
174 20 µmol N L⁻¹ and 1100 and 1800 µmol C L⁻¹, respectively, when resuspended with 20 ml of filtered
175 site water (0.2 µm, Sartorius).

176 *Labeled substrate application*

177 Treatments were applied in August of 2017 (austral spring) to intertidal sediment at sites
178 within 2 km straight line distance upstream from the river mouths of the Richmond (28° 52' 30" S,
179 153° 33' 26" E), Tweed (28° 11' 43.1" S 153° 32' 49.2" E), and Brunswick (28° 32' 01.8" S 153° 33'
180 07.1" E) Rivers in New South Wales, Australia (Fig. 1A). Particular care was taken to select sites that
181 were similar within each estuary (i.e., similar location within estuary, sandy mud, similar inundation
182 times) to minimize environmental variability between sites. Areas within sites that were free of animal
183 burrows or wallows were selected to minimize 1) transfer of label between plots due to foraging, 2)
184 channeling of initial treatment application to deeper layers, or 3) uneven resuspension and flushing of
185 sediments during inundation due to wallows. At each site, four quadrats (one per treatment; 60 × 60
186 cm) were placed 2 m apart from one another at the same height within the intertidal zone. Quadrats
187 were divided into 9 plots (20 × 20 cm) to allow for 4 replicate samplings for 4 h and 24 h periods for
188 each treatment and leaving one blank per quadrat. Quadrats were anchored in two corners by 20 cm
189 lengths of PVC pipe pushed into the mud flat that remained in place between samplings to mark the
190 quadrat positions. Two loggers for temperature and light (HOBO) were attached to anchoring posts on
191 opposite corners of the outside quadrats across each site.

192 Treatments were applied to bare sediment on a falling tide during mid-day sunlight conditions
193 to ensure maximum time (1.5-4 h) for uptake of treatment applications prior to tidal inundation
194 flushing the majority of the unincorporated application from the sediment. Each application was
195 sprayed onto the sediment using mechanical sprayers to evenly disperse 20 ml of treatment onto each
196 of 8 plots within the treatment quadrat. Labeling dosages ranged from 10 to 113 $\mu\text{mol C m}^{-2}$ and 0.1
197 to 1.9 $\mu\text{mol N m}^{-2}$ (supplemental Fig. 1) amongst the treatments. Concentrations were higher than
198 those naturally occurring to provide a pulse of labeled material that would remain detectable after
199 tidal flushing. Treatment applications were applied individually to avoid any cross contamination
200 between treatments. Within treatment quadrats, individual plots were randomly allocated to 4 h and
201 24 h samplings with 4 replicates taken for each sampling period. This sampling design has been
202 previously found to be statistically independent (Riekenberg et al. 2017) and does not represent a
203 reduction of variance between samplings (pseudoreplication) due to the high variability found in
204 sediment at the millimeter to centimeter scale. The majority of unincorporated labelled material was
205 removed from the sediment due to replacement and turnover of porewater during tidal inundation
206 prior to sampling. Removal of both ^{13}C and ^{15}N at 4 h was confirmed by the highest remaining $\delta^{13}\text{C}$
207 and $\delta^{15}\text{N}$ values left in sediment organic matter being 194‰ and 591‰, respectively, which reflect a
208 small fraction (1.3% and 0.6%) of the initial treatment remaining. Any remaining, but unincorporated
209 material from treatment applications was found to be minimal (26±50‰ and 7±25‰ for $\delta^{15}\text{N}$ and
210 $\delta^{13}\text{C}$, respectively) through comparison between raw and KCl-extracted (for ^{15}N) or acidified (for ^{13}C)
211 sediments from the 0-1 cm layer at 24 h.

212 *Sample collection*

213 Prior to treatment application four cores (9 cm diameter \times 3cm depth) were sampled from
214 between the quadrats and sectioned into 1 cm depths to 3 cm, including a 1 mm top scrape and a 1cm³
215 sample for chlorophyll-*a* analysis, serving as controls for sediment characterization and $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$
216 values. All sections were placed in individual bags and stored frozen (-20°C), while the 0-1 cm depth
217 was immersed in liquid nitrogen and flash frozen directly to ensure preservation of the microbial
218 community for biomarker analysis prior to transport back to the laboratory. Within the quadrats, at 4 h
219 and 24 h after the treatment applications, four cores were sampled, similarly sectioned, and stored
220 frozen until freeze drying occurred within the laboratory. Prior to label applications, water from site
221 porewaters and the overlying water column was collected and passed through a 0.7 μm syringe filter
222 and frozen prior to analysis for baseline nutrient concentrations (Lachat, 1994) and $\delta^{15}\text{N}$ values for
223 NO₃⁻ via the bacterial denitrifier method (Sigman et al., 2001).

224 *Sample analysis*

225 Chlorophyll *a* was measured by colorimetry (Lorenzen, 1967) within the 0-1 cm depth for
226 both control and treatment application cores at each site. MPB-C biomass was calculated using a
227 conversion factor of 40 for C:chlorophyll-*a* which is in the range reported for microalgae in

228 subtropical Australian estuaries (30-60; (Ferguson et al., 2004; Riekenberg et al., 2018). MPB-C
229 biomass estimated in this way was only used as a measure to compare biomass between estuaries and
230 treatment applications and was not used for subsequent determination of ^{13}C or ^{15}N uptake or
231 retention.

232 Frozen sediment was freeze dried, homogenized, and a 3 cm³ subsample was extracted with
233 (2 M KCl inundation and centrifugation to remove 3× followed with 3× inundation and
234 centrifugation with double distilled water), or a 20 to 40 mg subsample acidified (two drop 2N HCl
235 into silver cups) to remove any adsorbed inorganic and organic N and inorganic C, respectively.
236 Washed or acidified sediment was dried (60°C) and weighed into tin or silver cups for analysis of %N
237 and $\delta^{15}\text{N}$ values or %C and $\delta^{13}\text{C}$ values, respectively, using a Flash 200 elemental analyzer coupled to
238 a Delta V Advantage isotope ratio mass spectrometer (Thermo Scientific, Bremen). Reproducibility of
239 $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ values was $\pm 0.1\text{\textperthousand}$ and $\pm 0.2\text{\textperthousand}$ for samples with enrichment less than 100‰ and
240 decreased for enrichment beyond that range.

241 To partition the uptake of ^{15}N into microphytobenthos and heterotrophic bacteria, we used
242 amino acid analysis of alanine to determine the relative uptake of ^{15}N between the D- and L- forms. A
243 subsample of sediment from the 0-1 cm depth (3 g) was acidified (2N HCl, 15 minutes or until
244 bubbling stopped) to remove carbonates, hydrolyzed overnight in 6N HCl at 110°C, purified using
245 cation exchange chromatography (Dowex 50WX8-100), and derivatized into N-pivaloyl-amino acid-
246 i-propyl esters. The concentrations and $\delta^{15}\text{N}$ values of the N contained in the chiral forms of alanine
247 (D/L-Ala) were examined via gas chromatography-combustion isotope ratio mass spectrometer using
248 50 m of CP-Chirasil Val columns (2 25 m × 0.25 mm × 0.12 μm , Agilent) mounted in an Agilent 6890
249 gas chromatograph connected to a Delta V advantage isotope ratio mass spectrometer via a
250 combustion III interface. Daily check standards were performed using an in-house standard mix of D-
251 and L-alanine and norleucine to confirm adequate separation between the chiral forms and to monitor
252 the reference standard (norleucine) $\delta^{15}\text{N}$ values to ensure adequate oxidation of the reactor across
253 runs. Samples for D/L-Ala were only examined from the 0-1 cm depth due to previous work having
254 confirmed that most ^{15}N uptake is confined to the 0-1 cm depth (>85%; Riekenberg et al. 2017) and
255 limitations of time and cost associated with the laborious nature of this analysis.

256 *Calculations*

257 ^{13}C and ^{15}N data are presented as excess μmol ^{13}C or ^{15}N per square meter of sediment to
258 allow for direct comparison between estuarine sediments with different porosities, calculated as:

259 1) Excess ^{15}N or ^{13}C = ((at% X_{sample} – at% X_{control}) × (μmol N or C in sample) / g dry
260 sediment) * g sediment m⁻²

261 where at% X is the percentage of C or N atoms present as ^{13}C or ^{15}N in the sample or control
262 sediment, and g sediment m⁻² is the dry weight of sediment remaining when porosity is accounted for
263 in each estuarine sediment. To account for the difference in atom % for both ^{13}C and ^{15}N in the algal

264 DOM treatment and the other treatments, we adjusted the values by 20 \times and 2.7 \times , respectively. To
265 provide $^{13}\text{C}/^{15}\text{N}$ ratios, the sum of excess ^{13}C and ^{15}N was calculated across the 0-3 cm sediment
266 depths examined.

267 Statistical analysis:

268 Variance for ^{13}C and ^{15}N areal uptake and $^{13}\text{C}/^{15}\text{N}$ ratios was significantly different (Levene's
269 HOV, $p<0.05$) for each of the variables (treatment, estuary, sampling) time regardless of the
270 transformations indicated by the bestNormalize package in R (Peterson and Cavanaugh, 2020).
271 Therefore, comparisons between organic and inorganic treatments among estuaries were made using
272 non-parametric pairwise Wilcoxon sign-rank tests. ANOVAs were used to compare biomass
273 contributions for heterotrophic bacteria, where variance was homogenous (Levene's $F_{2,69}=1.8$, $p=0.2$).
274 Biomass contributions were similar between the 4 and 24 hour periods (3-way ANOVA: $F_{1,71}=2.6$,
275 $p=0.1$, Hour) with no significant interaction between Hour and Estuary or Treatment. We therefore
276 proceeded with 2-Way ANOVAs for Estuary*Treatment. For the contribution of heterotrophic
277 bacteria (HB%) from D/L-Ala measurements we found a significant 3-way interaction (3-way
278 ANOVA: $F_{6,70}=2.3$, $p=0.05$, Estuary*Treatment*Hour). Variance was homogenous within the data set
279 (Levene's $F_{3,67}=1.52$, $p=0.2$) and Hour was not significantly different ($p=0.06$) as a variable or in
280 interactions with Estuary and Treatment so we performed separate 2-Way ANOVAs for
281 Estuary*Treatment for the 4 and 24 h periods. To further examine the between site variability from
282 this study and the variability observed across the wider literature, we used the R package cvequality
283 (Version 0.1.3; Marwick and Krishnamoorthy 2019) to test for significant differences between the
284 coefficient of variation for uptake rates for C and N substrates.

285 **Results**

286 *Site characteristics*

287 Surface water and porewater nutrients were dominated by NH_4^+ concentrations ranging from 2 to 4
288 $\mu\text{mol N L}^{-1}$ and 27 to 124 $\mu\text{mol N L}^{-1}$, respectively (Fig.1B), with NH_4^+ comprising >98% of
289 porewater DIN and 54 -100% of surface water DIN. Water column and porewater concentrations for
290 NO_3^- were 0 to 1.2 $\mu\text{mol N L}^{-1}$ with $\delta^{15}\text{N}$ values of $6.6\pm10.5\text{\textperthousand}$ and $29.8\pm2\text{\textperthousand}$ across all three sites.
291 Across all sites and depths, sediment was dominated by coarse and medium sand (75-90%), with mud
292 fractions ranging from <0.5% in the Brunswick to 13% in the Tweed. Sediment density was higher
293 and porosity was lower from the Tweed to the Brunswick, with the Richmond intermediate in both
294 comparisons, ranging from 0.3 to 1.2 g m^{-3} and 0.9 to 0.3 g ml^{-1} . Both sediment density and porosity
295 varied significantly among estuaries (One-way ANOVA; density $F_{2,8}=45.7$, $p<0.001$; porosity
296 $F_{2,8}=125$, $p<0.001$).

297 MPB-C biomass estimated from chlorophyll-*a* in the control cores ranged from 271 ± 71 to
298 $50\pm23 \text{ mmol C m}^{-2}$ and was highest in the Richmond and comparably lower in the Brunswick and the
299 Tweed estuaries (One-way ANOVA; $F_{2,17}=27.5$, $p<0.001$). There was no significant difference
300 between MPB-C biomass in control and treatment applications in the Richmond (Tukey's HSD all

301 $p>0.05$; One-way ANOVA, $F_{4,34}=6$, $p=0.001$), the Brunswick (One-Way ANOVA, $F_{4,37}=1.3$, $p=0.3$),
302 or the Tweed estuaries (One-Way ANOVA, $F_{4,37}=2.7$, $p>0.05$).

303 *Sediment uptake and short-term processing*

304 Between the organic treatments (treatments 1 & 2), uptake of ^{13}C m^{-2} was higher for algal
305 DOM (treatment 1) than for the amino acid mixture (treatment 2) within all three estuaries (all
306 $p=0.036$, Fig. 2A). Within each estuary there was no change in ^{13}C m^{-2} (all $p>0.05$) from 4 to 24 hour
307 for either treatment. For data pooled across estuaries, we again observed higher uptake of ^{13}C m^{-2} for
308 algal DOM than amino acids ($p=0.009$ for both) at both 4 and 24 h (Fig. 2B) and no change in ^{13}C
309 uptake between 4 and 24 h ($p>0.1$). In contrast to ^{13}C uptake in the organic treatments (1 & 2), uptake
310 of ^{15}N m^{-2} was higher from the amino acids than for algal DOM within all three estuaries (all $p=0.036$,
311 Fig. 3A). Uptake of ^{15}N m^{-2} was similar (all $p>0.1$) within each individual estuary for each treatment
312 at 4 and 24 h after application. By pooling across estuaries, we found that uptake of ^{15}N m^{-2} was
313 higher for the amino acids treatment than for algal DOM at 24 h than at 4 h (Fig. 3B) and that uptake
314 of algal DOM was higher at 24 h than at 4 h ($p<0.01$) but all other treatments had similar uptake
315 ($p>0.1$) between the two time periods.

316 Between the inorganic N treatments (treatment 3 & 4), uptake of ^{13}C was higher for glucose
317 than for NaHCO_3^- in the Brunswick ($p=0.036$), approached significantly higher in the Tweed
318 ($p=0.06$), but did not differ in the Richmond River estuary (Fig. 4A). Within each estuary there was
319 no change in ^{13}C m^{-2} (all $p>0.05$) from 4 to 24 hour for either treatment. For data pooled across
320 estuaries, we found that uptake of ^{13}C m^{-2} was similar between treatments at 4 h but was higher for
321 glucose than for NaHCO_3^- at 24 h. Additionally, uptake of treatment 3 was nearly significantly higher
322 at 4 h than at 24 h (Fig. 4B; $p=0.06$) but was similar across the two periods for treatment 3. Between
323 treatments 3 and 4, uptake of ^{15}N m^{-2} (Fig. 5A) was similar in the Brunswick, higher for treatment 4 in
324 the Richmond ($p=0.036$) and higher for treatment 3 in the Tweed ($p=0.036$). By pooling across
325 estuaries, we found that uptake of ^{15}N m^{-2} was similar between treatments 3 and 4 between estuaries
326 (Fig. 5A; $p>0.1$) and time (Fig. 5B; $p>0.1$).

327 $^{13}\text{C}/^{15}\text{N}$ uptake ratios did not differ (all $p>0.1$) between 4 and 24 h samplings within each
328 individual estuary. At all sites $^{13}\text{C}/^{15}\text{N}$ ratios were higher for algal DOM than for the amino acid
329 treatment (treatment 1 vs 2; Fig. 6A; all $p=0.036$) but were only higher for the glucose/ NH_4^+ treatment
330 in the Tweed (Treatment 3 vs 4; Fig 6C; $p=0.036$). $^{13}\text{C}/^{15}\text{N}$ ratios were higher for the algal DOM
331 treatment than for the amino acids at 4 and 24 h (treatment 1 vs 2; Fig 6B; $p=0.009$ for both) but were
332 only higher for the glucose and NH_4^+ treatment at 24 h (treatment 3 vs 4; Fig. 6D; $p=0.009$).
333 Comparing the coefficient of variation between the ^{13}C ($F_{1,12}=19.4$, $p=0.047$) and ^{15}N ($F_{1,12}=13$,
334 $p=0.29$) applications indicated significantly different amounts of variance among the C substrate
335 applications.

336 Bacterial biomass indicated by D/L-Ala was marginally different between treatments (Fig 7A;
337 $F_{2,71}=2.9$, $p=0.04$), but a post hoc Tukey's test indicated no difference amongst treatments ($p>0.05$;

338 range 76 to 85%). Between estuaries, bacterial biomass was higher in the Richmond than the Tweed
339 ($F_{2,71}=6.6, p<0.01$; range 75-86%). Bacterial uptake indicated from the ratio of uptake of ^{15}N into the
340 D and L forms of alanine had a significant 3-way interaction that required applying 2-way ANOVAs
341 for Estuary*Treatment to the 4 and 24 h periods separately (Fig. 7B; 4 h, $F_{5,35}=4.8, p<0.01$; 24 h,
342 $F_{5,34}=2.9, p=0.03$). For the 4 h period, uptake of ^{15}N was dominated by bacteria in all treatments
343 except for glucose and NH_4^+ (62-65% vs. 23%, respectively), but by 24 h there was no significant
344 difference between treatments. However, the Tweed estuary had lower bacterial uptake of ^{15}N than the
345 other two estuaries (25% vs 53-55%) indicating higher retention of ^{15}N in MPB.

346

347 **Discussion:**

348 Through the application of dual labelled (^{13}C and ^{15}N) substrates and mixtures, we have
349 identified the short-term uptake and fate from each organic (algal DOM; treatment 1 and amino acid
350 mixture; treatment 2) and inorganic (NH_4^+ , glucose; treatment 3 and NH_4^+ , NaHCO_3 ; treatment 4)
351 application to reveal differences in the relative lability of organic materials and the dominant
352 processing pathways during uptake by MPB and heterotrophic bacteria. This work highlights the
353 considerable variability encountered from in situ sediment applications despite the effort taken in
354 targeting sediments from geographically and locally similar settings (e.g. river-dominated muddy
355 sands). This study represents a well-replicated comparison of short-term uptake rates for C and N
356 from multiple substrate types within intertidal sediments, sampling river dominated sub-tropical
357 estuaries across relatively small spatial and temporal scales. Although this work is spatially limited in
358 scope (subtropical East coast of Australia), it helps to clarify the extent of heterogeneity during uptake
359 and short-term processing of C and N across latitudes in similar intertidal settings (Table 1).

360 Studies using isotope labelling have observed considerable variability between both between
361 replicates and studies (%CV; 5-63% ^{13}C ; 7-60% ^{15}N) in short term uptake and processing of inorganic
362 or organic labeled substrates, regardless of whether the isotope-labelled substrate was applied to
363 slurries or intact sediments (Table 1 and included references). Studies included here represent the bulk
364 of the work undertaken in intertidal mudflats using rare stable isotope labelling techniques for carbon
365 and nitrogen. The variability observed across these 20 studies ranges across orders of magnitude thus
366 often making it difficult to satisfy requirements for statistical analysis, such as homogeneity of
367 variance, even following data despite transformation. Even coefficient of variation is significantly
368 different amongst ^{13}C substrate uptake ($F_{1,24}=53.2, p<0.001$) and ^{15}N substrates ($F_{1,23}=34.5, p=0.044$)
369 using a modified signed-likelihood ratio test (Krishnamoorthy and Lee, 2014) on studies with
370 adequate replication for inclusion. Variability does not appear to be a result of climate (e.g. latitude),
371 as variance in ^{15}N and ^{13}C uptake remains considerable across studies performed in temperate,
372 subtropical and for the single sub-Arctic study (Oakes et al., 2016). Variability during uptake may
373 reflect heterogeneity resulting from the combination of overlying water flow rates, nutrient
374 availability (Høgslund et al., 2023; Riekenberg et al., 2020), sediment composition (Pryshlak et al.,

375 2015; Wallace et al., 2020), substrate availability and quality, and the composition of the microbial
376 community (Carreira et al., 2013). In this study, MPB-C biomass measurements confirm that there
377 was no short-term (24 h) change in biomass resulting from treatment applications between the
378 estuaries examined here. However, despite similar biomass being present across our study, we
379 observed considerable variability in the uptake of labelled substrates. This indicates that the uneven
380 distribution of ^{13}C and ^{15}N between replicates across other studies may be due to 1) variable diffusion
381 of substrates through porewater or 2) heterogenous distribution of the microbial community that may
382 have caused the development of hotspots of uptake and biogeochemical processing (McClain et al.,
383 2003; Santner et al., 2015).

384 *Uptake from algal DOM and amino acids*

385 Amino acids can be a large portion of the organic matter pool in marine settings (Cowie and
386 Hedges, 1992). Microbial community uptake of amino acids can vary from minimal, resulting in
387 downward transport of newly fixed autotrophic material via the microbial shunt in the open sea
388 (McCarthy et al., 2004), to complete use by sponge symbionts (Shih et al., 2019). In estuarine
389 sediment, amino acids are expected to be quickly taken up and used by the microbial community as a
390 relatively labile molecule, but use of C and N from these compounds can be either coupled or
391 uncoupled (Veuger and Middelburg, 2007) depending on whether applied amino acids are directly
392 incorporated into biomass (coupled) or degraded into NH_4^+ and the accompanying carbon skeleton
393 prior to use by the microbial community. Higher retention of ^{13}C from algal DOM than amino acids
394 across the 4 and 24 h periods in the current study (Fig. 2) indicates that ^{13}C from DOM was
395 preferentially incorporated into biomass, whereas C from amino acids was more likely to be shunted
396 into supporting respiration and exported from the sediments.

397 In the organic treatments more ^{13}C was taken up from algal DOM than from amino acids
398 (~ 150 to $300 \mu\text{M C m}^{-2}$; Fig. 2A&B), which contrasts with previous work in sediment slurries that
399 observed preferential uptake of carbon from algal-derived amino acids vs urea, a relatively labile
400 substrate (Veuger and Middelburg, 2007). Higher use of ^{13}C from algal DOM indicates immediate
401 preferential use of ^{13}C vs amino acids (Fig. 2A), likely due to a more complex mixture of ^{13}C
402 compounds (e.g. short chain fatty acids, acetate, pyruvate) being incorporated into wider range of
403 biomass rather than being shunted towards supporting metabolism. Uptake was followed with
404 relatively higher retention of ^{13}C from algal DOM, a complex mixture of molecules, than from amino
405 acids across all three estuaries (Fig. 2B). Relatively low uptake of ^{13}C and higher uptake of ^{15}N from
406 amino acids indicates decoupled use between C and N and the dominant uptake pathway for AAs is
407 likely via extracellular amino acid oxidation. Oxidation of amino acids allows for the preferential use
408 of N from AAs and leaves the carbon skeletons as labile DOC to be respired by heterotrophic bacteria
409 and subsequently exported to the water column as DIC (Mulholland et al., 1998). The uptake rates for
410 both organic treatments (Fig. 2) were lower than those observed for glucose (~ 530 to $2100 \mu\text{M C m}^{-2}$

411 Fig. 4), which likely reflects the requirement for algal DOM to be degraded by enzymes before
412 becoming available for uptake.

413 High $^{13}\text{C}/^{15}\text{N}$ ratios for both organic treatments in the Richmond River and for algal DOM in
414 the Brunswick (Fig. 6A) indicate disproportionately higher immediate uptake and short-term retention
415 of ^{13}C from DOM than from the amino acid treatment. Initially high decoupled uptake of ^{13}C followed
416 with lower $^{13}\text{C}/^{15}\text{N}$ values at 24 h in the Brunswick reflect immediate use of labile carbon. These C
417 substrates are potentially carbohydrates and sugars included in the algal DOM treatment and carbon
418 skeletons from amino acids to support immediate respiration. Similarly high ^{13}C values at 24 h across
419 all three estuaries (Fig. 2) for algal DOM indicate considerable short-term retention of the newly fixed
420 ^{13}C observed at 4 h. Higher uptake and retention likely indicates incorporation of newly fixed carbon
421 into microbial biomass or refractory molecules within sediment. $^{13}\text{C}/^{15}\text{N}$ ratios for algal DOM were
422 lower than the applied amino acid treatment (4.4) in the Brunswick and Tweed estuaries (2.2±1.5; Fig.
423 6A). Higher uptake of N from amino acids indicates decoupled initial uptake of ^{15}N and ^{13}C reflecting
424 higher rates of microbial N use followed with preferential retention of ^{15}N within either biomass or
425 molecules in the bulk sediment. The presence of excess ^{15}N in D-alanine at 4 h (Fig. 7A) confirms that
426 uptake from both organic treatments was dominated by heterotrophic bacteria. Differences in
427 substrate use between estuaries reflect the balance between selective uptake into biomass due to
428 substrate quality, respiration, and remineralization in the short-term.

429 Uptake from glucose, bicarbonate, and ammonium

430 Between the treatments comparing NH_4^+ use in the Brunswick and Tweed estuaries,
431 considerably more ^{13}C was taken up from the glucose and NH_4^+ treatment than was newly fixed into
432 MPB-C from the NaHCO_3^- and NH_4^+ treatments (Fig. 4A). Tight coupling in the microbial
433 community is expected in conditions where carbon is limited to production by MPB under nutrient
434 limited conditions (Riekenberg et al., 2020) and glucose is therefore the predominant form of newly
435 fixed carbon (Moerdijk-Poortvliet et al., 2018) with limited alternative sources, especially during
436 sediment exposure. In the Richmond River estuary ^{13}C uptake was similar between the two treatments
437 indicating similar use of newly fixed MPB-C and labile sugars with uptake happening in a comparable
438 range to the Tweed. The Richmond River has been previously observed to be carbon limited (Oakes
439 and Eyre, 2014), so it is not surprising that there was similar uptake of ^{13}C between the two
440 treatments, indicating quick uptake and recycling of labeled substrates by MPB and heterotrophic
441 bacteria regardless of substrate form. This suggests comparable use of carbon between newly fixed
442 MPB-C and glucose which would be expected from a microbial community with tight coupling
443 between MPB and heterotrophic bacteria, with heterotrophic bacteria processing extracellular
444 polymeric substances and producing NH_4^+ that then supports MPB (Cook et al., 2007).

445 In contrast, in the Brunswick and Tweed, higher uptake by the microbial community of ^{13}C
446 from glucose than from newly fixed MPB-C indicates stimulated uptake of carbon in excess of what

447 occurs during production of MPB-C. This may be a result of the bacterial community using glucose to
448 support alternative processing pathways such as denitrification (Morelle et al., 2022), but this
449 possibility remains speculative without inclusion of a direct measurement of N₂ fluxes in this study.
450 Despite this limitation, these values serve as an *in situ* comparison between heterotrophic and
451 autotrophic C uptake and processing. Previous work incorporating direct measurements of N₂ fluxes
452 have identified contributions from denitrification to the nitrogen budget in the Richmond (~3%;
453 Riekenberg et al., 2020) and higher contributions in the Brunswick (~21%; Eyre et al., 2016a). The
454 similar uptake of ¹⁵N from observed for both NH₄⁺ treatments in the Brunswick may be due to export
455 of ¹⁵N via coupled nitrification-denitrification for treatment 3 as evident from the disproportionately
456 high uptake of ¹³C from glucose and increased incorporation of ¹⁵N in treatment 4 (NaHCO₃, Fig. 7A).
457 The comparable uptake of ¹⁵N between treatments is deceptive since application rates for NH₄⁺ were
458 ~3× higher in the glucose treatment. Low uptake and retention of ¹⁵N could potentially indicate either
459 considerably higher export rates or a low threshold for ¹⁵N saturation in these systems, but saturation
460 is unlikely due to higher uptake of ¹⁵N found in the amino acid treatment. Decreased uptake and
461 retention of ¹⁵N potentially indicates export from heterotrophic bacteria likely via coupled
462 nitrification-denitrification. Processing along this pathway leaves more of the total ¹⁵N contained in
463 MPB, effectively leaving MPB as a N reservoir in situations where denitrifying bacteria have been
464 stimulated, but again, this remains speculative without measured N₂ efflux between the treatments.

465 Apart from stimulated uptake of ¹³C from glucose in the Brunswick, ¹³C/¹⁵N ratios for
466 treatments 3 and 4 were equivalent across estuaries and time periods (Fig. 6C & D). This similarity
467 indicates that ¹³C uptake and processing occurs at similar rates regardless of the form of C that the
468 heterotrophs encounter, either simple sugars or newly fixed MPB-C. Despite the similarity in uptake
469 rates, the variability between replicates was considerably higher for the glucose application largely
470 driven by variable ¹³C uptake (Fig. 4). This suggests that hotspots (McClain et al., 2003) for simple
471 sugar uptake and processing occur across all of the sediments examined, where heterotrophic bacteria
472 have limited access to labile carbon but have the ability to readily shift if labile substrates become
473 available episodically. High variability in uptake occurs across sediments on the cm scale across all
474 the intertidal sediment examined here and likely reflects a flexible life strategy for a microbial
475 community that experiences routine variations in substrate availability. This flexibility in supporting
476 metabolism is well suited to intertidal environments where tidal inundation and exposure occur and
477 substrates are likely to be briefly available and then promptly limiting depending on the tidal cycle, or
478 in estuarine settings where rainfall can quickly change inputs to the system.

479 Generalities between organic and inorganic processing

480 Carbon uptake and processing is more variable than expected and ranged from almost
481 complete exclusion in the case of amino acids to glucose uptake as high as 3.5 mg m⁻² in a single
482 replicate in the Brunswick estuary. The wide ranges and considerable variability throughout this study

483 for the uptake of ^{13}C highlight both the varied pathways that support for MPB and heterotrophs and
484 how tightly coupled MPB production and heterotrophic processing are, depending on substrate
485 availability. Direct stimulation of heterotrophs with the addition of a labile sugar results in variable
486 uptake of ^{13}C and lower ^{15}N uptake and retention considering the 3 \times higher NH_4^+ application between
487 treatments 3 and 4. ^{13}C uptake was highly variable between replicates amongst the glucose
488 applications and suggests that hotspots for processing quickly develop in areas where heterotrophic
489 communities using the coupled nitrification-denitrification pathway quickly make use of the added
490 substrates. In contrast, uptake and processing via carbon fixation as MPB-C was lower and less
491 variable across the estuaries indicating more homogeneous incorporation and processing between
492 MPB and heterotrophic bacterial communities. It is unsurprising that MPB-C fixation is the dominant
493 pathway for carbon support of the microbial community in photic intertidal sediments, and this is
494 therefore reflected in the wider data set. Similarly, the relatively lower but consistent uptake of ^{13}C
495 from algal DOM implies that this substrate mixture was less available than newly fixed MPB-C but
496 was still readily incorporated and processed via heterotrophy. In contrast, amino acid C was generally
497 preferentially excluded with the notable exception of 3 replicates in the Richmond estuary, indicating
498 the dominant retention of N from these labile substrates, except where carbon limitation necessitates
499 the uptake of amino acid C. It is unclear from the current work what C limitation threshold prompts
500 the switch to amino acid C use rather than exclusion or complete export of amino acid C via
501 respiration. However, the limited use of amino acid C nevertheless implies an energetic disadvantage
502 to processing this material, suggesting that additional steps (enzymatic or membrane transfer) are
503 required to make amino acid C energetically profitable to support respiration.

504 Conclusion

505 This work used well replicated applications of organic and inorganic substrates across three estuaries
506 to explore 1) the relative uptake and processing between amino acids and a relatively less labile algal
507 DOM mixture and 2) the processing of labile sugars versus newly fixed MPB-C using two
508 applications of $^{15}\text{NH}_4^+$ coupled with glucose and $\text{NaH}^{13}\text{CO}_3^+$. In contrast to our expectations, amino
509 acid ^{13}C was generally excluded, although not completely, indicating the preferential use of N from
510 these compounds except in rare situations of exceptional C limitation. ^{15}N uptake from amino acids
511 was higher than from algal DOM and reflects the relatively more refractory nature of the algal DOM
512 mixture and supporting our hypothesis of labile materials being more readily used, but both substrates
513 demonstrated clear use and retention across 24 h. Uptake and processing of ^{13}C from glucose
514 (dominated by heterotrophic bacteria) was considerably more variable and somewhat higher than for
515 ^{13}C from NaHCO_3^+ (dominated by MPB) which likely indicates increased processing from hotspots of
516 coupled nitrification-denitrification occurring across the estuarine sediments. However, this result
517 remains speculative due to not measuring N_2 fluxes simultaneously in this study, despite observing
518 comparable ^{15}N uptake and retention for NH_4 between applications that were scaled 3 times higher for

519 the glucose treatment. In contrast to our expectations, uptake and processing routed through MPB via
520 fixation pathways was generally lower for ^{13}C , but variable for ^{15}N across estuaries indicating that
521 local processes likely contribute to regulating the rates of fixation and processing MPB-derived
522 organic matter even amongst similar river-dominated intertidal sediments.

523 Implications

524 The considerable variability in substrate use and processing observed in this study highlights a major
525 issue in the investigation of process rates in estuarine samplings. Hotspots and hot moments, localized
526 areas or times of disproportionately high biogeochemical processing, are recognized to routinely
527 develop and persist in sediments largely regulated by substrate availability (McClain et al., 2003) and
528 contribute to the considerable variability observed in uptake and processing within labelling studies.
529 This processing couples carbon and nitrogen use, but the majority of the studies examining
530 denitrification only characterize the processing of N, and not C (Douglas et al., 2022; Peng et al.,
531 2022). Coupled examination of both C and N can help identify which substrates are available and how
532 substrate quality changes affect the microbial community composition and processing pathways. This
533 work further highlights the need for in situ studies coupling gas measurements to treatment
534 applications to ensure that gas flux rates for quantification of target biogeochemical pathways are
535 possible. MPB-dominated sediments can display a wide range of uptake and processing rates resulting
536 with considerable variability (Table 1) that is difficult to adequately investigate using traditional
537 statistical techniques. High variability between replicates is characteristic of substrate processing in
538 sediments, and lowered labeling concentrations that better reflect environmental conditions within in
539 situ applications appear to exacerbate this problem. Future work should aim for sufficient replication
540 to allow for statistical analysis while accounting for limitations of labour and cost associated with
541 laboratory analysis, while also acknowledging that replicate variability may very well still interfere
542 with data analysis. Further work, development, and application of analytical techniques used for data
543 from systems that tend to be episodic and demonstrate high variability would help to further separate
544 and identify signals from biogeochemical processing (i.e., hot spots) from the noise (replicate
545 variability).

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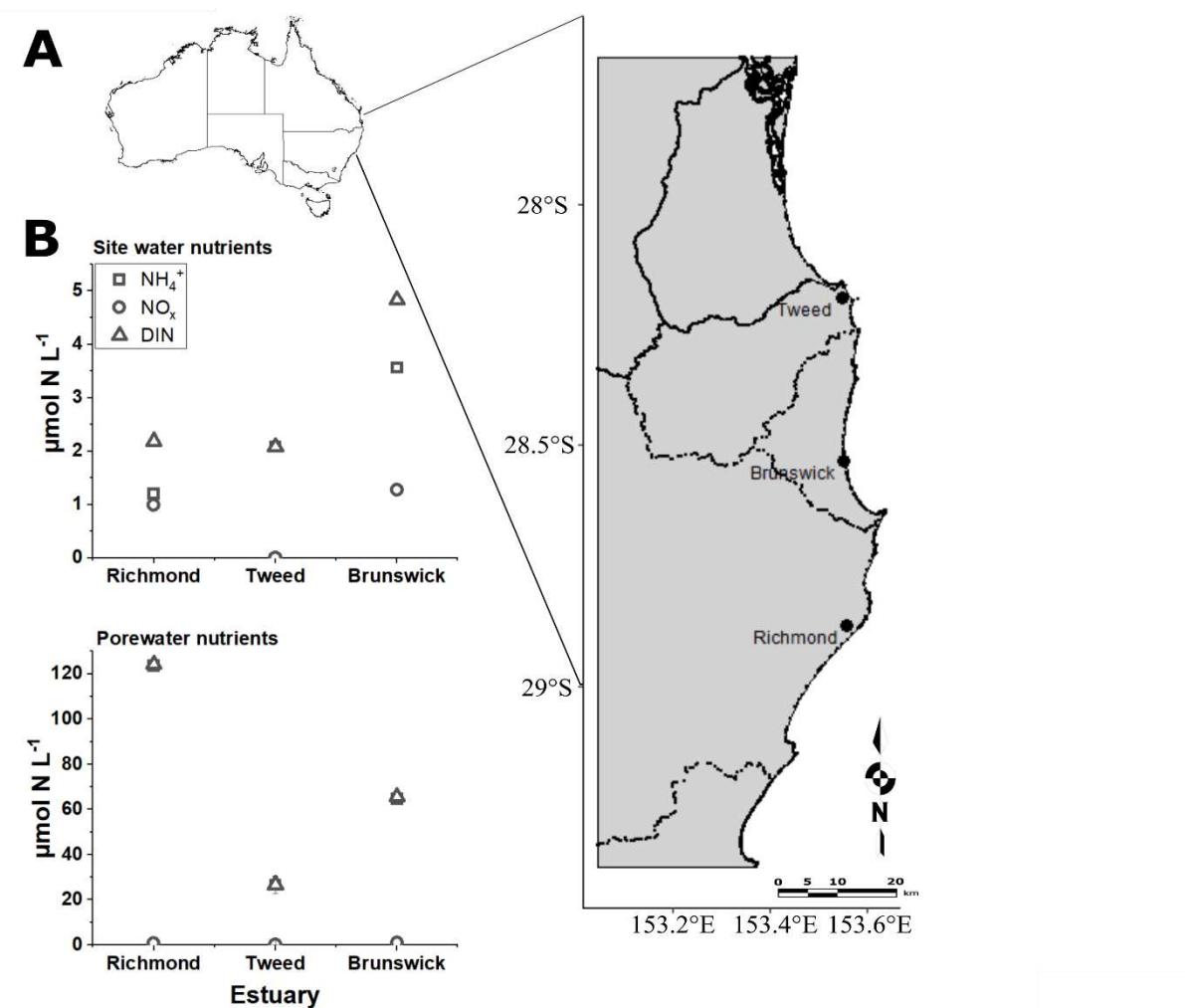
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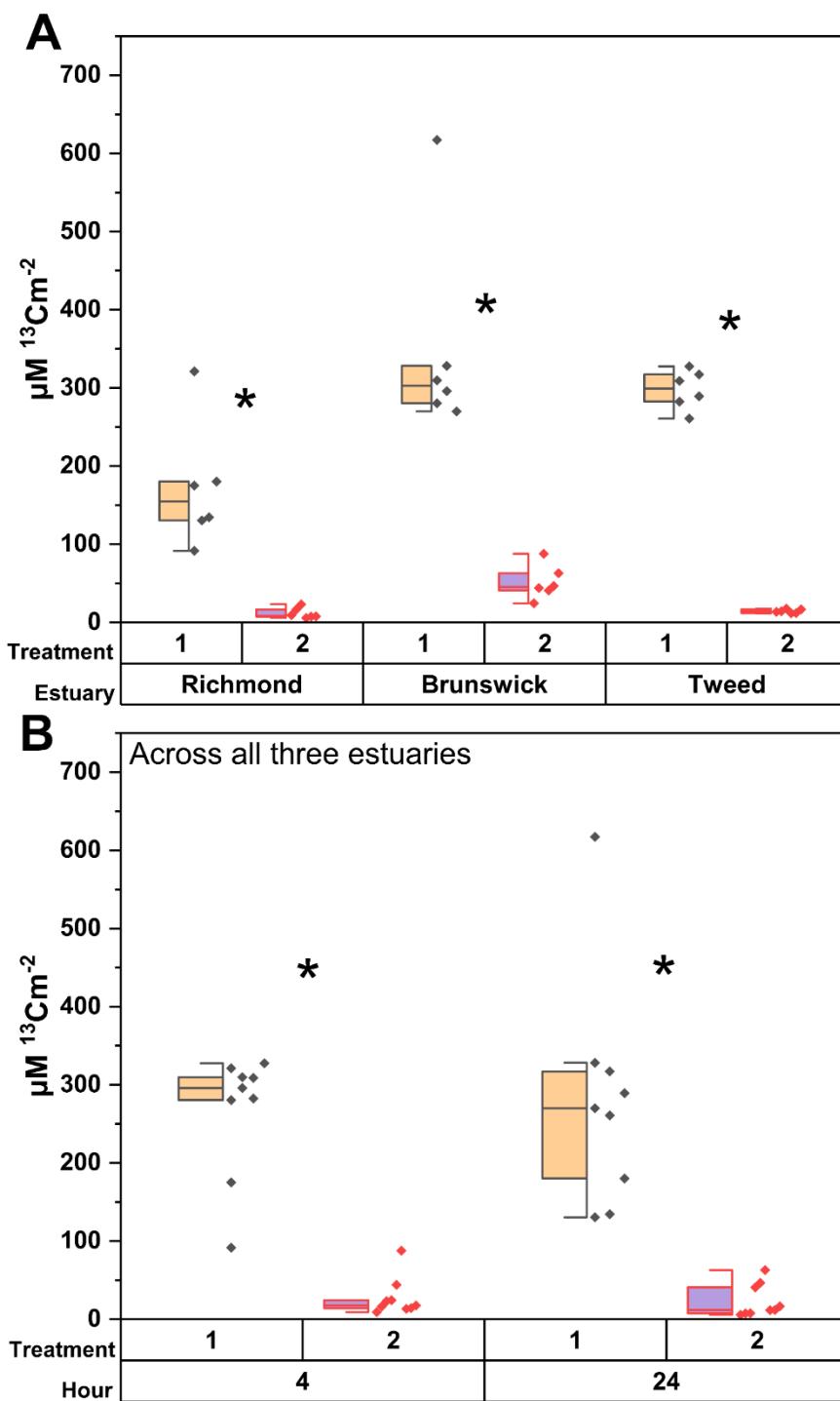
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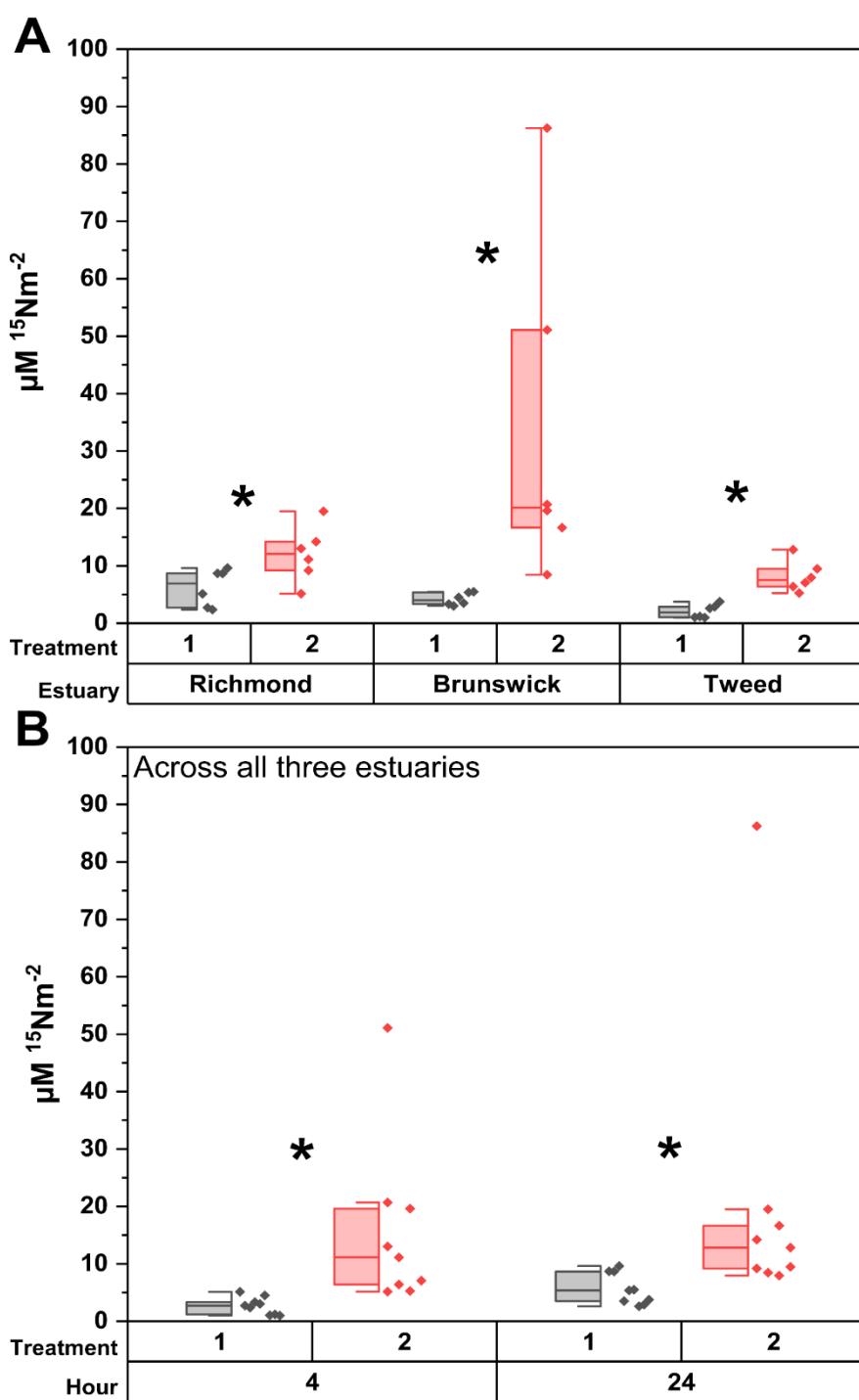


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Figure 1: A) Study site locations in New South Wales, Australia in the Tweed, Brunswick and Richmond River catchments (black outlines) and B) site water and porewater NH_4^+ , NO_x , and DIN ($\text{NH}_4^+ + \text{NO}_x$) values. Sampled mudflats were within 2 km straight line of the river mouths to the Pacific Ocean within each estuary.

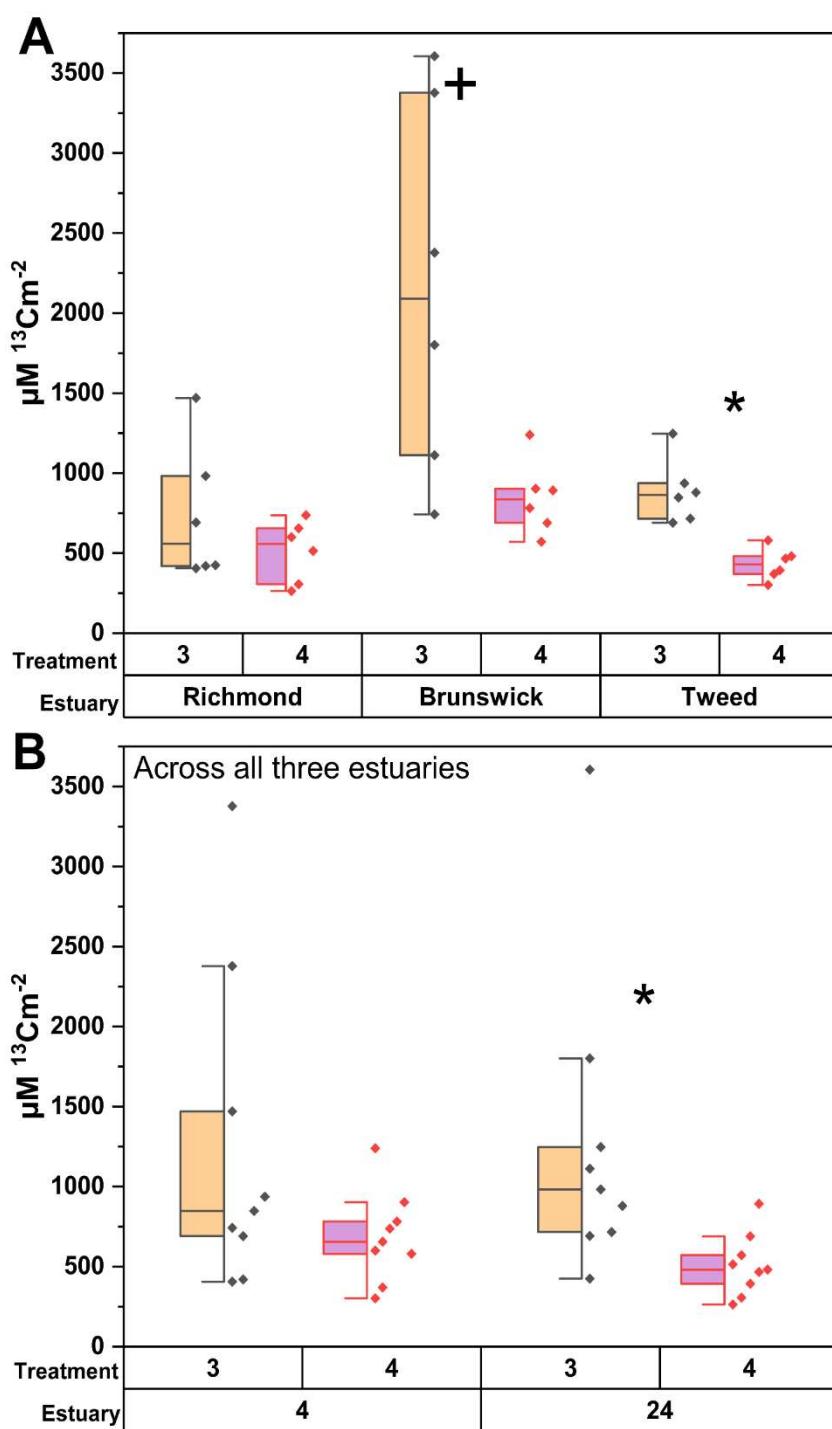


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751 Fig. 2: Uptake of ^{13}C from organic treatment applications applied to intertidal sediment A)
752 within each estuary and B) after 4 and 24 h across all three estuaries. Dual labelled (^{15}N and
753 ^{13}C) treatment applications contain: 1) algal dissolved organic matter and 2) an amino acid
754 mixture with uptake concentrations scaled to allow direct comparison between treatments. An
755 asterisk indicates a significant difference ($p < 0.05$) between pairings using non-parametric
756 pairwise Wilcoxon sign rank tests.
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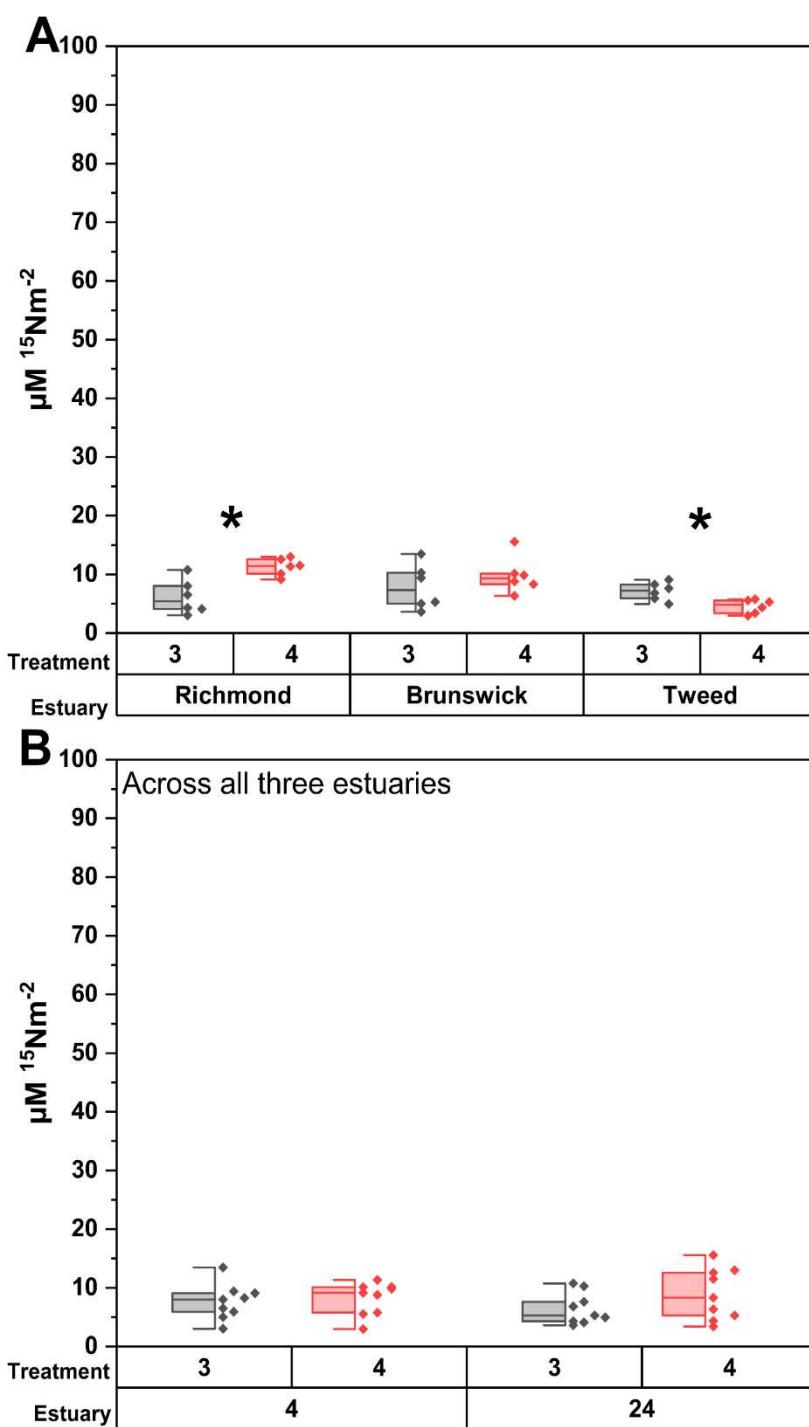
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759 Figure 3: Uptake of ^{15}N from treatment applications applied to intertidal sediment A) within
760 each estuary and B) after 4 and 24 h across all three estuaries. Dual labelled (^{15}N and ^{13}C)
761 treatment applications contain: 1) algal dissolved organic matter, and 2) an amino acid
762 mixture. Uptake concentrations have been scaled to allow direct comparison between
763 treatment applications. Asterisks indicate a significant difference ($p < 0.05$) between pairings
764 using non-parametric pairwise Wilcoxon sign rank tests.



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766 Fig. 4: Uptake of ^{13}C from inorganic treatment applications applied to intertidal sediment A)
767 within each estuary and B) after 4 and 24 h across all three estuaries. Dual labelled (^{15}N and
768 ^{13}C) treatment applications contain 3) glucose and NH_4^+ , and 4) NaHCO_3^- and NH_4^+ with
769 uptake concentrations scaled to allow direct comparison between treatments. An asterisk
770 indicates a significant difference ($p < 0.05$) between pairings and the cross mark indicates a
771 close to significant difference ($p = 0.06$) using non-parametric pairwise Wilcoxon sign rank
772 tests.



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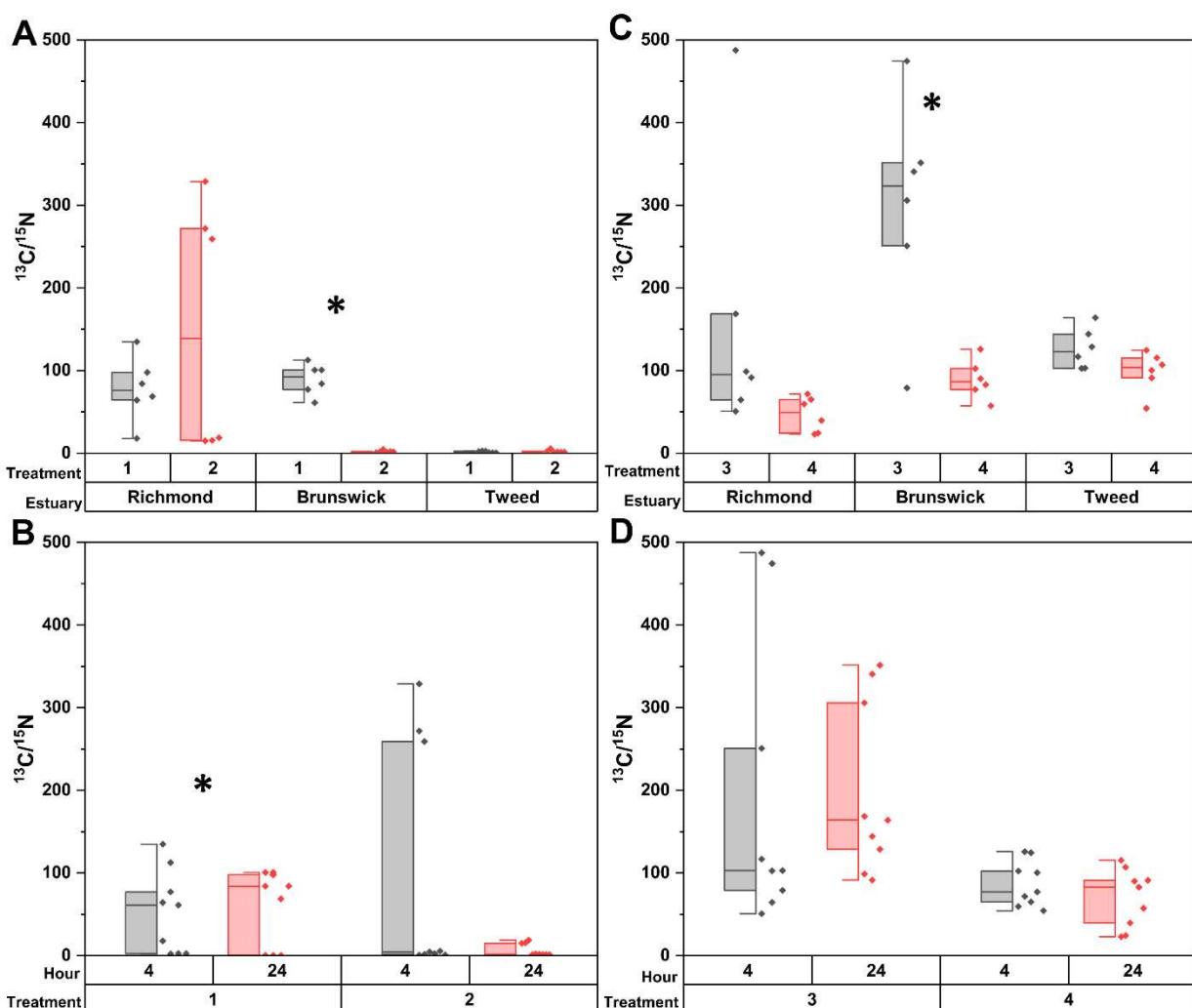
774 Fig. 5: Uptake of ^{15}N from inorganic treatment applications applied to intertidal sediment A)
775 within each estuary and B) after 4 and 24 h across all three estuaries. Dual labelled (^{15}N and
776 ^{13}C) treatment applications contain: 3) glucose and NH_4^+ , and 4) NaHCO_3 and NH_4^+ with
777 uptake concentrations scaled to allow direct comparison between treatments. An asterisk
778 indicates a significant difference ($p < 0.05$) between pairings using non-parametric pairwise
779 Wilcoxon sign rank tests.

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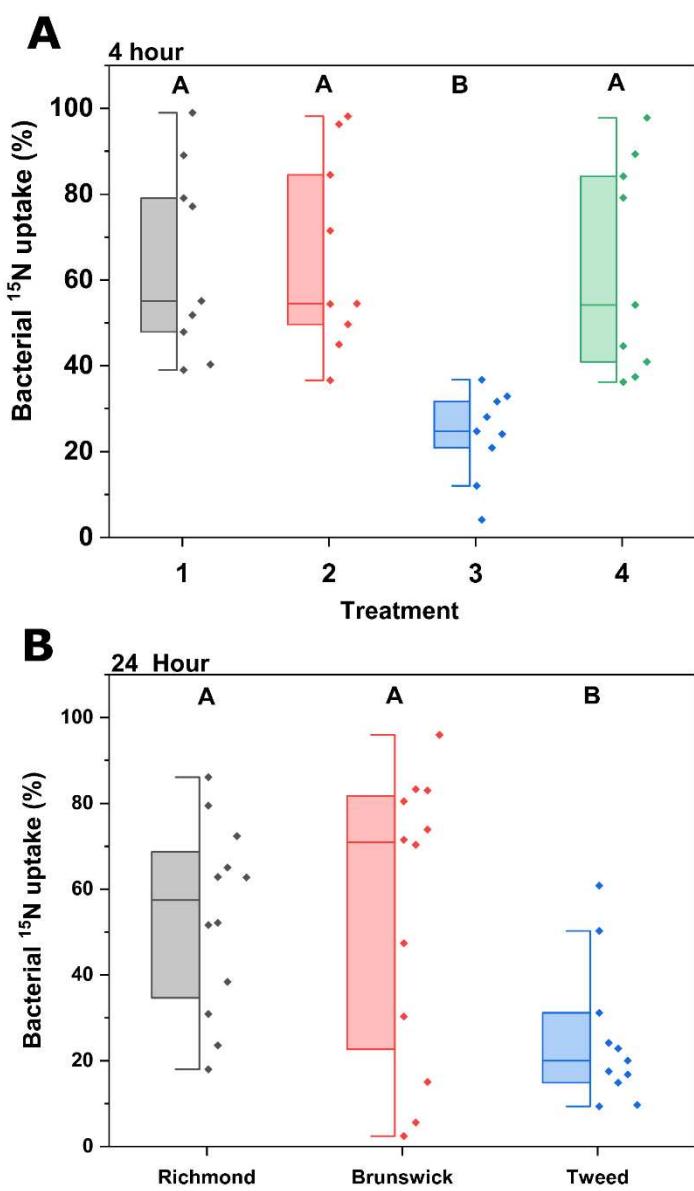
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Figure 6: ^{13}C to ^{15}N ratios retained in sediment from treatment applications applied to intertidal sediment A) within each estuary and B) at 4 and 24 hours across all three estuaries. Dual labelled (^{15}N and ^{13}C) treatment applications contain: 1) algal dissolved organic matter, 2) an amino acid mixture, 3) glucose and NH_4^+ , and 4) NaHCO_3 and NH_4^+ . Uptake concentrations have been scaled to allow direct comparison between organic (1 and 2) and NH_4^+ and glucose or NaHCO_3 (3 and 4) treatment applications. An asterisk indicates a significant difference ($p<0.05$) between pairings using non-parametric pairwise Wilcoxon sign rank tests.



810 Figure 7: Percentage of total ^{15}N incorporated into heterotrophic bacteria within the microbial
811 community. Bacterial uptake is derived from the ^{15}N content of D/L-alanine in diatoms and
812 heterotrophic bacteria.

813 Table 1: Compilation of studies examining uptake of ^{13}C and/or ^{15}N from organic or inorganic substrates into bulk measured intertidal sediments. SD
 814 indicates standard deviation where applicable.

Estuary	Country	Sample replicates	Time after application (h)	retained label N (umol m ⁻²)	SD	%CV	retained label C (umol m ⁻²)	SD	%CV	sediment depth (cm)	form N	Form C		
Schedt	the Netherlands	4	96				16651	5888	35.4	8	HCO_3^-	Middelburg et al. 2000		
Hel	Poland	2	24				766			3	HCO_3^-	Evard et al 2012		
Oosterschelde	the Netherlands	2	4				11250	540	4.8	1.5	HCO_3^-	Moerdijk-Poortvliet et al. 2018 (April high)		
Oosterschelde	the Netherlands	2	4				1190	60	5.0	1.5	HCO_3^-	Moerdijk-Poortvliet et al. 2018 (October low)		
Plum Island	USA	3	96				2770	515	18.6	0.5	HCO_3^-	Spivak and Ossolinski 2016		
Richmond	Australia	3	12				1549	140	9.0	10	HCO_3^-	Riekenberg et al 2018; Ambient		
Brunswick	Australia	2	60				892	32	3.6	2	HCO_3^-	Oakes et al 2010		
Brunswick	Australia	2	48				1809	110	6.1	10	HCO_3^-	Oakes et al. 2012		
Richmond	Australia	2.4 to 12					12000	4500	37.5	10	HCO_3^-	Oakes et al. 2014		
Nuup Kangerlua fjord	Greenland	3.4 to 12					7700	200	2.6	10	HCO_3^-	Oakes et al. 2016		
Richmond	Australia	3	4	10.2	1.1	10.8	664	69	10.4	3 NH_4^+	HCO_3^-	This study		
Brunswick	Australia	3	4	9.6	0.7	7.3	974	237	24.3	3 NH_4^+	HCO_3^-	This study		
Tweed	Australia	3	4	4.8	1.6	33.3	417	145	34.8	3 NH_4^+	HCO_3^-	This study		
Molenplaat	the Netherlands	1	24				11500			10	glucose	van Oevelen et al. 2006		
Schedt	the Netherlands	1	144				106			5	glucose	Veugel et al. 2006		
Richmond	Australia	3	4	5.8	2.6	44.8	765	610	79.7	3 NH_4^+	glucose	This study		
Brunswick	Australia	3	4	9.3	4.2	45.2	2166	1330	61.4	3 NH_4^+	glucose	This study		
Tweed	Australia	3	4	7.8	1.7	21.8	824	125	15.2	3 NH_4^+	glucose	This study		
Hel	Poland	2	1				1715		0.0	3	phytodetritus	Evard et al 2012		
Plum Island	USA	3	3				259	87	33.6	0.5	Spartina detritus	Spivak and Ossolinski 2016		
Richmond	Australia	3	24	88.4	42	47.5				3 Algal DOM		Riekenberg et al. 2017; DON 250 Intact		
Richmond	Australia	3	4	3.4	1.5	44.1	196	116	59.2	3 Algal DOM	Algal DOM	This study		
Brunswick	Australia	3	4	3.6	0.8	22.2	295	14.7	5.0	3 Algal DOM	Algal DOM	This study		
Tweed	Australia	3	4	1.1	0.1	9.1	306	22.7	7.4	3 Algal DOM	Algal DOM	This study		
Richmond	Australia	3	4	9.8	4.1	41.8	16.2	7.1	43.8	3 amino acid	amino acid	This study		
Brunswick	Australia	3	4	30.5	17.9	58.7	52	32.5	62.5	3 amino acid	amino acid	This study		
Tweed	Australia	3	4	6.2	0.9	14.5	15.1	2.2	14.6	3 amino acid	amino acid	This study		
Sylt	Germany	3	24	1148	214	18.6	1645	106	6.4	5 NaNO_3	HCO_3^-	Evard et al 2008, 2010		
Brunswick	Australia	2	24	665	182	27.4				10 KNO_3		Eyre et al 2016		
Richmond	Australia	2	4	9500						2 NH_4^+		Oakes et al 2020		
Richmond	Australia	6	24	276	103	37.3				3 NH_4^+		Riekenberg et al. 2017; DIN 500 Intact		
Richmond	Australia	6	12	216.7	45	20.8				10 NH_4^+		Riekenberg et al 2020; Ambient		
Intact sediments with no areal conversion		Country	Time after application (h)	retained label N (umol N gdw ⁻¹)	SD		retained label C (umol C gdw ⁻¹)	SD		sediment depth (cm)	form N	Form C		
Chesapeake Bay	USA	1	24	0.163			0.233			1 NH_4^+	HCO_3^-	Hardison et al 2011		
Oosterschelde	the Netherlands	1	12				0.24			1.5	HCO_3^-	Miyatake et al 2014		
Sediment Slurries		Country	Time after application (h)	retained label N (nmol N gdw ⁻¹)	SD		retained label C (nmol C gdw ⁻¹)			sediment depth (cm)	form N	Form C		
Richmond	Australia	6	24	124	22.8	18.4					Algal DOM		Riekenberg et al. 2017; DON 250 Slurry	
Richmond	Australia	6	24	737	135	18.3					NH_4^+		Riekenberg et al. 2017; DIN 500 Slurry	
Schedt	the Netherlands	1 0 to 1		40						0.1	urea		Veugel and Middelburg 2007	
Schedt	the Netherlands	1 0 to 2		60						0.1	amino acid		Veugel and Middelburg 2007	
Schedt	the Netherlands	2	48	245	16	6.5	526	40	0.08	10 NH_4^+	glucose		Veugel et al. 2012	